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## UTILIZATION OF *IN VITRO* CYTOCHROME P450 INHIBITION DATA FOR PROJECTING CLINICAL DRUG–DRUG INTERACTIONS

JANE R. KENNY, DERMOT F. MCGINNITY, KEN GRIME, AND  
ROBERT J. RILEY

*AstraZeneca R&D Charnwood, Loughborough, United Kingdom*

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## 22.1 INTRODUCTION

Adverse drug reactions (ADRs) are recognized as a significant cause of hospitalizations and up to 2.8% of hospital admissions may be a consequence of drug–drug interactions (DDIs) [1, 2]. The possibility of DDIs presents whenever two or more drugs are administered simultaneously; an ever more likely occurrence in an aging population often exposed to polypharmacy [3]. Moreover, it has been estimated that >90% of ADRs are also associated with dose-dependent pharmacokinetic/pharmacodynamic (PK/PD) events, which are one of the top five causes of death in the United States, resulting in over 100,000 deaths each year [1, 2, 4]. While this is alarming, it is noteworthy that of the drugs associated with preventable drug-related hospital admissions, only four drug groups account for more than 50% of ADRs [5]. Indeed, drug concentration-dependent pharmacological reactions (classified as type A adverse reactions) account for approximately 75% of all ADRs and in principle should be preventable [6]. Such statistics highlight the need for improved mechanistic understanding and clear risk avoidance strategies, both during the drug discovery/development process and in the clinic.

DDIs may occur at the metabolic level with impact on drug clearance at the site of gastrointestinal absorption, on plasma or tissue binding, by inhibition or induction of hepatic drug metabolizing enzymes, and on carrier mediated transport (including renal and hepatic uptake and biliary efflux) [7]. All these interactions may induce changes in the PK profiles of the drugs involved, resulting in altered PD response or promotion of a toxicological effect not associated with the primary pharmacology [8, 9]. Several clinically relevant and severe DDI cases, such as those resulting in the withdrawal of terfenadine and cerivastatin [10], have increased the focus on metabolism-based interactions, particularly those involving cytochrome P450 (CYP) enzymes. Indeed, CYP inhibition is reported to account for as much as half of all reported DDIs [11]. These metabolic interactions can occur whenever the metabolism of one “victim” drug is inhibited by the presence of another “perpetrator” drug. The resultant decrease in clearance of the victim drug leads to increased systemic concentrations that are pronounced if the inhibited pathway is the major route of elimination. Several CYP mediated DDIs have resulted in severe clinical reactions and fatalities in extreme cases [10, 12], so real emphasis is placed on the prediction and hence avoidance of DDIs in drug discovery [7, 13–15]. Indeed, there has been substantial academic and industrial effort directed toward the accurate prediction of DDIs with the aim of reducing their impact in the clinic by designing out such interactions.

DDIs as a potential source or contributor to drug attrition, either directly or indirectly, has contributed to the industry focus on this area. In the late 1980s poor drug metabolism and pharmacokinetics (DMPK) properties and toxicity were predominantly responsible for the discontinued development of new chemical entities (NCEs) during late stage clinical trials [16, 17]. In response to this, and to increase competitive advantage (e.g., by avoiding product interaction labels), the pharmaceutical industry has increasingly aligned DMPK to the early drug discovery phase. The failure of NCEs at a late stage is a cost the pharmaceutical industry can ill afford; it is estimated that the total preapproval costs for a new drug are in excess of 800 million U.S. dollars [18]. The industry has responded to both the clinical and

economic factors by front-loading DDI screens early in the discovery phase to predict the risk of DDI for NCEs. High throughput automated assays to evaluate the impact of CYP inhibition are now common as early as the hit identification stage [19]. Increased importance is placed on evaluation of the risk for NCEs participating in DDIs, as a victim and/or perpetrator of such interactions. However, despite the focus that DDIs and potentially associated ADRs have received over the last two decades, it was recently reported that there has been little reduction in their clinical impact [6, 17], although with a drug development time in excess of ten years it is perhaps too early to identify the impact of such DDI screens on these clinical statistics.

In the following sections several examples of DDIs are introduced to demonstrate the clinical consequences of CYP inhibition. The kinetics of reversible and irreversible inhibition are presented along with discussion of *in vitro* methodology. Accurate *in vitro* kinetic data is an absolute prerequisite for *in vitro-in vivo* extrapolation (IVIVE), and the current literature models for predicting clinical inhibition are presented. Overall, this chapter aims to summarize the current knowledge base and the complexities of CYP inhibition and to provide a guide for embarking on prediction of clinical DDIs with particular emphasis on the drug discovery setting.

## 22.2 CYTOCHROME P450 AND ITS ROLE IN DRUG-DRUG INTERACTIONS

The CYP family of enzymes are the primary facilitators of oxidative biotransformation in the liver and are responsible for the metabolism, at least in part, of the majority of human drugs [20]. These heme-containing enzymes are primarily situated in the lipid bilayer of the endoplasmic reticulum of hepatocytes, but are also located in extrahepatic tissues, including the gastrointestinal tract (GIT), kidney, and lung [21]. Fifty-seven different human CYP enzymes have been identified [22] and of these, 23 human isoforms are from the CYP1, CYP2, and CYP3 families that are recognized as important in xenobiotic metabolism [23]. However, it has been estimated that approximately 90% of human drug oxidations can be attributed to just five CYPs, namely, CYP1A2, CYP2C9, CYP2C19, CYP2D6, and CYP3A4/5 [24, 25]. CYP3A4 is the most abundant of these and can constitute up to 50% of total hepatic CYP content [26]. This enzyme is also the most prolific, with approximately 60% of all oxidative biotransformations being attributed to CYP3A4 [27]. The CYP family metabolize a very broad range of substrates, from endogenous steroids and vitamins to small drug molecules of all charge types [28]. For example, CYP3A4 metabolizes large lipophilic molecules including testosterone, erythromycin, and cyclosporine as well as smaller molecules like midazolam and nifedipine. CYP2C9 favors medium sized acidic or neutral molecules such as diclofenac and tienilic acid; CYP2D6 favors basic, lipophilic molecules such as dextromethorphan and metoprolol, while CYP1A2 has affinity for neutral planar molecules such as theophylline and phenacetin [28]. Since CYPs are the major enzyme family in xenobiotic/drug metabolism, they retain a central role in the mediation of clinically relevant DDIs.

### 22.2.1 Clinical Consequences of Cytochrome P450 Mediated Drug–Drug Interactions

There is a wealth of literature around CYP mediated DDIs. The most documented DDIs resulting in severe clinical adverse reactions are listed in Table 22.1. In the following section examples of drugs as either victims or perpetrators of DDIs will be presented with specific cases exemplifying (1) reversible and irreversible inhibition at the major CYPs and (2) the role of intestinal metabolism and enzyme polymorphisms within the patient population.

The antihistamine terfenadine, a victim of DDI, was withdrawn because of patient fatality following coadministration with ketoconazole [10]. Competitive inhibition of CYP3A4-dependent terfenadine methyl-hydroxylation by ketoconazole resulted in up to a 40-fold increase in the area under the plasma concentration–time curve (*AUC*) of terfenadine. This increased concentration manifested itself clinically as QT prolongation and torsades de pointes [37]. Other victims of DDI are simvastatin and lovastatin. Both are 3-hydroxy-3-methylglutaryl coenzyme A (HMG-CoA) reductase inhibitors, which although exceedingly well tolerated, incur a small risk of myopathy or potentially fatal rhabdomyolysis, particularly when coadministered with medications that increase their systemic exposure [44]. These statins both have a substantial CYP3A4 component to their metabolism, and large (hepatic and intestinal) first-pass metabolism makes them particularly susceptible to interactions via CYP3A4 inhibition. Intestinal metabolism may be effectively abolished in the presence of a potent oral CYP3A4 inhibitor, resulting in an increase in plasma concentration of up to threefold for drugs such as cyclosporine, midazolam, lovastatin, and simvastatin [44–46]. This was elegantly confirmed in clinical studies where midazolam was administered either via the oral or intravenous route alongside the CYP3A4 inhibitors itraconazole and fluconazole [46]. Following an intravenous dose the *AUC* of midazolam was increased 3.5-fold in the presence of itraconazole; however, the *AUC* increased sevenfold when midazolam was administered orally [46]. Clearly, for drugs with a high hepatic extraction, the impact of metabolic inhibition will give rise to large changes in oral exposure, compounded by any effect of inhibition on intestinal metabolism.

Another statin, cerivastatin, was anticipated to have less potential for CYP3A4 mediated DDIs than the other statins due to the additional role of CYP2C8 in its metabolism [47]. Prior to launch it was thought that cerivastatin would provide a safer alternative to simvastatin and lovastatin due to its alternative metabolic pathway. Indeed, interaction studies with erythromycin and itraconazole, two potent inhibitors of CYP3A4, have shown little impact on cerivastatin plasma levels [48–50]. However, coadministration with gemfibrozil increases the *AUC* of cerivastatin about sixfold [33]. This is assumed to be due to inhibition of CYP2C8 by gemfibrozil and its glucuronide metabolite. Interestingly, the glucuronide metabolite demonstrates a 20-fold increase in inhibitory potency at CYP2C8 over its parent ( $IC_{50}$  of 80  $\mu$ M and 4  $\mu$ M for gemfibrozil and its glucuronide metabolite, respectively [51]) and is actively concentrated in the liver [51, 52]. It is also worth noting that gemfibrozil also inhibits the OATP1B1-mediated hepatic uptake of statins and that this mechanism could well have contributed to this and indeed other gemfibrozil–statin interactions [52, 53]. During cerivastatin monotherapy, the incidence rate of rhabdomyolysis was 10–100 times higher than with the other statins and gemfibrozil

**TABLE 22.1 Examples of Marketed Drugs Involved in Clinically Significant Drug-Drug Interactions**

Predominant CYP Isoform	Victim Drug	Perpetrator Drug	Maximum Clinical AUC Increase of Victim Drug	Clinical Consequences	References
CYP1A2	Theophylline	Ciprofloxacin	Variable	Theophylline toxicity	29
	Tizanidine	Ciprofloxacin	10-fold	Excess hypotensive and sedative effects	30
CYP2C8	Tacrine	Fluvoxamine	~sixfold	Hepatotoxicity	31, 32
	Cerivastatin <sup>a</sup>	Gemfibrozil	sixfold	Rhabdomyolysis	33
CYP2C9	Tolbutamide	Sulfaphenazole	5.3-fold	Hypoglycemia	34
	Warfarin	Fluconazole	50% decrease in clearance	Enhanced hypoprothrombinemic effect	35
CYP2D6	Desipramine	Fluoxetine	~sixfold		31
	Metoprolol	Mibefradil <sup>a</sup>	4-5-fold	Effect on cardioselectivity	36
CYP3A4	Terfenadine <sup>a</sup>	Ketoconazole	~40-fold	Torsades des pointes	2, 10, 37, 38
	Lovastatin	Itraconazole	~20-fold	Rhabdomyolysis	39, 40
	Simvastatin	Mibefradil <sup>a</sup>	5-10-fold	Rhabdomyolysis	41
	Midazolam	Erythromycin	threefold	Excess sedation	42
	Cisapride <sup>a</sup>	Erythromycin	Variable	QT prolongation	12, 43

<sup>a</sup>Withdrawn from the market between 1990 and 2001.

greatly increased this risk; the number of rhabdomyolysis cases reported to the U.S. Food and Drug Administration with the gemfibrozil–cerivastatin combination was 533 [53]. As a result, cerivastatin was voluntarily withdrawn from the market in 2003.

Adverse events can be manifest without particularly large changes in the *AUC* of the victim drug. In epileptic patients stabilized on carbamazepine therapy, addition of the antidepressant agent viloxazine has resulted in side effects of carbamazepine intoxication (fatigue, dizziness, and ataxia). This is associated with an *AUC* elevation of carbamazepine (approximately twofold) and carbamazepine 10,11-epoxide (approximately 1.3-fold) [54]. The most likely mechanism for this interaction is CYP3A4 inhibition by viloxazine [55]. This example shows how a modest elevation can be detrimental if the therapeutic window of the victim drug is narrow; in this case resulting in an exaggerated pharmacological effect.

Mibefradil, a novel calcium antagonist for the treatment of hypertension and angina and a perpetrator of DDI, was voluntarily withdrawn from the market in 1997, less than a year after its launch [36]. Several severe adverse reactions were observed during mibefradil therapy in combination with various drugs including simvastatin and lovastatin [36, 41]. During its development, mibefradil had been identified as having inhibitory effects on CYP3A4 and CYP2D6, consequently coadministration was contraindicated with terfenadine and metoprolol (a CYP2D6 substrate) and the labeling carried a list of drugs that might require dose adjustment. When mibefradil was withdrawn, an additional complication arose; life-threatening reactions occurred if therapy was replaced with  $\beta$ -blockers or calcium channel antagonists within 5 days of discontinuing mibefradil therapy [36]. This was attributed to the long elimination half-life of mibefradil, but subsequently, it has been observed that mibefradil is a time-dependent inhibitor of CYP3A4 [56, 57] and also inhibits P-gp *in vitro* ( $K_i$  of 1.6  $\mu$ M [58]). The time-dependent nature of its CYP3A4 inhibition could explain the long “wash-out” period of mibefradil mediated inhibition. There are suggestions in the literature that some of the severe interactions observed with mibefradil could have been predicted from the *in vitro* and *in vivo* data available at the time [36, 45, 59].

The extent of DDIs can vary substantially within the population due to variability in enzyme expression. This variation is observed in the ciprofloxacin–theophylline interaction and has implications for the severity of the clinical outcome. The inhibitory effect of ciprofloxacin at CYP1A2 can result in varying changes in theophylline levels across a patient population. In sensitive patients, theophylline clearance can decrease significantly, increasing the risk of theophylline related toxicity [29, 60]. In another case, ciprofloxacin inhibits the CYP1A2 mediated metabolism of tizanidine, resulting in *AUC* increases of up to 10-fold, dangerously escalating its hypotensive and sedative effects [30]. CYP1A2 is the primary clearance mechanism for both theophylline and tizanidine [61, 62], and there is considerable interindividual variability in the *in vivo* metabolic activity of CYP1A2 (demonstrated by caffeine clearance [63]); therefore, inhibition results in significant changes in plasma levels in sensitive patients only.

As is the case for all polymorphic CYPs, the CYP2D6 related DDIs are complicated by the genetic polymorphisms observed for this enzyme (between 5% and 10% of the Caucasian population are CYP2D6 poor metabolizers). Investigation of potential DDIs observed via CYP2D6 inhibition may be confounded by the pres-

ence of individuals with the poor metabolizer phenotype within the study group [64]. It is evident that poor metabolizers are unlikely to suffer DDIs through CYP2D6 inhibition and therefore will not display the same magnitude of interaction predicted from *in vitro* studies [6]. CYP2D6 is inhibited by antipsychotic drugs and the propensity for interactions is increased for this patient population as psychopharmacologic medications are often taken for long periods of time and commonly coprescribed with other medications [65]. For example, the CYP2D6 inhibitor fluoxetine has been demonstrated to substantially increase the plasma levels of desipramine (between four- and sixfold) [31].

The clinical interactions presented here highlight the need to assess an NCE's involvement in potential DDIs from *both* perspectives of inhibition, either as a victim drug where metabolism is altered by a coadministered inhibitor or as the perpetrator of DDI. The cases also highlight the importance of fully defining the elimination and metabolic profile of an NCE: reaction phenotyping can lend greater understanding to potential interactions as can an understanding of the target patient population and likely coadministered drugs.

### 22.2.2 Mechanisms of Cytochrome P450 Inhibition

There are different mechanisms through which CYP inhibition may occur, all of which are a function of the metabolic capacity of the enzyme. In nearly all cases of clinically relevant *reversible* inhibition, the perpetrator is a substrate competing for the same CYP mediated metabolism. Simple Michaelis–Menten kinetics often satisfactorily describes CYP metabolism *in vitro*, provided that the enzyme and substrate are in thermodynamic equilibrium.

**Enzyme Kinetics** In the reaction scheme shown in Fig. 22.1, a steady-state equilibrium is assumed to be reached (the ES complex rapidly reaches a constant value such that  $d[\text{ES}]/dt = 0$ ). This assumption is key to the derivation of the Michaelis–Menten equation, which underpins not only prediction of metabolic clearance from *in vitro* data but also prediction of DDIs (since the prediction of DDI magnitude from *in vitro* data always starts with a modification of the Michaelis–Menten equation). The reason that the investigator needs to be aware of this is that experimental conditions are important in making this assumption. For  $d[\text{ES}]/dt = 0$ , the reverse reaction rate should be negligible and the formation of the [ES] complex should not significantly alter the substrate concentration [S]. In other words, the substrate concentration should greatly exceed the enzyme concentration and the substrate concentration should not be significantly depleted during the incubation (less than 20% metabolism). To this end, it is noteworthy that in a typical human liver microsome (HLM) preparation, CYP3A4 concentration may be as much as 0.25  $\mu\text{M}$  when the incubation contains 1 mg/mL HLM.

Equations 22.1 and 22.2 are the Michaelis–Menten equation (22.1) and its rearrangement (22.2), where  $v$  is the observed rate of metabolite formation,  $V_{\text{max}}$  is the maximum rate of this metabolite formation,  $K_m$  is the Michaelis–Menten constant representing the enzyme:substrate affinity and is the concentration that supports half the maximum rate of metabolite formation, and [S] is the substrate concentration at the enzyme active site. The derivation of these equations is shown in Appendix A.





**FIGURE 22.1** Schematic representing Michaelis–Menten kinetics, where E is enzyme, S is substrate,  $k_{+1}$  is the association rate constant describing ES formation from E + S,  $k_{-1}$  is the dissociation rate constant from ES to E + S, ES is substrate-bound enzyme, P is the product of enzyme, and  $k_2$  is the dissociation rate constant from ES to E + P.

$$v = \frac{V_{\max} \cdot [S]}{K_m + [S]} \quad (22.1)$$

$$= \frac{V_{\max}}{1 + K_m/[S]} \quad (22.2)$$

Only simple one-site binding can be accurately described by hyperbolic Michaelis–Menten kinetics but there is substantial literature evidence that atypical kinetics and multisite binding exist for the CYP enzymes. This is most clear for CYP3A4 [66] but is also observed for CYP2C9 [67, 68]. Positive or negative cooperative effects, where a bound substrate molecule influences the binding of a second molecule, have implications in terms of probe substrate selection and make full kinetic characterizations complex. It must be noted, however, that while cooperativity is clearly demonstrable *in vitro*, there is little evidence of its influence *in vivo* [69, 70]. Only simple one-site models of inhibition will be discussed here and detailed explanation of atypical (or sigmoidal) kinetics in inhibition can be found elsewhere in the scientific literature [14, 66, 71–73].

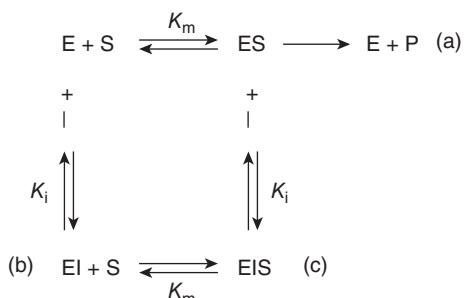
**Reversible Inhibition** Having explained the basis of simple CYP kinetics, the details of inhibition can be described. Reversible CYP inhibition can be categorized into three main forms: competitive, noncompetitive, and uncompetitive. Figure 22.2 describes each type of inhibition and the rate equations associated with them are presented below. In competitive inhibition the substrate and inhibitor compete for the same binding site within the enzyme; inhibition therefore decreases as the enzyme becomes saturated with substrate. Competitive inhibitors interfere with substrate binding so as to raise  $K_m$  by a factor of  $[I]/K_i$  without affecting  $V_{\max}$  (i.e., the affinity of the substrate for the enzyme,  $K_m$ , is decreased by a factor dependent in magnitude on the ratio of inhibitor concentration and the affinity of the inhibitor for the enzyme,  $K_i$  (Eq. 22.3)):

$$v = \frac{V_{\max} \cdot [S]}{K_m(1 + [I]/K_i) + [S]} \quad (22.3)$$

It is common to relate the potency of competitive inhibition to  $IC_{50}$  (the concentration of inhibitor for which the reaction rate is suppressed by 50%). Equation 22.4 is derived from Eq. 22.3 when  $v$  becomes  $v/2$  and  $I$  becomes  $IC_{50}$ .

$$IC_{50} = K_i(1 + [S]/K_m) \quad (22.4)$$





**FIGURE 22.2** Schematic of competitive inhibition, representing Michaelis–Menten kinetics. Pathway (a), competitive inhibition; pathway (b), uncompetitive inhibition; pathway (c), noncompetitive inhibition. E is enzyme, S is substrate, ES is substrate-bound enzyme, P is the product of enzyme and substrate, I is inhibitor, EI is inhibitor-bound enzyme, EIS is inhibitor- and substrate-bound enzyme,  $K_m$  is the Michaelis–Menten constant, and  $K_i$  is the inhibition constant. Note that  $K_m$  can be described as  $(k_2 + k_{-1}/k_{+1})$ —see Appendix A. However, if it is assumed that rapid formation of the ES complex occurs when the enzyme and substrate are mixed;  $k_{+1} \gg k_2$  and  $K_m$  can be described as  $k_{-1}/k_{+1}$ .

In noncompetitive inhibition the inhibitor binds at a site distinct from the substrate binding site and can interact with either the substrate-free or the substrate-bound form of the enzyme. It is assumed that the inhibitor binds to both states of the enzyme with the same affinity and so the degree of inhibition is independent of substrate concentration. The binding affinity ( $K_m$ ) is unaffected but the inhibitor acts to remove active enzyme and consequently the maximal reaction velocity ( $V_{\max}$ ) is decreased. The metabolic rate of noncompetitive inhibition is described by

$$v = \frac{V_{\max}/(1 + [I]/K_i) \cdot [S]}{K_m + [S]} \quad (22.5)$$

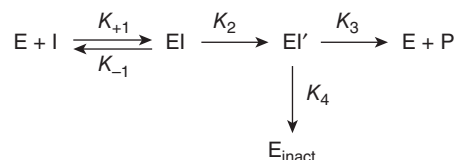
The substrate-independent nature of inhibition is further demonstrated by the  $IC_{50}$ , which is equivalent to the inhibitor constant  $K_i$  ( $IC_{50} = K_i$ ), derived from Eq. 22.5 when  $v$  becomes  $v/2$  and  $[I]$  becomes  $IC_{50}$ .

In uncompetitive inhibition (which is extremely rare for CYP metabolism [71]), only the substrate-bound form of the enzyme is open to inhibitor binding; the inhibitor cannot bind to free enzyme and so the degree of inhibition will increase with substrate concentration. The metabolic rate in the presence of an uncompetitive inhibitor can be described by Eq. 22.6 and the corresponding  $IC_{50}$  by Eq. 22.7:

$$v = \frac{V_{\max}/(1 + [I]/K_i) \cdot [S]}{K_m(1 + [I]/K_i) + [S]} \quad (22.6)$$

$$IC_{50} = K_i(1 + K_m/[S]) \quad (22.7)$$

**Mechanism-Based Inhibition** Irreversible inhibition or mechanism-based inactivation is an unusual occurrence with most enzymes. It is, however, observed in reactions catalyzed by CYP in somewhat higher frequency [74]. In contrast with



**FIGURE 22.3** Schematic of mechanism-based inhibition, where E is enzyme, I is inhibitor, EI is inhibitor-bound enzyme, EI' is the initial complex formed between metabolic intermediate and enzyme, P is the product of enzyme and inhibitor complex that escapes reactive binding in the active site,  $E_{\text{inactive}}$  is the inactive enzyme with irreversibly bound metabolic intermediate,  $k_{+1}$  is the association rate constant for E + I forming EI,  $k_2$  is the association rate constant for EI forming EI',  $k_3$  is the dissociation rate constant for EI' to E + P, and  $k_4$  is the rate constant from EI' to  $E_{\text{inact}}$ .

reversible inhibition, the effects of mechanism-based inhibition (MBI) are more profound after multiple dosing and the recovery period, typically several days, is independent of continued exposure to the drug [75, 76]. This MBI transpires when a CYP is quasi-irreversibly or irreversibly inhibited by reactive intermediates formed during the metabolic process. The inhibitory intermediates may form covalently bound adducts to the CYP heme or apoprotein; alternatively, they may form a metabolic inhibitory complex (MIC) in the active site. This MBI is characterized by NADPH-, concentration-, and time-dependent inactivation. Additionally, the inactivation rate is diminished in the presence of a competing substrate, enzyme activity is not restored by dialysis or filtration (as inhibitor is covalently bound), and *de novo* synthesis of protein is required to recover metabolic activity [77]. Figure 22.3 shows a schematic representation of this inhibition [78] and the inhibition for the initial rate of enzyme deactivation can be described as follows.

From Fig. 22.3 it can be seen that the change in concentration over time of inhibitor-bound enzyme, [EI], the initial complex formed between metabolic intermediate and enzyme, [EI'], and the inactive enzyme with irreversibly bound metabolic intermediate, [ $E_{\text{inact}}$ ], respectively can be expressed as

$$d[\text{EI}]/dt = k_{+1}[\text{I}] \cdot [\text{E}] - (k_{-1} + k_2)[\text{EI}] \quad (22.8)$$

$$d[\text{EI}']/dt = k_2[\text{EI}] - (k_3 + k_4)[\text{EI}'] \quad (22.9)$$

$$d[E_{\text{inact}}]/dt = k_4[\text{EI}'] \quad (22.10)$$

where [E] is the concentration of the unbound active enzyme and [I] is the concentration of the mechanism-based inhibitor.

The parameters  $k_{\text{inact}}$  (the maximal inactivation rate constant) and  $K_I$  (the inhibitor concentration that supports half the maximal rate of inactivation) are described by Eqs. 22.11 and 22.12, respectively, using the association constants shown in Fig. 22.3.

$$k_{\text{inact}} = k_2 k_4 / (k_2 + k_3 + k_4) \quad (22.11)$$

$$K_I = \frac{(k_3 + k_4)(k_{-1} + k_2)}{(k_2 + k_3 + k_4)k_1} \quad (22.12)$$

The apparent inactivation rate constant of the enzyme,  $k_{\text{obs}}$ , is described by

$$k_{\text{obs}} = \frac{k_{\text{inact}}[\text{I}]}{[\text{I}] + K_{\text{I}}} \quad (22.13)$$

The derivation of these equations is included in Appendix B to show the mathematical assumptions that must be made to allow such parameters to be defined experimentally.

The partition ratio ( $r$ ) is a useful additional parameter to  $k_{\text{inact}}$  and  $K_{\text{I}}$ . It is defined as the number of moles of reactive metabolite escaping the enzyme without inactivating it (product) per number of moles of reactive metabolite inactivating the enzyme (trapped product) ( $r = k_3/k_4$ ; Fig. 22.3). As the partition ratio approaches 0, inhibition is more efficient; for every cycle of the enzyme and substrate that results in product, a cycle also results in inactivated enzyme. For a potent inhibitor, the reactive species is less likely to escape the active site without adduct formation. While the partition ratio may be helpful in ranking compounds in order of inhibitory efficiency (as is the  $k_{\text{inact}}/K_{\text{I}}$  ratio), any use beyond this is limited.

## 22.3 ASSESSMENT OF CYTOCHROME P450 INHIBITION *IN VITRO*

Over the last decade there have been tremendous advances in the understanding of both the theoretical and technical aspects of generating *in vitro* kinetic data to engender predictions of DDIs [7, 11, 13, 21, 71, 79, 80]. The advantages of using *in vitro* assays to study inhibition include allowing inexpensive and rapid assessment of inhibition potential and consequent evaluation of DDI risk, facilitating detailed kinetic analysis of drugs known to cause DDIs in the clinic, and helping to further develop our understanding of these interactions and ultimately striving toward the avoidance of DDIs for NCEs. However, despite the undoubted usefulness of *in vitro* models, there are limitations, primarily due to the fact that, to date, there is no consensus toward a model that quantitatively predicts interactions in humans from *in vitro* data. Indeed, there is no single method for generating *in vitro* data. Despite the lack of consensus, there are some fundamentally common elements to the existing methodologies and these form the basis of the remainder of this chapter. Prior to modeling data, accurate *in vitro* assessments must be made. There are many parameters to consider when embarking on inhibition studies to generate good *in vitro* data, the key elements of which are discussed here.

### 22.3.1 Enzyme Systems

Historically, the majority of inhibition studies have been conducted using subcellular fractions of liver. Clear species differences exist in hepatic enzyme content and function [81]. In most cases a homolog of the human enzyme is present in the pre-clinical species. For example, CYP3A1 and CYP3A2 are the major 3A isoforms in the rat and share substrate crossover to human CYP3A4 and CYP3A5. Likewise CYP2C11 in rodent liver is homologous to CYP2C9 in humans. However, the species-specific isoforms of CYP1A, CYP2C, CYP2D, and CYP3A do show appreciable interspecies differences in terms of catalytic activity [81]. While laboratory

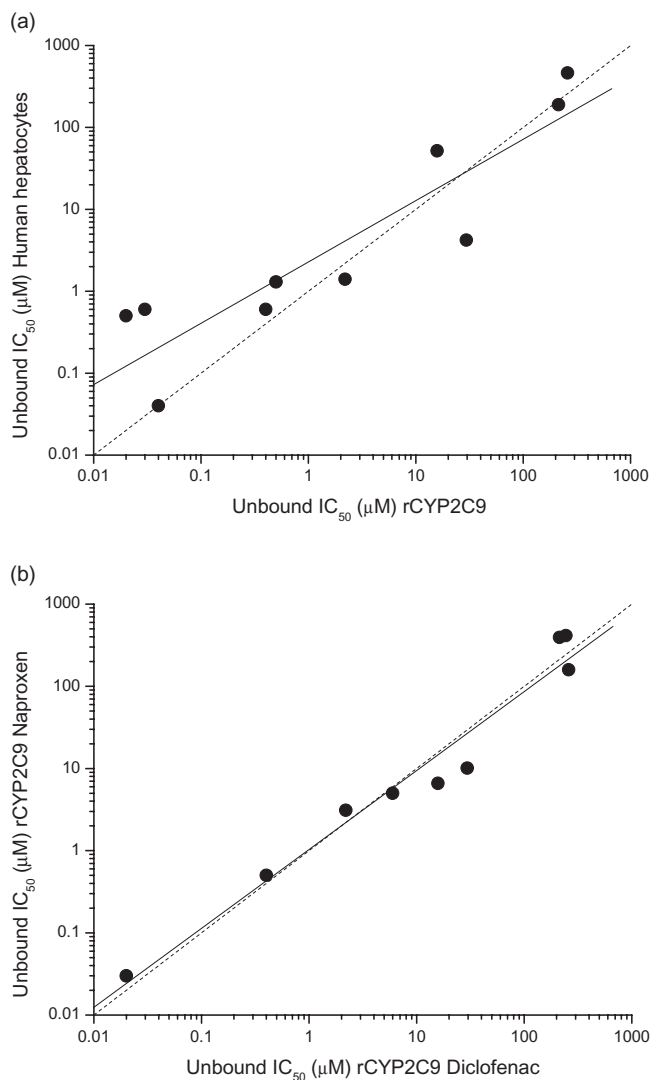
species can be a useful tool for investigating pharmacokinetic species differences and for providing confidence in IVIVE, clinical DDI potential must be investigated *in vitro* using human derived tissues and/or enzymes because of such species differences.

Human liver microsomes (HLMs) are prepared from liver homogenates by differential ultracentrifugation, first removing the cytosolic (S9) fraction, then with further ultracentrifugation collecting the subcellular fraction, which includes membrane-bound enzymes [82]. The utility of HLMs is clear; they can be isolated from a relatively small amount of human liver, can be stored almost indefinitely at  $-80^{\circ}\text{C}$ , contain functional CYPs (and other enzymes such as UDP-glucuronosyl-transferases and flavin monooxygenases), and can be used as a single donor or pooled to provide a representation of enzyme expression in the population. However, due to the presence of multiple CYP isoforms with overlapping substrate selectivity, kinetic investigations must be thorough and carefully planned as inhibitors may be substrates for more than one CYP isoform. Consequently, for CYP inhibition studies using HLMs, the use of selective probe reactions is paramount and often relies on careful adjustment of substrate concentration to ensure selectivity.

In the last decade, the use of recombinant human CYP (rCYP) proteins, expressed in yeast, baculovirus-infected insect cells, or *Escherichia coli*, has become widespread and indeed now forms the backbone of screening efforts in the pharmaceutical industry. There are considerable advantages to using a single rCYP enzyme in an assay as it allows manipulation of substrates and inhibitors without the complications of enzyme selectivity. A notable example of this is CYP3A5, which has broad substrate crossover with CYP3A4, making it difficult to characterize reactions that are specific to or of equal affinity at CYP3A5 with respect to CYP3A4 [83]. A sometimes-perceived disadvantage of rCYP is the increased enzyme activity per unit of membrane protein compared to that of HLM. However, the high CYP expression levels give rise to low incubational protein content and low non-specific substrate and inhibitor binding, resulting in more accurate estimations of unbound  $K_{+1}$  values. Finally, the utility of rCYP in reaction phenotyping and inhibition studies has been well validated [84–86] and the versatility of rCYPs lends them to enhanced throughput screening.

Isolated human hepatocytes are an alternative *in vitro* tool and can be used to add refinement to data generated using subcellular fractions, as the impact of drug transporter activity on  $K_i$ , accounting for intracellular free drug concentrations, can be investigated. As whole cells, hepatocytes express not only all membrane-bound enzymes but also hepatic cytosolic enzymes; moreover, they contain the necessary cofactors. Under the correct conditions, human hepatocytes retain some functionality for many days, express the transcriptional machinery to regulate enzymes, and contain functional proteins for active drug transport across membranes [87]. The utility of hepatocytes for investigation of metabolic stability [88–90] and DDIs mediated by CYP induction [87, 91, 92] is well established. More recently, hepatocytes have been used to describe the impact of reversible [68, 93, 94] and time-dependent CYP inhibition *in vitro* [95, 96]. Primary hepatocytes in culture provide the closest *in vitro* model to human liver and may offer advantages when predicting clinical DDI. Metabolites of one pathway may lead to inhibition of another; this phenomenon is indiscernible using single rCYPs and exemplifies the usefulness of hepatocytes [97].

There is little published comparative data of inhibition constants generated in rCYPs and human hepatocytes. Data generated in this laboratory demonstrates  $IC_{50,apparent}$  values generated in human hepatocytes, using probe substrates specific to the CYP isoform, were systematically higher than those determined using the respective rCYP [68]. This was predominantly the result of greater nonspecific binding in hepatocytes compared to rCYPs, as for the majority of compounds tested there was a good concordance between the respective  $IC_{50,unbound}$  values (Fig. 22.4a).



**FIGURE 22.4** (a) Relationship between  $IC_{50,unbound}$  for CYP2C9 in human hepatocytes and rCYPs. The dotted line is unity. The solid line indicates linear regression of the data ( $r^2 = 0.88$ ,  $p < 0.0001$ ) (b)  $IC_{50,unbound}$  comparisons using naproxen and diclofenac as substrates for rCYP2C9. The dotted line is unity. The solid line indicates linear regression of the data ( $r^2 = 0.77$ ,  $p < 0.002$ ). (Part (b) from Ref. 68.)

It can be speculated that the impact of hepatic transporters may explain some discrepancies in  $IC_{50,unbound}$  between rCYP and hepatocytes as compound can be actively effluxed from or taken up into the hepatocyte, altering the concentration of drug exposed to the enzyme [97]. Additionally, the impact of time-dependent CYP inhibition in primary cultures of human hepatocytes has been evaluated, again using a cocktail of CYP substrates, which allows the maximum amount of data to be extracted from this limited resource [95].  $K_I$  and  $k_{inact}$  values were estimated from nonlinear regression analysis and compared with values obtained from rCYPs and HLMs (Table 22.2). The values generated are also comparable to several literature reports of  $K_I$  and  $k_{inact}$  estimates using rCYP and HLMs and the results from our laboratory using rCYP, HLMs, and human hepatocytes were consistent with these

**TABLE 22.2 Kinetic Parameters of Time-Dependent Inhibitors in Different *In Vitro* Matrices**

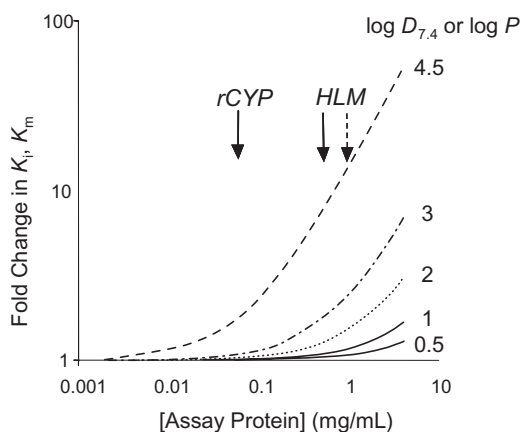
Compound	CYP Isoform	Enzyme Source	$K_I$ ( $\mu$ M)	$k_{inact}$ ( $min^{-1}$ )	Reference
Tienilic acid	CYP2C9	rCYP2C9	2	0.19	95
		rCYP2C9 (2C10)	4	0.20	99
		Human hepatocytes	2	0.05	95
AZ1	CYP2C9	rCYP2C9	30	0.02	95
		Human hepatocytes	19	0.02	95
Fluoxetine	CYP2C19	rCYP2C19	0.4	0.5	95
		HLM	8 <sup>a</sup> (0.8)	0.03	95
		Human hepatocytes	0.2	0.04	95
	CYP3A4	rCYP3A4	2	0.03	95
		HLM	5 <sup>a</sup> (0.5)	0.01	95
		HLM	5 <sup>a</sup>	0.02	100
		Human hepatocytes	1	0.01	95
Erythromycin	CYP3A4	rCYP3A4	9	0.12	98
		rCYP3A4	5	0.12	101
		HLM	16 <sup>a</sup>	0.07	102
		HLM	82 <sup>a</sup>	0.07	103
		HLM	13 <sup>a</sup>	0.02	104
		HLM	10 <sup>a</sup>	0.08	105
		HLM	11 <sup>a</sup>	0.05	106
		HLM	15 <sup>a</sup>	0.07	96
		Human hepatocytes	11	0.07	95
		Troleandomycin	CYP3A4	rCYP3A4	0.3
rCYP3A4	0.2			0.15	101
HLMs	2 <sup>a</sup>			0.03	96
Human hepatocytes	0.4			0.05	95

<sup>a</sup>Apparent  $K_I$  values. All other values are *unbound*  $K_I$  estimates.

Source: From Ref. 95.

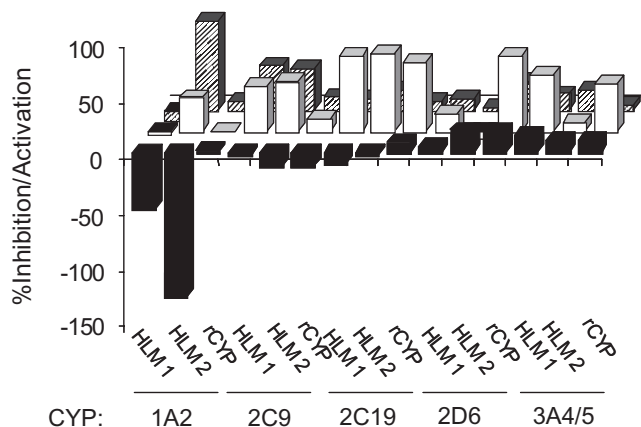
reports. Indeed, the CYP3A4-dependent  $K_I$  and  $k_{\text{inact}}$  values in human hepatocytes are comparable to values determined in rCYP3A4 and HLMs (Table 22.2) when corrected for nonspecific binding in the incubation.

Finally, an element of interlaboratory variability can be expected for different tissue preparations. However, an interlaboratory comparison found the rank order of five HLM preparations was conserved across five laboratories, all using different methodologies, despite differences in absolute values (based on protein content, CYP content, and activity) [107]. Irrespective of the enzyme system used, there are some common factors that must be considered for all *in vitro* inhibition incubations. The assay must be designed such that linear initial rate conditions for the reaction under investigation are assured. To make certain of this, preliminary experiments will be required to ensure product formation is linear with respect to both time and enzyme (protein) concentration [72]. Additionally, protein concentration should be kept as low as reasonably possible to reduce the impact of nonspecific binding [108]. As discussed with respect to hepatocytes and rCYPs, the fraction unbound in the incubation ( $f_{\text{uinc}}$ ) can have a pronounced effect on the apparent  $K_i$ ,  $K_I$ , and  $IC_{50}$  determined (Fig. 22.5). This is especially the case for lipophilic inhibitors such as ketoconazole, for which the apparent  $IC_{50}$  can vary between 0.005 and 0.3  $\mu\text{M}$  [109]. It is important that such drug binding terms be incorporated by default when making IVIVE to improve the accuracy of *in vitro* data [88], and the importance of considering the fraction unbound in an *in vitro* incubation ( $f_{\text{uinc}}$ ) when relating *in vitro* and *in vivo* data was proposed many years ago by Gillette (1963). The term itself can readily be determined and indeed predicted with accuracy from physicochemical properties [88, 110, 111]. Another important factor is the amount of solvent vehicle, which must be minimized ideally to 1% (or less) of the final incubation volume, as activation or inhibition of CYP activity can be manifest through solvent effects alone (Fig. 22.6). Acetonitrile appears the most favored solvent, having minimal inhibitory effects, despite it being able to activate CYP1A2; although DMSO is commonly used to ensure adequate NCE solubility, it can result in inhibition of metabolism, especially for basic CYP3A4 substrates [88, 112]. Solvent effects



**FIGURE 22.5** Simulated effects of microsomal or hepatocyte concentration on  $K_i$  or  $K_m$  as a function of  $\log P$  or  $\log D_{7.4}$ -dependent nonspecific binding. (From Ref. 88.)





**FIGURE 22.6** Overview of potential effects of commonly used organic solvents (1%, v/v) on CYP activities with recombinant proteins and HLMs (solid bars = acetonitrile, open bars = DMSO, hatched bars = methanol). (From Ref. 112.)

can also be selective for probe and enzyme source [113], so it is advisable to check the effects of an alternative solvent for key studies.

### 22.3.2 Probe Substrate Selection

An important component of *in vitro* inhibition study design is the selection of appropriate probe substrates. Nonselective probes in HLM or human hepatocyte assays can result in hybrid inhibition measurements where multiple enzymes (with varying sensitivity to the inhibitor) can contribute to the reaction under investigation [14]. To avoid such complications, a selective marker substrate is required for each CYP isoform being assessed. These probes should meet the following criteria: possess high affinity for the specific enzyme; be metabolized to a specific product metabolite; provide the ability to detect the product separately from the substrate (using the analytical system of choice); and, finally, the turnover of the probe substrate should be optimized in each enzyme system and be linear with respect to time resulting in less than 20% loss of parent compound to maintain constant reaction velocity throughout the experiment [106, 114]. Industry representatives and regulatory bodies have made recommendations for preferred probe substrates for each CYP isoform (Table 22.3), many of which have the additional benefit of being clinically relevant [117, 120]. These substrates have been rigorously validated for human CYPs [115] and the usefulness and caveats associated with them are well documented [14].

An advantage to the availability of highly selective probes is that, coupled with the improved analytical sensitivity afforded by advances in detection systems such as LC/MS-MS, a cassette of substrates can be used to evaluate the inhibitory potential of a compound upon multiple CYP isoforms in a single incubation. The usefulness of cassette assays derives from increasing the throughput of inhibition screening, although they do require careful optimization and validation to ensure selectivity is maintained [116, 118, 121].

**TABLE 22.3 Specific Probe Substrate Reactions for Individual CYP Enzymes**

CYP	Substrate	Reaction	Analytical Method	Comments <sup>a</sup>	References
CYP1A2	Phenacetin <sup>b</sup> Ethoxyresorufin (EROD)	O-deethylation O-deethylation	LC/MS-MS Fluorescence	High specificity at <100 μM ( $K_m \approx 50 \mu\text{M}$ ) $K_m \approx 0.2 \mu\text{M}$ ; not clinically relevant	115, 116 117
CYP2A6	Caffeine Coumarin	N3-demethylation 7-Hydroxylation	UV-HPLC LC/MS-MS	Low affinity $K_m \approx 600 \mu\text{M}$ High specificity $K_m \approx 0.5 \mu\text{M}$	117 115, 117
CYP2B6	Bupropion <sup>b</sup> S-Mephenytoin	Hydroxylation N-demethylation	Fluorescence LC/MS-MS LC/MS-MS	High specificity $K_m \approx 90 \mu\text{M}$ Not CYP2B6 specific $K_m \approx 2000 \mu\text{M}$ (CYP2C9 high affinity $K_m \approx 150 \mu\text{M}$ )	115, 118 14
CYP2C8	Amodiaquine <sup>b</sup>	N-deethylation	LC/MS-MS	High specificity $K_m \approx 2 \mu\text{M}$	115, 118
CYP2C9	Diclofenac <sup>b</sup> Tolbutamide	4'-Hydroxylation Hydroxylation	LC/MS-MS LC/MS-MS	Good specificity $K_m \approx 2 \mu\text{M}$ (some CYP2C19 $K_m \approx 80 \mu\text{M}$ ) $K_m \approx 200 \mu\text{M}$ (some CYP2C19 $K_m \approx 500 \mu\text{M}$ )	115, 116 115, 119
CYP2C19	S-Mephenytoin <sup>b</sup>	4'-Hydroxylation	Radiometric LC/MS-MS	High specificity $K_m \approx 40 \mu\text{M}$	115, 116
CYP2D6	Bufuralol <sup>b</sup>	1-Hydroxylation	LC/MS-MS	High affinity component for CYP2D6 $K_m \approx 5 \mu\text{M}$ (CYP2C19 and CYP2C9 at high concentrations)	116
CYP2E1	Dextromethorphan	O-demethylation	LC/MS-MS	Questionable specificity $K_m \approx 700 \mu\text{M}$	115
CYP3A4	Chlorzoxazone Midazolam <sup>b</sup> Testosterone	7-Hydroxylation 1'-Hydroxylation 6β-Hydroxylation	LC/MS-MS LC/MS-MS UV-HPLC	High specificity benzodiazepine $K_m \approx 3 \mu\text{M}$ High specificity steroid $K_m \approx 50 \mu\text{M}$	115, 117 56, 115, 116 56, 115, 117
	Nifedipine	Oxidations	Radiometric UV-HPLC	High specificity dihydropyridine $K_m \approx 50 \mu\text{M}$	87

<sup>a</sup>Comments from Ref. 14.

<sup>b</sup>Primary Substrate used in this laboratory.

An additional aspect to be considered for *in vitro* assays is that CYP inhibition can be substrate dependent, particularly for CYP3A4 [122]. CYP3A4 is thought to bind substrates and inhibitors in multiple modes and/or binding sites [66, 123]. As a result, inhibitory interactions observed with one probe substrate may not be representative of those observed with other substrates [19]. This is due to the cooperative nature of substrate–inhibitor interactions within the active site [124]. Kenworthy et al. [122] classified 10 CYP3A4 substrates into three distinct groups: a benzodiazepine group (including diazepam, midazolam, triazolam, and dextromethorphan), a large molecular weight group (including testosterone, erythromycin, and cyclosporine), and finally nifedipine and benzyloxyresorufin (BROD), which fit in neither of the previous groups and are distinct from each other. These substrate groups were classified according to analysis of their behavior as probe reactions for the effect of 34 compounds on CYP3A4 mediated metabolism, with different inhibitor profiles being observed with different substrate groups. For example, the  $IC_{50}$  for inhibition of CYP3A4 mediated nifedipine oxidation by haloperidol was 0.1  $\mu\text{M}$ , while for dextromethorphan N-demethylation it was  $>100 \mu\text{M}$  [122]. It is postulated that haloperidol is able to inhibit CYP3A4 by binding at more than one site, with the effect being dependent on the substrate, reflecting the cooperative nature of CYP3A4.

There is also increasing evidence that similar multisite kinetics and substrate-dependent interactions are observed for CYP2C9 *in vitro* [67–69]. Therefore, while there is little evidence of auto-/heteroactivation *in vivo* in humans [69, 70, 125], it is good practice to evaluate the inhibitory potential of NCEs against CYP3A4 and CYP2C9 using at least two structurally dissimilar probe substrates, to afford a better predictive quality to the *in vitro* data as *in vivo* effects can be substrate dependent. In our laboratory, routine screens are conducted with drug-like probes for both CYP3A4 (midazolam) and CYP2C9 (diclofenac); for key compounds, a second and potentially a third probe is used to fully evaluate inhibition of these isoforms, for example, erythromycin and naproxen for CYP3A4 and CYP2C9, respectively. Figure 22.4b shows the effect of using either diclofenac or naproxen as probe substrate to measure the  $IC_{50}$  for a range of compounds demonstrating CYP2C9 inhibition. On average, it was found that the  $IC_{50,\text{apparent}}$  generated in an assay using naproxen as a substrate could be approximately 1.5-fold times that using diclofenac. This could be corrected from a consideration of  $IC_{50,\text{unbound}}$  [68] (see Fig. 22.4b).

### 22.3.3 Analysis Methods

Several analytical methods are available for determining the impact of inhibition on a probe substrate. The selection of an analytical endpoint depends on the probe under investigation, the availability of the detection system, and the throughput required. LC/MS-MS is probably the system of choice for the pharmaceutical industry due to the selectivity and sensitivity it affords [115, 126]. This, combined with the fact that LC/MS-MS is the backbone of analysis within DMPK laboratories and therefore readily available, has driven its increased use in CYP inhibition assays. There are several LC/MS-MS methods that enable routine analysis of metabolites generated in microsomal and rCYP incubations, and importantly LC/MS-MS facilitates the use of clinically relevant probes. Additionally, the sensitivity afforded by LC/MS-MS has allowed multiple reaction monitoring of substrate cassettes, facilitating the investigation of test inhibitors for multiple CYP isoforms [20, 116, 121]. In

our laboratory we routinely screen for CYP inhibition using a substrate cassette and a rCYP cocktail; the details of this and other assays used are shown in Table 22.4.

An alternative to LC/MS-MS is HPLC-radioflow. This allows separation of metabolites of  $^{14}\text{C}$  or  $^3\text{H}$  labeled probe substrates. It is, however, a time-consuming technique requiring meticulous optimization of conditions and long run times, and LC/MS-MS is a more rapid alternative [115, 116, 118, 126]. A higher throughput radiometric endpoint assay is available and is flexible, sensitive, robust, and free from analytical interference [119, 127]. This laboratory uses a range of automated assays with [*O*- or *N*-methyl- $^{14}\text{C}$ ]-substrates, which liberate [ $^{14}\text{C}$ ]-formaldehyde as the product of enzyme-specific oxidative demethylation [119]. However, the extra considerations involved with the use of radioactivity, such as the highly regulated disposal of isotopes and additional safety requirements, does constrain their use perhaps to second substrate selection.

Fluorometric assays are commonly used for CYP inhibition studies, especially in early drug discovery, as their speed and cost advantages lend them to enhanced throughput (compared to detailed manual assays) or true HTS (miniaturized assays employing robotics to utilize 96-, 384-, or 1536-well plate formats with rapid endpoints). Indeed, they are widely used throughout the pharmaceutical industry. Fluorescent-based assays are very amenable to multiwell plate assays; but despite their utility in rapidly generating large sets of inhibition data there are drawbacks. There can be fluorescent interference from the test compound or its product and issues with fluorescent quenching as well as lack of probe molecule specificity for an individual CYP [20]. The nonselective nature of fluorescent probes can, of course, be overcome by the use of single rCYPs. A potential concern for these assays is the non-drug-like qualities of most fluorescent probes, particularly in terms of clinically relevant CYP interactions. A detailed study evaluating inhibition data generated using fluorescent probes compared to that of conventional drug probes suggested that in general only a weak correlation between the two endpoints existed, with both endpoints not detecting inhibition by a significant number of compounds [19, 88]. Cohen et al. [19] suggest that concerns over fluorogenic probes not resembling drug molecules are unfounded and that it is the architecture of the CYP active site, permitting binding of structurally diverse molecules, that underlies interprobe differences in  $\text{IC}_{50}$  values, be they fluorescent or not. In our laboratory, fluorometric assays have been widely replaced by LC/MS-MS-based assays but are still used as an alternative substrate assay to cross-check inhibition data for CYP3A4 and CYP2C9, as it is prudent to assess inhibition in detailed studies using multiple probes and alternative endpoints as least for these enzymes.

Finally, luminescence-based CYP selective probes have recently become commercially available, such as those used in the Promega P450-Glo™ assays. These assays provide a luminescent method for measuring CYP activity from recombinant and native sources. A conventional CYP reaction is performed by incubating the CYP with a luminogenic CYP substrate and NADPH regeneration system. The substrates in the P450-Glo assays are derivatives of beetle luciferin ((4*S*)-4,5-dihydro-2-(6-hydroxybenzothiazolyl)-4-thiazolecarboxylic acid) and are substrates for CYP but not luciferase. The derivatives are converted by CYP to a luciferin product that is detected in a second reaction with a luciferin detection reagent. The amount of light produced in the second reaction is directly proportional to the activity of the CYP (<http://www.promega.com>). Such assays lend themselves to true

**TABLE 22.4 Assay Conditions Used for IC<sub>50</sub> Generation in the Authors' Laboratory Utilizing Fluorescent or Radiometric Endpoints and a Single rCYP Isoform and LC/MS-MS Endpoint with a Cocktail of rCYPs and Cassette of Substrates<sup>d</sup>**

Isoform	Substrate	Final [Substrate] (μM)	[CYP] (pmol/mL)	Incubation Time (min)	Endpoint	Vehicle (1% v/v)	Inhibitor (IC <sub>50</sub> ; μM)
CYP1A2	Ethoxyresorufin	0.7	17	15	λ <sub>ex</sub> 530 nm; λ <sub>em</sub> 590 nm	DMSO	Fluvoxamine (0.17)
CYP2C9	7-Methoxy-4-(trifluoromethyl) coumarin	50	50	20	λ <sub>ex</sub> 410 nm; λ <sub>em</sub> 535 nm	DMSO	Sulfaphenazole (0.3)
CYP2C19	7-Methoxy-4-(trifluoromethyl) coumarin	50	57	20	λ <sub>ex</sub> 405 nm; λ <sub>em</sub> 535 nm	MeCN	Omeprazole (3)
CYP2D6	3-[2-( <i>N,N</i> -diethyl- <i>N</i> -methylammonium)ethyl]-7-methoxy-4-methylcoumarin	10	56	15	λ <sub>ex</sub> 530 nm; λ <sub>em</sub> 590 nm	DMSO	Quinidine (0.060)
CYP3A4	7-Benzoyloxy-4-(trifluoromethyl) coumarin	10	9	15	λ <sub>ex</sub> 405 nm; λ <sub>em</sub> 535 nm	MeOH /MeCN	Ketoconazole (0.007)
CYP2C9	Naproxen	109	70	15	Liquid scintillation counting	DMSO	Sufaphenazole (0.45)
CYP2C19	Diazepam	21	40	15	Liquid scintillation counting	DMSO	Omeprazole (3)
CYP2D6	Dextromethorphan	5	20	15	Liquid scintillation counting	DMSO	Quinidine (0.025)
CYP3A4	Erythromycin	19	25	10	Liquid scintillation counting	DMSO	Ketoconazole (0.035)
5 CYP cocktail	1A2-Phenacetin	26	15				α-Naphthoflavone(0.014)
	2C9-Diclofenac	2	5				Sulfaphenazole (0.26)
	2C19-S-mephenytoin	31	3	10	LC/MS-MS	DMSO	Tranylcypromine (3.60)
	2D6-Bufurololol	9	5				Quinidine (0.018)
	3A4-Midazolam	3	5				Ketoconazole (0.0036)
			<i>Total 33</i>				
3 CYP cocktail	2B6-Bupropion	20	18				Ticlopidine (0.55)
	2C8-A modiaquine	2	1	10	LC/MS-MS	DMSO	Quercetin (5.12)
	3A5-Midazolam	2	5				Ketoconazole (0.17)
			<i>Total 24</i>				

<sup>d</sup>See Refs. 116, 118, and 119. Data is mean *n*> 50.

**TABLE 22.5** Some Considerations in the Selection of CYP Assay Endpoints

Consideration	Assay Type			
	Radiolabeled	Fluorescent	LC/MS-MS	Luminescent
Cost	++	+++	+	++
Throughput	+	+++	++	HTS
Accessibility	++	++	+	++
Interference	+++	+	++	+
Health and safety	+	+++	+++	+++
Published information	++	+++	+++	+
Marketed drug substrates	++	—	+++	—

HTS capabilities, providing quick readouts on multiple well plates. Table 22.5 shows a comparison of LC/MS-MS, fluorometric, radiometric, and luminescent endpoints to aid consideration of assay choice.

Using fluorescent or luminescent assays for enhanced/HTS and rapid generation of  $IC_{50}$  data may be invaluable for the development of structure–activity relationships (SARs) for the different CYPs [69, 128]. The utility of high throughput CYP inhibition screens has allowed generation of large datasets, facilitating development of statistical *in silico* approaches. To this end, quantitative *in silico* models now exist for prominent CYPs, often based on physical-chemical properties alone. These models can classify compounds as “noninhibitors” with reasonable accuracy (~80% of compounds were classified correctly for CYP3A4 and CYP2D6 [128]) and can guide medicinal chemistry away from CYP interactions at early stages of drug discovery programs [128, 129]. In theory, these *in silico* models allow predictions of inhibitory potency to be made even before a compound has been synthesized [129].

## 22.4 GENERATION OF *IN VITRO* CYTOCHROME P450 INHIBITION DATA

Fundamentally, *in vitro* kinetic data must be accurate, meaningful, and relevant. As discussed previously, all assays should be preoptimized for time and protein linearity to ensure the reaction proceeds under linear conditions. The objective is to determine  $K_i$  for predicting the likelihood and extent of DDIs. Experimentally, this is achieved by either defining the  $IC_{50}$  or by directly determining a  $K_i$ . From Eqs. 22.4 and 22.7, it is clear that if  $[S] = K_m$ , the  $K_i$  can be estimated to within twofold from an accurate  $IC_{50}$  determination. This may well be less than the error in determining  $K_i$  by an extensive (and laborious) fully inhibited Michaelis–Menten study. Furthermore, the mechanism of inhibition must be defined and, if irreversible, generation of  $K_I$  and  $k_{inact}$  is required for accurate assessment of DDIs.

When generating *in vitro* inhibition data, particularly when evaluating the propensity of NCEs to behave as CYP inhibitors, it is important to include the appropriate positive controls. To this end, known inhibitors should be included in any incubations as markers, and it has been suggested that  $K_i$  or  $IC_{50}$  values generated

for these standard inhibitors should fall within threefold of the median literature value for an assay to be deemed acceptable [130].

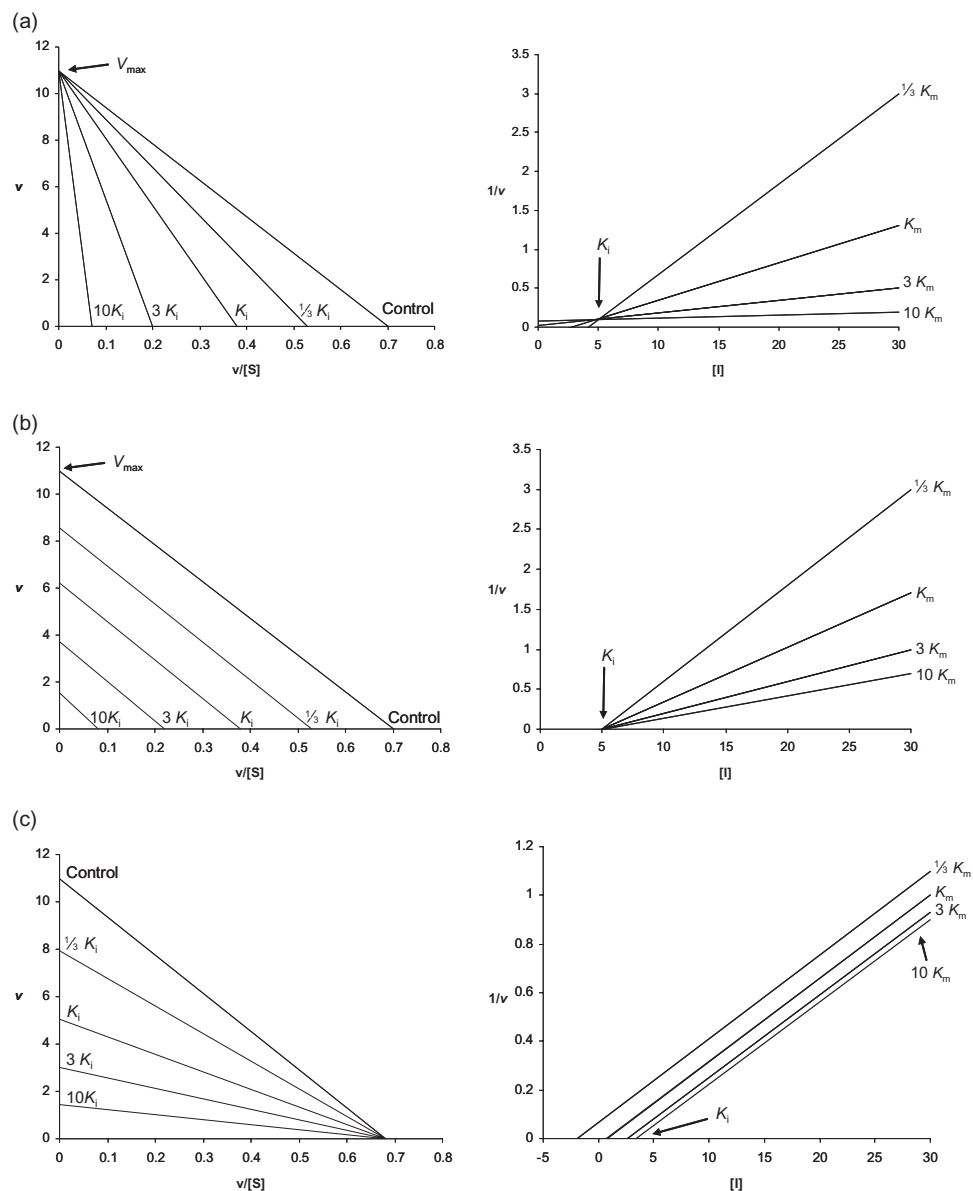
#### 22.4.1 Reversible Inhibition ( $IC_{50}$ and $K_i$ )

An  $IC_{50}$  is generated using a single concentration of substrate. Inhibitor concentrations should cover at least two log units (e.g., 0, 0.1, 0.3, 1, 3, 10  $\mu$ M) to ensure a range of measurements from negligible to virtually complete inhibition. The percentage of total enzyme activity remaining (from control incubations) is plotted against log inhibitor concentration [I] to generate an  $IC_{50}$  curve. Plotting log [I] results in more accurate confidence intervals when using nonlinear regression, as the data is equally spaced on a log axis, allowing symmetrical confidence intervals to be calculated. In the authors' laboratory, an automated assay to determine  $IC_{50}$  is adopted;  $K_i$  is estimated to be half the  $IC_{50}$  (as  $[S] = K_m$ ). The automated assay utilizes a pool of recombinant CYPs (CYP1A2, CYP2C9, CYP2C19, CYP2D6, and CYP3A4) to minimize nonspecific binding in the incubation, at protein concentrations and incubation times designed to ensure the selective metabolism of a substrate cassette at concentrations equivalent to the  $K_m$  (phenacetin, diclofenac, *S*-mephenytoin, bufuralolol, and midazolam, respectively). LC/MS-MS analysis of specific metabolite formation allows comparison of peak area in the presence of increasing concentration of test inhibitor against a DMSO control. Recently, an additional assay was developed for CYP2B6, CYP2C8, and CYP3A5 due to the emerging importance of these enzymes [118]. This second automated assay again uses a pool of rCYPs and a substrate cassette of bupropion, amodiaquine, and midazolam, respectively. Table 22.4 shows the reaction conditions for these assays.

An alternative approach is to experimentally define the inhibition constant  $K_i$ . To do this accurately, the inhibitor concentration should span a range around the suspected  $K_i$  (based on a prior  $IC_{50}$  estimation) using at least five concentrations, for example,  $0K_i$ ,  $0.3K_i$ ,  $K_i$ ,  $3K_i$ , and  $10K_i$ . Typically, 8–10 substrate concentrations should be used to investigate each concentration of inhibitor and these should span from  $K_m/3$  to  $3K_m$ . The range of substrate concentration needs to be expanded for increased inhibitor concentrations, since the  $K_m$  shifts up when the substrate's metabolism is competitively inhibited.  $K_i$  determinations therefore require large numbers of incubations, as all parts of the velocity–substrate concentration profile need to be defined for each inhibitor concentration. Accurate determinations of any one of the parameters ( $K_m$ ,  $V_{max}$ ,  $K_i$ ) depends on accurate determination of all three.

Once generated, the data can be visually inspected for the mechanism of inhibition; this is commonly achieved using two types of graphical representation—the Eadie–Hofstee plot ( $v$  against  $v/[S]$ ) and the Dixon plot ( $1/v$  against [I]). The type of inhibition can be determined from the shape of the linearized data and examples of these plots for each type of reversible inhibition are shown in Fig. 22.7. Kinetic parameters should never be determined from linear transforms such as these as doing so generally leads to inaccurate estimation of  $K_i$  (the two-dimensional nature of the data fit does not fully describe the effect of changing inhibitor concentrations on different concentrations of substrate). The plots are solely for the purpose of aiding interpretation of the inhibition type. Consequently, nonlinear regression software is an absolute requirement to facilitate iterative modeling of changing sub-





**FIGURE 22.7** Graphical representation of inhibition plots: Eadie–Hofstee plots ( $v$  against  $v/[S]$  on the left) and Dixon plots ( $1/v$  against  $[I]$  on the right) for (a) competitive inhibition, (b) noncompetitive inhibition, (c) and uncompetitive inhibition.

strate and inhibitor concentrations, fitting the untransformed data to the appropriate rate equation (for competitive, noncompetitive, or uncompetitive inhibition, Eqs. 22.3, 22.5, and 22.6, respectively). In practice, this means nonlinear regression modeling of all the substrate concentration, velocity, and inhibitor concentration data simultaneously using the equation for competitive inhibition (Eq. 22.3). The data

should be weighted by  $1/y$ , to compensate for unequal experimental variance in the different experimental points. (Velocity–substrate concentration plots tend to have a constant relative error associated such that the residual—the difference between predicted and observed velocity—will be larger at higher reaction velocities. This is true because the error is the same at low and high velocities, giving rise to larger standard deviations from the mean observed velocity at higher velocities.) The same dataset should then be reanalyzed using the equation for noncompetitive inhibition (Eq. 22.5). The two mathematical models can then be discriminated by evaluation of “goodness-of-fit” criteria. For a full explanation, see Mannervik [131]. Generally, goodness-of-fit criteria (e.g., Akaike values) and accuracy of parameter estimates (standard deviation and coefficient of variance) should be compared for both models and the most appropriate model (type of inhibition) chosen.

#### 22.4.2 Mechanism-Based Inhibition ( $K_I$ , $k_{inacts}$ and $IC_{50}$ )

True irreversible inhibition or mechanism-based inhibition (MBI) of CYPs can only be defined according to the criteria detailed previously and may be established by mechanistic studies such as dialysis of the inhibited CYP or spectral shifts in absorbance [100]. MBI is not readily detected in standard *in vitro* assay formats. Time-dependent inhibition (TDI) is a collective term for a change in potency of CYP inhibitors during an *in vitro* incubation or dosing period *in vivo*. Such changes usually result in an increase in inhibitory potency; potential mechanisms include the formation of more inhibitory metabolites and MBI. In the following assays, it is the time-, concentration-, and/or NADPH-dependent nature of the inhibition under investigation.

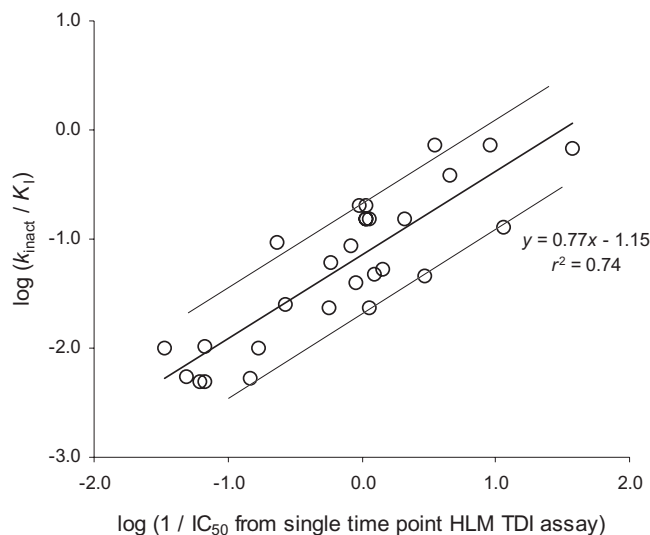
In essence, an assay to detect TDI involves a preincubation step, where enzyme is incubated with and without NADPH and in the presence of varying inhibitor concentrations. This reaction is allowed to proceed over a predetermined time course, before dilution of the reaction mixture into a second incubation that includes a probe substrate to assess the degree of enzyme inactivation [77]. To accurately define the inhibitory constants of TDI, the conditions governing the two separate incubations must be carefully chosen. A useful critical evaluation of experimental design, methodology, and data analysis for TDI protocols in published literature highlighted not only the variation between laboratories but the subsequent impact on the predicted effects of TDI *in vivo* if the *in vitro* kinetic parameters are incorrectly defined [79].

The experimental design of a TDI assay must adhere to the same conditions as those for reversible inhibition (linear enzyme reactions, etc.) and protocols have to be adjusted to correctly define the kinetics of different inhibitors. These adjustments are dependent on the potency of the inhibitor under investigation, as potent inhibitors require considerably shorter preincubation times than weaker inhibitors [75, 79]. To this end, the time course of the pre-incubation must be such that the inhibition at each concentration of inhibitor is linear with respect to time—at least over the initial preincubation period. Allowing the preincubation reaction to proceed over a large time course with multiple dilution points into the second assay (0, 0.5, 1, 2, 3, 5, 10, 20, and 30 min) should gauge this; however, this is not necessarily practical due to the large reaction volume required in the preincubation to sustain multiple dilution points. In practice, two separate assays may have to be conducted: an

initial assessment where preincubation occurs over an intermediate time course (such as 5, 10, and 20 min), and a second assay where times are adjusted for more accurate definition of the inhibition profile (based on inspection of the data generated in the initial assessment). In addition to this, it is vital to ensure all the CYP inactivation occurs in this preincubation, to allow accurate measurement of inhibited enzyme with respect to incubation time, without concurrent inhibition and probe substrate metabolism in the second incubation. To this end, the dilution of the preincubation reaction mixture into the second incubation should be as large as possible (in general, 10–50-fold) while still guaranteeing analytical sensitivity for the probe substrate metabolite. Additionally, the concentration of CYP selective substrate in the final incubation should be high relative to its  $K_m$  (in general, 5–10-fold), again to minimize further CYP inactivation after the preincubation step, and the final incubation time should be minimized. Finally, since the enzyme can undergo some inactivation in the absence of inhibitor, the enzyme activity remaining following each preincubation time point should be determined by comparison to control (enzyme incubated without inhibitor). In the authors' laboratory, two automated assays with an LC/MS-MS endpoint are used to assess TDI potential [98] and have been modified to allow detection of weaker time-dependent inhibitors [75]. An initial screen generates an  $IC_{50}$  value using HLMs and a cassette of CYP specific probe substrates. A single inhibitor concentration is tested using a single preincubation time of 30 min and a 20-fold dilution. If inhibition is detected in this  $IC_{50}$  assay, a second assay is then used to generate values for  $K_I$  and  $k_{inact}$  from rCYPs. This assay adheres to the strict criteria detailed earlier in order to accurately define the kinetic parameters. Five preincubation times (up to 23 min) are used with six inhibitor concentrations and a dilution factor of 20-fold. There is a good correlation between  $k_{inact}/K_I$  ratio (a measure of inhibitory efficiency) generated in rCYP and the  $IC_{50}$  from the HLM assay, thus allowing NCE evaluation from a reasonable throughput assay [98] (see Fig. 22.8).

The analysis of data generated in TDI assays must be carefully conducted. First, the natural log of the percent control activity remaining is plotted against time. The slope of the linear portion for each inhibitor concentration is equivalent to the inactivation rate constant,  $k$  (or  $k_{obs}$ ). Graphical representations of these plots are shown in Fig. 22.9. Using nonlinear regression, this data can be fitted to Eq. 22.13 to estimate  $k_{inact}$  and  $K_I$ . A common fault is to use linear Kitz–Wilson plots rather than nonlinear regression, as two-dimensional fitting of the data cannot yield accurate estimations for  $k_{inact}$  and  $K_I$ . [79]. It is also worth noting that if all the CYP inactivation occurs in the preincubation step, the point where the log-linear fits of each inhibitor concentration cross the y axis (time = 0) represents the reversible inhibition component. However, if the dilution of the preincubation is not sufficient, considerable reversible inhibition may occur during the second incubation, making data interpretation more problematic [98].

Finally, an additional utility of a TDI screen in drug discovery may be as a “flag” for reactive metabolite formation. Electrophilic species are often generated during CYP MBI and there is a high chance that these may escape from the CYP active site and react elsewhere in the body, either via detoxification processes (glutathione deactivation) or more worryingly via reaction with proteins or nucleic acids. It is not at all surprising that the same chemical groups implicated in reactive metabolite formation are also known to inhibit CYPs irreversibly [75]. Perhaps linked to this,



**FIGURE 22.8** Plot of log unbound  $IC_{50}$  against  $\log(k_{inact}/K_i)$  for 28 time-dependent inhibitors of CYP3A4.  $IC_{50}$  values were generated using the automated single time point, single inhibitor concentration human liver microsomal time-dependent inhibition screen [98]. These values were adjusted for incubational binding to give unbound  $IC_{50}$  values (according to the mathematical model of Austin et al. [111]).  $K_i$  and  $k_{inact}$  were estimated using the 3 time point, 6 inhibitor concentration, recombinant CYP automated time-dependent inhibition assay. The solid line is the line of best fit, with dashed lines representing threefold deviation from the line. (From Ref. 98.)

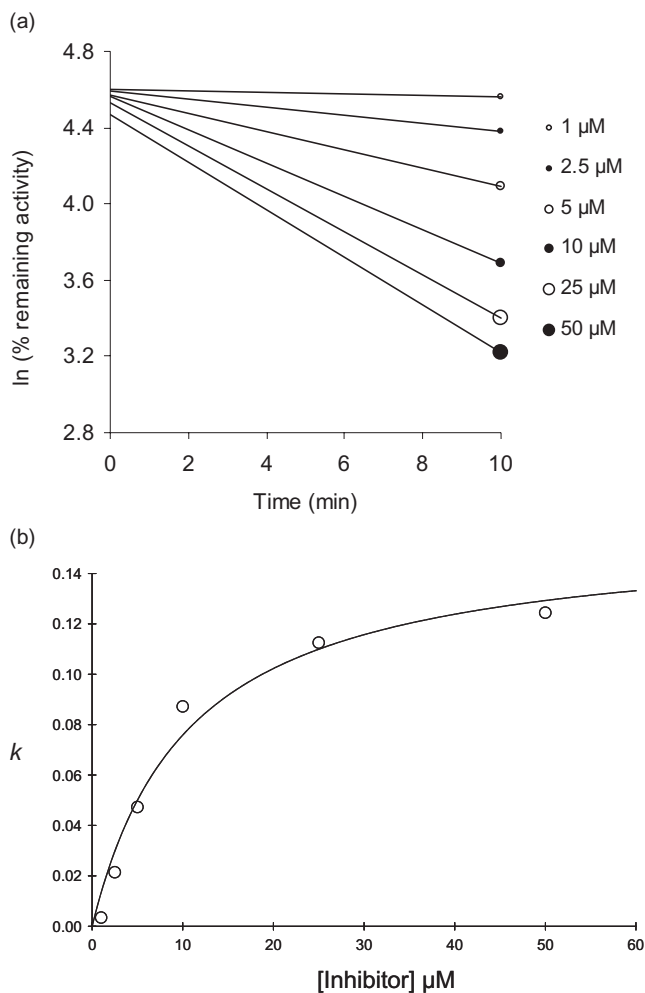
it is found that many drugs exhibiting TDI are also associated with adverse drug reactions (e.g., erythromycin, diclofenac, tacrine, valproic acid, halothane, suprofen, ticlopidine, tienilic acid, isoniazid, zileuton) [132, 133]. Using TDI screening, alongside cold trapping and/or radiolabeled binding studies, may be advantageous in minimizing the potential for reactive metabolite formation as early as possible in a drug discovery screening cascade.

## 22.5 IN VITRO–IN VIVO EXTRAPOLATION OF CYTOCHROME P450 BASED DRUG–DRUG INTERACTIONS

The clinical implications of DDIs and the importance of generating accurate *in vitro* data have been discussed; from this it follows that IVIVE requires careful data handling using the approach outlined below to generate predictions of clinical impact. Over the last 10–15 years, predicting the extent of DDI in a semiquantitative way has been the subject of many publications and the most effective models are presented here.

### 22.5.1 Approach for Reversible Inhibitors

If the metabolism of a drug is reversibly inhibited, the magnitude of the effect will be  $1 + [I]/K_i$  (sometimes termed the inhibition index,  $I_i$ ), provided that the concen-



**FIGURE 22.9** Graphical representation of irreversible (time-dependent) inhibition. (a) Inhibition of control activity against time for increasing inhibitor concentration; and (b) nonlinear regression of  $k$  against inhibitor concentration giving  $k_{\text{inact}}$  of  $0.16 \text{ min}^{-1}$  and  $K_i$  of  $10.6 \mu\text{M}$  (circles are observed data points and line is predicted data fit).

tration of substrate undergoing inhibition is well below its  $K_m$  value and that it is eliminated by the single pathway being inhibited [134]. This is because the magnitude of effect is dependent not only on how much affinity the inhibitor has for the enzyme ( $K_i$ ) but also on the concentration of inhibitor available. For competitive inhibition, the  $K_m$  of the victim drug will be increased to  $K_m + K_m [I]/K_i$  (or  $K_m(1 + [I]/K_i)$ ) and therefore the intrinsic clearance ( $CL_{\text{int}}$ ) (as  $CL_{\text{int}} = V_{\text{max}}/K_m$ ) will be decreased to  $V_{\text{max}}/(K_m(1 + [I]/K_i))$  or  $CL_{\text{int}}/(1 + [I]/K_i)$ .

For an IV administered victim drug, the following is true:

$$CL = \text{Dose}/AUC \quad (22.14)$$

$$CL/CL' = AUC'/AUC \quad (22.15)$$

where  $CL$  is the *in vivo* clearance in the absence and  $CL'$  in the presence of inhibitor, respectively, and  $AUC$  is area under the plasma–concentration time curve in the absence and  $AUC'$  in the presence of inhibitor, respectively.

$$\frac{AUC'}{AUC} = \frac{Q_h \cdot CL_{int} \cdot f_{ub}}{Q_h + CL_{int} \cdot f_{ub}} \cdot \frac{Q_h + CL'_{int} \cdot f_{ub} \cdot 1/I_i}{Q_h \cdot CL'_{int} \cdot f_{ub} \cdot I_i} \quad (22.16)$$

$$= \frac{Q_h^2 \cdot CL_{int} \cdot f_{ub} + Q_h^2 \cdot CL_{int} \cdot f_{ub} \cdot 1/I_i \cdot CL_{int} \cdot f_{ub}}{Q_h^2 \cdot f_{ub} \cdot CL_{int} \cdot 1/I_i + Q_h^2 \cdot f_{ub} \cdot CL_{int} \cdot f_{ub} \cdot CL_{int} \cdot 1/I_i} \quad (22.17)$$

$$= \frac{Q_h + (f_{ub} \cdot CL_{int})/I_i}{Q_h/I_i + (f_{ub} \cdot CL_{int})/I_i} \quad (22.18)$$

$$\frac{AUC'}{AUC} = \frac{Q_h \cdot I_i + f_{ub} \cdot CL_{int}}{Q_h + f_{ub} \cdot CL_{int}} \quad (22.19)$$

where  $Q_h$  is hepatic blood flow and  $I_i$  is the inhibition index  $1 + [I]/K_i$ . If  $f_{ub} \cdot CL_{int} \ll Q_h$ , there is a high inhibitory effect and the  $AUC$  ratio approximates to  $I_i (1 + [I]/K_i)$ . If  $f_{ub} \cdot CL_{int} \gg Q_h$ , there is little effect since the  $AUC$  ratio approximates to 1.

For an orally administered victim drug, the fraction escaping hepatic first-pass extraction in the presence ( $F'$ ) or absence ( $F$ ) of inhibitor is

$$\frac{F'}{F} = \frac{1 - CL'_h/Q_h}{1 - CL_h/Q_h} \quad (22.20)$$

Rearranging the well stirred model [135],

$$CL_{int} = \frac{CL_h}{f_{ub}(1 - CL_h/Q_h)} \quad (22.21)$$

$$= \frac{CL_h}{F \cdot f_{ub}} \quad (22.22)$$

It follows that

$$F = \frac{CL_h}{CL_{int} \cdot f_{ub}} \quad (22.23)$$

Therefore,

$$F = \frac{CL_{int} \cdot f_{ub} \cdot Q_h}{CL_{int} \cdot f_{ub} + Q_h} \cdot \frac{1}{CL_{int} \cdot f_{ub}} \quad (22.24)$$

$$= \frac{Q_h}{CL_{int} \cdot f_{ub} + Q_h} \quad (22.25)$$

Putting this back into Eq. 22.20, we find

$$\frac{F'}{F} = \frac{Q_h}{CL_{int} \cdot f_{ub}/I_i + Q_h} \cdot \frac{CL_{int} \cdot f_{ub} + Q_h}{Q_h} \quad (22.26)$$

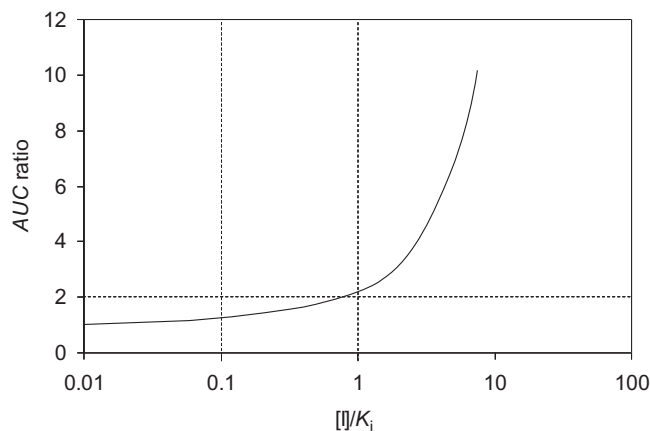
$$\frac{F'}{F} = \frac{CL_{int} \cdot f_{ub} + Q_h}{(CL_{int} \cdot f_{ub}/I_i) + Q_h} \quad (22.27)$$

So, when  $f_{ub} \cdot CL_{int} \ll Q_h$ , there is little inhibitor effect since the ratio of the fraction escaping first pass is approximately 1. When  $f_{ub} \cdot CL_{int} \gg Q_h$ , the ratio of the fraction escaping first pass in the presence and absence of inhibitor approximates to  $1 + [I]/K_i$ .

It is recognized as a “rule of thumb” that inhibitors with  $[I]/K_i < 0.1$  represent a low DDI risk, those with  $[I]/K_i$  between 0.1 and 1 are moderate risk, and those with  $[I]/K_i > 1$  are high risk. The relationship between the *AUC* ratio and  $[I]/K_i$  ratio is depicted in Fig. 22.10 and is the most simple model of CYP inhibition [80, 130, 136, 137]. A more detailed approach, that is, the fundamental basis of most IVIVE for CYP inhibition, expands on the inhibition index ( $1 + [I]/K_i$ ) by including the relative metabolic contribution (*fm*) of each inhibited CYP to the *in vivo* metabolism of the substrate under investigation:

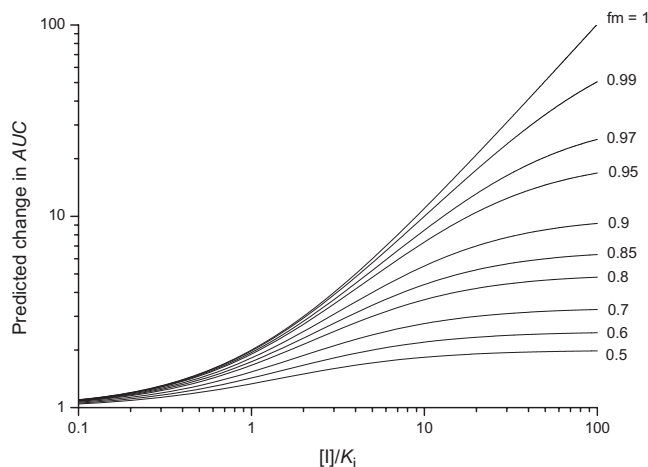
$$\frac{AUC'}{AUC} = \frac{1}{\frac{fm_{CYP}}{1 + [I]/K_i} + (1 - fm_{CYP})} \quad (22.28)$$

The impact of *fm* on the magnitude of inhibition can be visualized in Fig. 22.11 [56, 68, 138]. From this plot it is clear that  $fm > 0.5$  is required for inhibition to generate significant increases in *AUC* and that predictions are very sensitive to the *fm* used when it approaches 1. Within a population, *fm* will vary and individuals most at risk



**FIGURE 22.10** Qualitative zoning of the prediction of the magnitude and risk of drug–drug interactions mediated by reversible inhibition (*AUC* ratio) using  $[I]/K_i$  ratio. Low risk,  $<0.1$ ; medium risk,  $0.1$ – $1$ ; high risk,  $>1$ .





**FIGURE 22.11** Simulation of predicted change in AUC as a function of  $[I]/K_i$  with different  $f_m$  values. The simulation is based on Eq. 22.28, where  $f_m$  is the fraction of substrate clearance mediated by the inhibited metabolic pathway. (Adapted from Ref. 98.)

of DDI will be those with the highest  $f_m$ . The model can be expanded further to include inhibition of multiple CYP isoforms and the effect of each contribution to overall clearance of the victim drug and other routes of elimination need to be considered [136, 138].

The choice of which  $[I]$  to use in these predictions has been a source of much debate in the literature [7, 13, 80, 88, 139, 140]. Classical pharmacology states that only the unbound fraction of a drug is able to cross membranes and interact with a target protein; from this it follows that a drug associated with proteins in the plasma or tissue is unavailable for interaction. It follows that the use of unbound drug concentrations and inhibition constants should be the starting point in IVIVE and doing so can give good predictions when other relevant factors are included [88]. Changes in the unbound fraction of a drug in the blood ( $f_{ub}$ ) is in itself a potential mechanism of DDI;  $f_{ub}$  is increased when one drug is displaced by other drugs at the site of plasma protein binding. These interactions rarely cause serious clinical problems [7, 141]. However, despite the rationale for using corrections of unbound drug concentrations to predict clearance [108, 142], the use of drug binding terms in predictive IVIVE models remains controversial [13, 88, 97, 139].

Across the spectrum of literature on the subject, inconsistencies in the use of drug binding terms and the choice of *in vivo* inhibitor concentration may have clouded the general understanding and inadvertently reduced confidence in IVIVE for anything other than *qualitative* DDI predictions (average steady plasma concentration,  $[I]_{av}$ , peak steady-state plasma concentration,  $[I]_{max}$ , maximum hepatic input concentration  $[I]_{in}$ ; see Table 22.6) [7, 13, 80, 144]. The suggestion that total blood concentrations can be used with unbound *in vitro*  $K_i$  values to achieve good predictivity is less than helpful, since it leaves the reader potentially unsure as to the impact of blood binding *in vivo*. Binding to blood components may not reach equilibrium during the short time period in the portal vein, leading to a greater free-drug concentration than may be predicted and underestimation of DDI [136]. However,

**TABLE 22.6 Choice of *In Vivo* Inhibitor Concentration Advocated in the Literature in Recent Years**

Which [I]	Equation	$K_i$ Adjusted for $f_{\text{unc}}$ ?	Author Comments	References
Maximum <i>unbound</i> hepatic input concentration, $[I]_{\text{in}}$ , where $[I]$ is $[I]:f_{\text{ub}}$	$[I]_{\text{av}} = \frac{D/\tau}{CL/F}$ $[I]_{\text{in}} = [I]_{\text{av}} + \frac{k_a \cdot F_a \cdot D}{Q_h}$	No	The incidence of false-negative predictions was largest using $[I]_{\text{av}}$ for unbound drug concentration and smallest using $[I]_{\text{in}}$ for total drug concentration for each of the CYP enzymes. The use of $[I]_{\text{in}}$ is recommended because true negatives can be identified and, in contrast to the use of other $[I]$ values, false negatives eliminated.	137
Maximum <i>total</i> hepatic input concentration, $[I]_{\text{in}}$	$[I]_{\text{in}} = [I]_{\text{av}} + \frac{k_a \cdot F_a \cdot D}{Q_h}$	No	True negatives can be identified and, in contrast to the use of other values for $[I]$ , false negatives are eliminated. True positives are also predicted well and, although the incidence of false-positive predictions is quite high, the use of hepatic input concentration is recommended.	80
Maximum <i>total</i> hepatic input concentration $[I]_{\text{in}}$	$[I]_{\text{in}} = [I]_{\text{av}} + \frac{k_a \cdot F_a \cdot D}{Q_h}$	No	The use of $[I]_{\text{in}}$ , incorporating both $f_{m_{\text{CYP}}}$ and refined $k_a$ values resulted in the most successful prediction overall.	143
Average <i>total</i> steady plasma concentration, $[I]_{\text{av}}$ , or maximum <i>total</i> hepatic input concentration, $[I]_{\text{in}}$	$[I]_{\text{in}} = [I]_{\text{av}} + \frac{k_a \cdot F_a \cdot D}{Q_h}$ $[I]_{\text{av}} = \frac{D/\tau}{CL/F}$	Yes	Either the average systemic plasma concentration after repeated oral administration ( $[I]_{\text{av}}$ ) or the maximum hepatic input concentration ( $[I]_{\text{in}}$ )	56
Average <i>total</i> steady plasma concentration, $[I]_{\text{av}}$	$[I]_{\text{av}} = \frac{D/\tau}{CL/F}$	Yes	The use of $[I]_{\text{av}}$ as the $[I]$ surrogate generated the most successful predictions as judged by several criteria. Using $[I]_{\text{av}}$ incorporation of either plasma protein binding of inhibitor or gut wall CYP3A4 inhibition did not result in a general improvement of DDI predictions.	13
Maximum <i>unbound</i> hepatic input concentration, $[I]_{\text{in,us}}$ where $[I]$ is $[I]:f_{\text{ub}}$	$[I]_{\text{in}} = [I]_{\text{av}} + \frac{k_a \cdot F_a \cdot D}{Q_h}$	Yes ( $[\text{protein}]$ low so $f_{\text{unc}}$ assumed negligible)	The use of estimated unbound hepatic inlet $C_{\text{max}}$ during the absorptive phase yielded the most accurate predictions of the magnitudes of DDI.	139

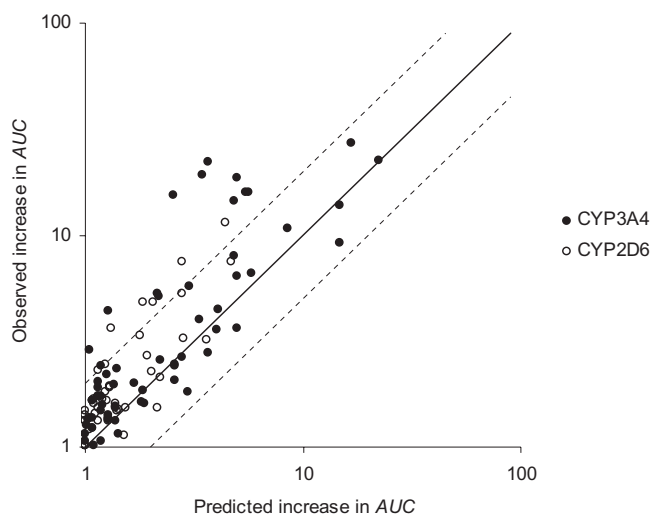
the “free-drug” theory should be regarded as the default situation and if successful predictions cannot be made, the reason(s) should be sought.

Equation 22.29 shows the maximum hepatic input of inhibitor (the sum of the contribution from the systemic circulation and the absorption phase;  $[I]_{in}$ ) calculated using the absorption rate constant ( $k_a$ ), the fraction absorbed ( $F_a$ ), and dose [7].

$$[I]_{in} = [I]_{av} + (k_a \cdot F_a \cdot \text{Dose})/Q_h \quad (22.29)$$

Often it is assumed that the rate of absorption is instantaneous (approaching  $0.1 \text{ min}^{-1}$ ); however, changes in  $k_a$  can account for marked differences in  $[I]_{in}$  and the value chosen should be carefully considered [143].

In the authors' laboratory, good predictions have been made using unbound maximum hepatic concentration  $[I]_{in,u}$  along with correction of  $K_i$  for binding in the incubation ( $f_{uinc}$ ) [88]. Using a published dataset of clinical DDIs [80], an 84% success rate was achieved for prediction of DDI at CYP3A4 and a 91% success rate predicting CYP2D6 interactions (Fig. 22.12) [88]. There were some noticeable outliers in the dataset, however; these highlighted that subtle changes in some parameters (e.g., plasma protein values for a highly bound drug) can result in large discrepancies in calculated unbound concentration and that measured  $AUC$  changes for drugs with low oral bioavailability are subject to large errors. Additionally, it was apparent that the impact of the  $[I]/K_i$  estimate on  $AUC$  change becomes small once  $f_{m_{CYP}}$  falls below 0.85 (Fig. 22.11) [88]. The clinical dataset is substantially smaller for CYP2C9 than for CYP3A4 and CYP2D6, but it was found that using  $[I]_{in,u}$  and  $K_{i,u}$  to predict changes in  $AUC$  resulted in 14 correct assignments and 1 false negative [68]. However, the desire to avoid false negatives in a drug discovery screening



**FIGURE 22.12** Observed and predicted  $AUC$  values for CYP3A4 and CYP2D6 DDIs. The square box represents twofold increase in  $AUC$ , the solid line is unity and the dashed lines are twofold errors in either direction. (Adapted from Ref. 88.)

cascade perhaps explains the popularity for using *total* concentration terms for [I], despite the decreased accuracy in absolute prediction afforded by doing so.

The multifaceted nature of CYP inhibition *in vivo* requires a great deal of understanding for accurate prediction, and the reasons behind the lack of a unifying model can be appreciated. As the IVIVE field progresses, it is anticipated that prediction accuracy will increase as there is profound interest and utility in such models within the pharmaceutical industry and considerable academic effort being made toward this goal. The ultimate aim is an integrated approach toward CYP inhibition prediction, which encompasses the role of all CYP isoforms and is underpinned by sound pharmacokinetic principles.

### 22.5.2 Approach for Mechanism-Based Inhibitors

Two models are currently presented in the literature and they rely on some key parameters for accurate predictions of DDI [7, 100]. As for reversible inhibition, the relevant concentration of inhibitor at the target CYP and the fraction of substrate clearance affected by inhibiting the enzyme are fundamental. In addition to this, predictions of MBI also employ the *in vivo* rate of enzyme synthesis ( $k_{\text{synth}}$ ) and of degradation ( $k_{\text{degrad}}$ ), and the *in vitro* inactivation kinetic parameters ( $K_I$  and  $k_{\text{inact}}$ ). Where the currently available models differ is in the complexity of predicting the effect of changing inhibitor concentration over a dosing interval and the impact of this on the overall amount of inactivated CYP.

A model first presented by Hall and co-workers and subsequently expanded by the same group [100, 145, 146] is used most widely. This model assumes that the rates of enzyme synthesis and degradation are unaffected by enzyme inactivation and rely on the steady-state concentration of active enzyme being proportional to the ratio of the rate of synthesis and the rate of degradation. In the presence of inhibitor, the rate of degradation is then the sum of the *in vivo* degradation rate constant ( $k_{\text{degrad}}$ ) and the rate constant describing the drug-induced inactivation ( $\lambda$ ), where

$$\lambda = ([I] \cdot k_{\text{inact}} / [I] + K_I) \quad (22.30)$$

The model including inactivation of intestinal CYP and the fraction of drug metabolized by the inhibited pathway in the absence of the inhibitor is given in by

$$\frac{AUC}{AUC'} = \frac{F'}{F} \cdot \left( 1 + \frac{fm}{([I] \cdot k_{\text{inact}}) / (k_{\text{degrad}} \cdot (K_I + 1))} + (1 - fm) \right)^{-1} \quad (22.31)$$

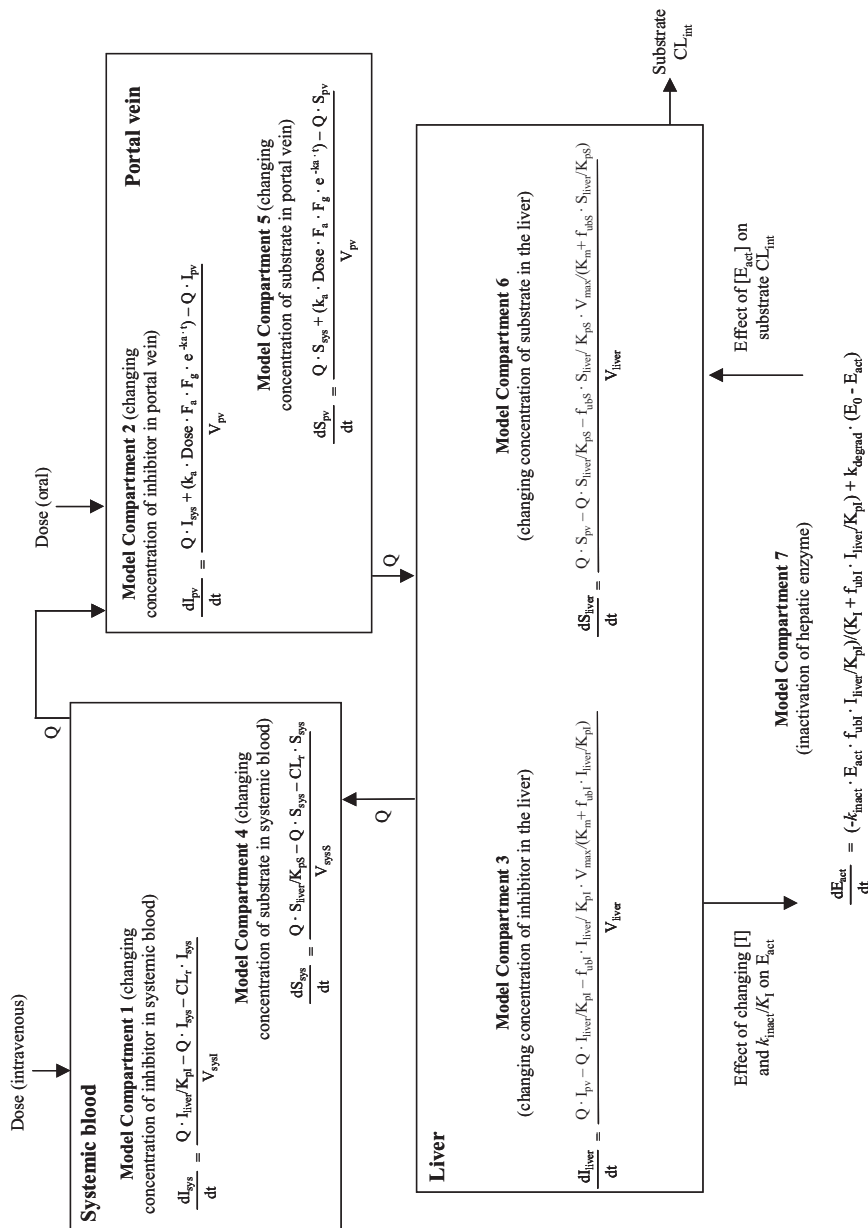
This model has been shown to predict well for verapamil [145] and other mechanism-based inhibitors [147]. The accuracy of prediction is dependent on the  $k_{\text{degrad}}$  used, especially for substrates undergoing substantial ( $fm > 0.9$ ) metabolism by the inhibited enzyme [56].

As with reversible CYP inhibition, a nonlinear regression model encompassing the various parameters predicting changing inhibitor and substrate concentrations affords the most sophisticated approach. To that end, an elegant model was proposed by Ito et al. [7], which predicts the effect of macrolide antibiotics on CYP3A4 [104]. A real advantage of this model is that it allows visualization of the plasma concentration–time profiles for the affected substrate and the inhibitor, and more importantly the change in active enzyme concentration over time. This model encompasses not only  $k_{\text{inact}}$  and  $K_I$  and the rate of resynthesis of active enzyme but also the changing concentration of the inhibitor as a function of its own disposition. Three compartments are represented: portal vein, liver, and systemic blood. The rate of change of inhibitor over time is modeled in each compartment with and without coadministered drug and a final compartment describes inactivation of hepatic CYP (Fig. 22.13). The use of this model requires nonlinear regression in a seven-compartment model which, while possible, is not necessarily practicable. Consequently, in the authors' laboratory, an analogous model has been written and implemented using a Microsoft Excel spreadsheet format [75], allowing simple manipulation and interrogation of this complex model.

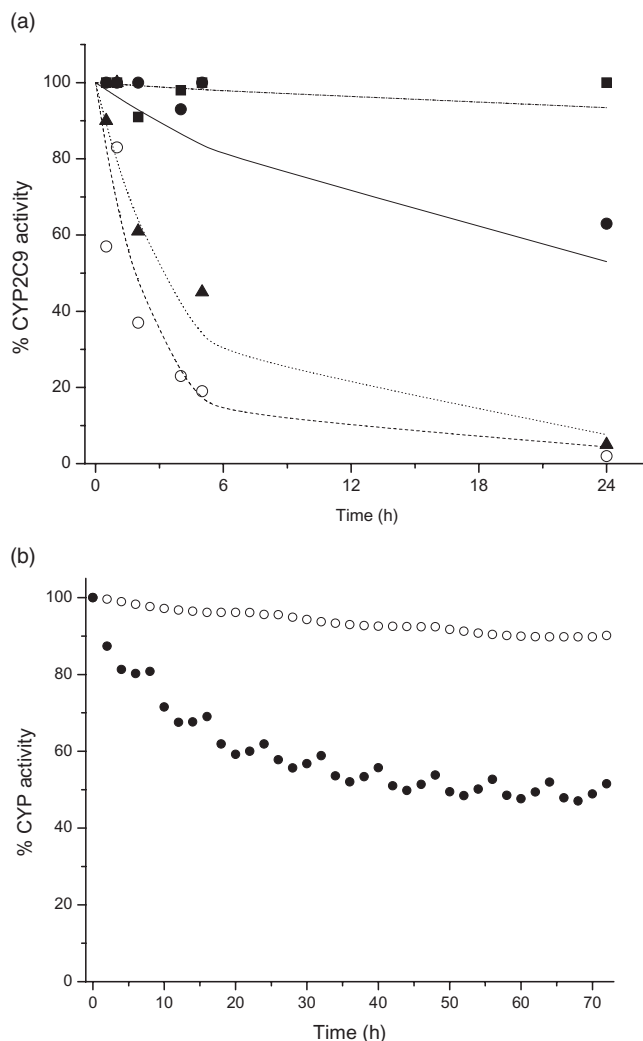
Alongside the validation of this model presented by Ito et al. [104], where the interaction between erythromycin and clarithromycin with midazolam in humans was accurately predicted, the isolated perfused rat liver model (IPRL) has been used to demonstrate the effect of known mechanism-based inhibitors and NCEs [148]. Using the IPRL allowed ready validation of the predicted impact of TDI *in vivo*. It provides a bridge between the *in vitro* effect and *in vivo* impact for compounds with existing clinical data and NCEs, using pre-clinical species to bring confidence to predicted effect in humans. An adapted five-compartment model was used to fit the data and the *ex vivo* effects, although modest, were well predicted. The interplay between inhibitory potency and pharmacokinetic parameters was demonstrated; a less potent inhibitor with longer PK duration can impact hepatic CYPs to the same extent as a much more potent inhibitor with short duration. This was confirmed for troleandomycin and erythromycin in the IPRL; troleandomycin was found to be over 100-fold more efficient at inhibiting CYP3A2 than erythromycin *in vitro*, yet in the IPRL only a two fold increase over the effect of erythromycin was observed, reflecting the clinical situation for these drugs [149, 150]. This point was further confirmed by a study in cultured human hepatocytes performed in the authors' laboratory [95], where a compound that is a relatively weak time-dependent inhibitor of CYP2C9 was observed to have a dramatic impact on CYP2C9 activity over a prolonged culture period due to the large unbound concentration in the incubation combined with its metabolic stability. However, when the *in vivo* effect was modeled using predicted human PK parameters and dose, only a 5% decrease in *in vivo* CYP2C9 activity was predicted (Fig. 22.14).

### 22.5.3 Key Concepts

The literature regarding IVIVE of pharmacokinetics and drug interactions is vast, not only in terms of predicting human pharmacokinetic parameters but also the impact of potential DDIs and the correct use of *in vitro* data to engender the most accurate prediction of clinical DDIs. A summary of the underlying key concepts not already covered for CYP mediated DDIs is presented here.



**FIGURE 22.13** Model present by Ito et al. [7] for predicting effect of TDI. Seven compartments are described representing the changing inhibitor concentration in systemic blood (compartment 1), the changing inhibitor concentration in the portal vein following an oral dose (compartment 2), the changing inhibitor concentration in the liver (compartment 3), the changing substrate concentration in systemic blood (compartment 4), the changing substrate concentration in the portal vein following an oral dose (compartment 5), the changing substrate concentration in the liver (compartment 6), and the inactivation of hepatic enzyme (compartment 7), where I and S are the concentrations of inhibitor and substrate, respectively;  $I_{sys}$ ,  $S_{sys}$  and  $I_{liver}$ ,  $S_{liver}$  are the concentration in the systemic circulation and liver, respectively;  $V_{sys}$ ,  $V_{sys}$  represent the volumes of distribution in the central compartment;  $K_{pi}$ ,  $K_{ps}$  are the liver-to-blood concentration ratios;  $f_{ubis}$ ,  $f_{ubil}$  are the fraction unbound in blood;  $I_{pv}$ ,  $S_{pv}$  represent the concentrations in the portal vein;  $V_{pv}$  represents the volume of the portal vein;  $Q$  is blood flow,  $CL_r$  represents renal clearance;  $k_a$  is the absorption rate constant;  $F_a$  is fraction absorbed from an oral dose; and  $F_g$  represents the intestinal availability.



**FIGURE 22.14** Effect of CYP2C9 inhibition observed *in vitro* and predicted *in vivo*. (a) CYP2C9 activity in cultured human hepatocytes after incubation with AZ1: CYP2C9-dependent diclofenac 4'-hydroxylation after incubation with AZ1 was determined after incubation with 0.1 μM (solid squares), 1 μM (solid circles), 5 μM (solid triangles), and 10 μM (open circles) AZ1. The dash-dot, solid, dotted, and dashed lines indicate nonlinear regression of the 0.1, 1, 5, and 10 μM AZ1 data, respectively. (b) Simulated profiles of CYP activity in human liver after oral dosing with AZ1 and erythromycin, respectively, using the methodology of Ito et al. [104]: CYP2C9 activity in human liver was simulated after oral dosing with AZ1 (70 mg every 24 h for 72 h open circles) and CYP3A4 activity in human liver after oral dosing with erythromycin (500 mg every 8 h for 72 h closed circles) (From Ref. 95.).

Making quantitative predictions of *in vivo* DDI using *in vitro* data requires consideration not only of inhibitor  $K_i$  (or  $K_I$  and  $k_{inact}$  for mechanism-based inhibition), but also  $f_{uinc}$ ,  $f_{u,b}$ ,  $CL$ , volume,  $fm_{CYP}$ , dose, and  $k_a$ . Use of such factors has resulted in relatively accurate predictions of DDI for reasonably large datasets [75, 88, 139, 147], with some exceptions (possibly compounds with an active component to their



hepatic uptake/efflux—clearly there is more to learn in this area). For detailed predictions of individual DDIs, computer-modeling software should be used to simulate not only the impact of the scenarios detailed earlier, but also the changing drug concentrations with time as a result of the individual PK parameters; this allows a detailed interrogation of the interaction. If a worst-case scenario is to be provided, the maximum unbound inhibitor concentration at the enzyme site should be used. For this, knowledge of the extent to which the inhibitor concentrates in the hepatocyte is important (if no information is available,  $10\times$  the maximum unbound blood concentration has been proposed [7]). Additionally, assuming that elimination of substrate is by a single hepatic CYP predicts a worse-case scenario for the inhibited pathway *in vivo*. These approaches are of utility in a discovery setting when selecting compounds for further progression, but it can be appreciated that it generally leads to an overestimation of the inhibitory effect, especially when combined with the assumption of complete and instantaneous absorption of the inhibitor from an oral dose.

Refinement and improved predictivity is afforded by including the contribution of the inhibited CYP to the total clearance of the substrate ( $fm_{CYP}$ ) and more realistic estimations of the fraction and rate of absorption for the inhibitor, essentially by estimating not only the concentration of inhibitor in the liver more accurately but also its impact on the metabolic pathway of the substrate. Information on the *in vivo*  $fm_{CYP}$  for specific substrates can be found in the literature [68, 138, 151]. As demonstrated by the clinical cases of DDI presented previously, CYP inhibition has greatest impact when a substrate is predominantly cleared by the inhibited pathway; consequently, multiple pathways for drug elimination reduce the chance of clinically significant DDI. However, consideration must be given to the target patient population as, for example, renal impairment can dramatically affect the elimination pathway of a drug.

Recently, an *in vitro* approach aimed at overcoming the difficulty of assessing inhibitor concentration at the enzyme was presented [152]. It involves using human hepatocytes suspended in human plasma to generate  $K_i$  without the need for correction for free concentration at the active site (with the caveat that active transport in hepatocytes *in vitro* may not fully represent that *in vivo*). Using this method, the authors claim accurate prediction of several clinically relevant CYP DDIs by including the CYP phenotypic metabolism for each substrate. This method requires a large range of inhibitor concentrations in the *in vitro* incubations and relies heavily on acute analytical sensitivity. A limitation of this and other methods utilizing  $fm_{CYP}$  is the level of understanding required for each substrate under investigation (optimal concentration range for determining  $K_i$  and an accurate estimation of the contribution of different CYPs to the substrates' *in vivo* metabolism). This method is one approach to minimizing the difficulty of predicting [I] at the active site.

Other points of consideration for IVIVE of DDI include using appropriate substrate-inhibitor combinations *in vitro* so the *in vivo* interaction is correctly assessed; this further emphasizes the importance of using clinically relevant probe substrates in the *in vitro* determinations. Consideration must also be given to the metabolites of an inhibitor, which can often have a significant inhibitory effect in their own right. If they persist in the systemic circulation or are actively concentrated in the liver, their influence can lead to underprediction of the magnitude of the interaction if they are discounted.



There are several pharmacokinetic parameters to be considered for both substrate and inhibitor when making predictions of DDIs. These considerations essentially require estimating levels of drug within the body, be it in the circulating plasma, liver, portal vein, or gastrointestinal tract. Dose and clearance are the primary determinants for these parameters. For drugs in preclinical development, accurate prediction of the relative impact of DDI relies on reasonable assessment of dose level, frequency, and route. It follows that by minimizing dose the potential of DDI will also be minimized. The route of administration of the victim drug must also be considered. A compound with high (intestinal and hepatic) first-pass extraction will behave very differently in the presence of an inhibitor when administered orally as opposed to intravenously. For a drug with a high extraction ratio, the  $C_{\max}$  will be significantly increased in the presence of an inhibitor but the elimination half-life will be quite similar in the presence and absence of inhibitor. Conversely, for a compound with a low extraction ratio, the same inhibitor will have limited effect on  $C_{\max}$  following an oral dose but the elimination half-life may be considerably larger; this will also be the case following an IV dose [130].

Predictions of CYP inhibition made in isolation of other contributing factors, albeit sometimes unknown, may only result in an approximation of *in vivo* effect. The ultimate goal is to be able to predict *in vivo* disposition by encompassing not only multiple mechanisms of CYP (and indeed other enzyme) inhibition but also the potential inductive capacity and transporter interactions for the compound under evaluation. Many drugs exhibit numerous interactive attributes, one such example is the antiviral protease inhibitor ritonavir. *In vitro* ritonavir has been demonstrated to induce CYP3A4 expression through activation of the Pregnane X receptor [153]; it is also a potent mechanism-based inhibitor of CYP3A4 [154] and a substrate and inducer of the P-glycoprotein transporter [155]. Despite accurate *in vitro* assessment of its inhibitory capacity, predictions of the *in vivo* effect of ritonavir are often erroneous due to varying contribution of all these factors and also are dependent on the time course of drug exposure. Initially, ritonavir exerts an inhibitory effect on CYP3A4 mediated metabolism *in vivo* but with prolonged exposure the inductive capacity of the drug predominates. For example, short-term low-dose administration of ritonavir produces a large and significant impairment of triazolam clearance and enhancement of clinical effects [156] with the *AUC* of triazolam increasing from 13.6 to 553 ng/mL·h following 1 day of ritonavir treatment but following 10 days of treatment falling to 287 ng/mL·h [157]. Accurate prediction of such situations becomes difficult. A unifying model allowing all factors to be encompassed, although complex, would be beneficial.

As discussed, a comprehensive, integrated, and systematic approach is required for IVIVE of CYP mediated DDIs. To this end there exists Simcyp® (<http://www.simcyp.com>), a population-based prediction software tool that offers such an evaluation for CYP mediated DDIs. This software incorporates physiological, genetic, and epidemiological information, which, together with *in vitro* data, facilitates the modeling and simulation of the time course and fate of drugs in representative virtual patient populations. Simcyp uses information from routine *in vitro* studies generated in most drug discovery departments and with this information predicts individual pharmacokinetic parameters (such as  $F_{\text{abs}}$ ,  $CL$ , and  $V_{\text{ss}}$ ) and simulates CYP mediated drug interactions and the likely associated population variability. This allows prediction of outcomes in relevant patient populations, identifying those

individuals at most risk from a DDI, not just a single value in an average individual. The software has the potential to simulate the effect of combined CYP induction and inhibition and can also be applied to DDIs involving time-dependent inhibition. Simcyp may assist in the evaluation and optimization of candidate drugs, in the estimation of early human doses and exposures in the clinic, and the prioritization and planning of suitable *in vitro* and clinical interaction studies in the development phase. Tools such as Simcyp are attracting a lot of interest in the pharmaceutical industry and may facilitate a more aligned strategy and as such are worthy of further investment and validation.

#### 22.5.4 FDA Guidelines

The FDA defines a significant DDI as a two fold (or more) increase in the *AUC* of the substrate drug in the presence of an inhibitor. The two fold threshold is given as inter-individual variation and polymorphisms in CYP expression within the population could account for changes less than this [158]. If a risk of DDI is predicted from *in vitro* data, a clinical study evaluating the risk and a thorough investigation of the mechanism of CYP inhibition (along with understanding of the potential coadministered drugs in the target patient population) may make it possible to launch a candidate drug, providing it carries a product label stating possible risk of DDI. For example, if a drug has been determined to be a “strong” inhibitor of CYP3A, a warning about an interaction with “sensitive CYP3A substrates” and CYP3A substrates with narrow therapeutic range may be required [158].

The FDA has released guidance documents, which reflect current thinking in the field specifically aimed at industry evaluating metabolism-based drug interactions; *Guidance for Industry: Drug Metabolism/Drug Interaction Studies in the Drug Development Process—Studies in Vitro*; and *Guidance for Industry: In Vivo Drug Metabolism/Drug Interaction Studies—Study Design, Data Analysis and Recommendations for Dosing and Labelling* [159]. The *in vitro* studies guide, as well as providing a summary of drug metabolism and relevant concepts, describes *in vitro* techniques for evaluating the potential of metabolism-based interactions, the correlation of *in vitro* and *in vivo* findings, and the timing of *in vitro* studies and subsequent product labeling. It defines the goals of evaluating *in vitro* drug metabolism as: (1) to identify all of the major metabolic pathways that affect the test drug and its metabolites, including the specific enzymes responsible for elimination and the intermediates formed; and (2) to explore and anticipate the effects of the test drug on the metabolism of other drugs and the effects of other drugs on its metabolism.

## 22.6 CONCLUSION

Undoubtedly, using *in vitro* data to make accurate predictions of clinical DDIs involving inhibition of CYP is possible providing that there is understanding of the various processes involved. The availability of clinical data is central to building predictive models and is required for feedback on their accuracy and further iteration of these algorithms. Misinterpretation of the risk of CYP inhibition can be

costly both in terms of patient safety and pharmaceutical resources; in a drug discovery setting predicting the worst-case scenario is a prudent approach.

In brief, CYP inhibition may/can result in serious clinical adverse reactions, the prediction of which is a maturing science. Accurate predictions are possible and the pitfalls and progress in using *in vitro* inhibition data to project clinical drug–drug interactions have been presented. Clearly, accurate predictions start with the production of high quality *in vitro* data, require rational examination of the kinetics of inhibition, and, finally, require the application of sound pharmacokinetic theory. The approaches discussed here should guide the researcher in implementing this best practice.

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## APPENDIX A

From Fig. 22.1 it can be seen that

$$d[\text{ES}]/dt = k_1 \cdot [\text{E}] \cdot [\text{S}] - k_{-1} \cdot [\text{ES}] - k_2 \cdot [\text{ES}] \quad (\text{A.1})$$

But at steady state,  $d[\text{ES}]/dt = 0$ ; therefore,

$$k_1 \cdot [\text{E}] \cdot [\text{S}] = k_{-1} \cdot [\text{ES}] + k_2 \cdot [\text{ES}] \quad (\text{A.2})$$

The total enzyme concentration is

$$[\text{E}]_0 = [\text{E}] + [\text{ES}] \quad (\text{A.3})$$

and the total substrate concentration is

$$[\text{S}]_0 = [\text{S}] + [\text{ES}] \quad (\text{A.4})$$

and therefore

$$[\text{E}] = [\text{E}]_0 - [\text{ES}] \quad (\text{A.5})$$

and

$$[\text{S}] = [\text{S}]_0 - [\text{ES}] \quad (\text{A.6})$$

Therefore, substituting Eqs. A.5 and A.6 into Eq. A.2 gives

$$k_1 \cdot [\text{S}]([\text{E}]_0 - [\text{ES}]) = (k_{-1} + k_2) \cdot [\text{ES}] \quad (\text{A.7})$$

Dividing through by  $[\text{ES}]$  gives

$$\frac{k_1 \cdot [\text{E}]_0 \cdot [\text{S}]}{[\text{ES}]} - k_1 \cdot [\text{S}] = k_{-1} + k_2 \quad (\text{A.8})$$

and therefore

$$k_{-1} + k_2 + k_1 \cdot [S] = k_1 \cdot \frac{[E]_0 \cdot [S]}{[ES]} \quad (\text{A.9})$$

and

$$[ES] = \frac{k_1 \cdot [E]_0 \cdot [S]}{k_1 \cdot [S] + k_{-1} + k_2} \quad (\text{A.10})$$

Dividing through by  $k_1$  gives

$$[ES] = \frac{[E]_0 \cdot [S]}{[S] + (k_{-1} + k_2/k_1)} \quad (\text{A.11})$$

since from the reaction scheme in Fig. 22.1 the overall dissociation rate constant for enzyme–substrate complex,  $[ES] = (k_{-1} + k_2)/k_1$ . This is referred to as the Michaelis–Menten constant  $K_m$ . Therefore, Eq. A.11 becomes

$$[ES] = \frac{[E]_0 \cdot [S]}{[S] + K_m} \quad (\text{A.12})$$

Referring again to the reaction scheme in Fig. 22.1, the rate of reaction,  $v$ , or product formation,  $d[P]/dt = k_2 \cdot [ES]$ . Substituting this into Eq. A.12 gives

$$\frac{d[P]}{dt} = v = \frac{k_2 \cdot [E]_0 \cdot [S]}{[S] + K_m} \quad (\text{A.13})$$

where  $[S] \gg K_m$  and  $v$  tends to  $k_2 \cdot [E]_0$  such that the maximal reaction rate can be described as  $V_{\max} = k_2 \cdot [E]_0$ . Equation A.13 therefore becomes

$$v = \frac{V_{\max} \cdot [S]}{K_m + [S]} \quad (\text{A.14})$$

$$= \frac{V_{\max}}{1 + K_m/[S]} \quad (\text{A.15})$$

## APPENDIX B

$$d[EI]/dt = k_{+1} \cdot [I] \cdot [E] - (k_{-1} + k_2) \cdot [EI] \quad (\text{B.1})$$

$$d[EI']]/dt = k_2 \cdot [EI] - (k_3 + k_4) \cdot [EI'] \quad (\text{B.2})$$

$$d[E_{\text{inact}}]/dt = k_4 \cdot [EI'] \quad (\text{B.3})$$

The *initial* rate of enzyme degradation,  $v_{\text{inact}}$ , is described by Eq. B.1, where  $k_{\text{obs}}$  is the apparent inactivation rate constant of the enzyme

$$v_{\text{inact}} = d[\text{E}_{\text{inact}}]/dt = k_4 \cdot [\text{EI}'] = k_{\text{obs}} \cdot [\text{E}]_0 \quad (\text{B.4})$$

It is assumed that  $[\text{E}_{\text{inact}}]$  is negligible before  $[\text{EI}']$  reaches steady state; therefore, the total enzyme concentration,  $[\text{E}]_0$ , remains the same whether it is free, in complex with inhibitor,  $[\text{EI}]$ , or inactivated  $[\text{EI}']$ :

$$[\text{E}]_0 = [\text{E}] + [\text{EI}] + [\text{EI}'] \quad (\text{B.5})$$

Referring to Fig. 22.3, when the levels of the inhibitor-bound enzyme are at steady state,  $d[\text{EI}]/dt = 0$ ,

$$k_1 \cdot [\text{E}] \cdot [\text{I}] = (k_{-1} + k_2) \cdot [\text{EI}] \quad (\text{B.6})$$

therefore,

$$[\text{EI}] = \frac{k_1 \cdot [\text{E}] \cdot [\text{I}]}{k_{-1} + k_2} \quad (\text{B.7})$$

and when the levels of the initial complex formed between metabolic intermediate and enzyme are at steady state,  $d[\text{EI}']/dt = 0$ ,

$$k_2 \cdot [\text{EI}] = (k_3 + k_4) \cdot [\text{EI}'] \quad (\text{B.8})$$

therefore,

$$[\text{EI}'] = \frac{k_2 \cdot [\text{EI}]}{k_3 + k_4} \quad (\text{B.9})$$

Substituting Eq. B.4 into Eq. B.6 gives

$$[\text{EI}'] = \frac{k_2 \cdot k_1 \cdot [\text{I}] \cdot [\text{E}]}{(k_3 + k_4) \cdot (k_{-1} + k_2)} \quad (\text{B.10})$$

Substituting Eqs. B.8 and B.9 into Eq. B.5 gives

$$[\text{E}]_0 = [\text{E}] + \frac{k_1 \cdot [\text{I}] \cdot [\text{E}]}{(k_{-1} + k_2)} + \frac{k_2 \cdot k_1 \cdot [\text{I}] \cdot [\text{E}]}{(k_3 + k_4) \cdot (k_{-1} + k_2)} \quad (\text{B.11})$$

To simplify Eq. B.11, the common denominator  $(k_3 + k_4) \cdot (k_{-1} + k_2)$  is used:

$$[\text{E}]_0 = \frac{(k_3 + k_4) \cdot (k_{-1} + k_2)}{(k_3 + k_4) \cdot (k_{-1} + k_2)} \cdot [\text{E}] + \frac{(k_3 + k_4) \cdot k_1 \cdot [\text{I}] \cdot [\text{E}]}{(k_3 + k_4) \cdot (k_{-1} + k_2)} + \frac{k_2 \cdot k_1 \cdot [\text{I}] \cdot [\text{E}]}{(k_3 + k_4) \cdot (k_{-1} + k_2)} \quad (\text{B.12})$$

$$= [E] \cdot \frac{(k_3 + k_4) \cdot (k_{-1} + k_2) + (k_3 + k_4) \cdot k_1 \cdot [I] + k_2 \cdot k_1 \cdot [I]}{(k_3 + k_4) \cdot (k_{-1} + k_2)} \quad (\text{B.13})$$

Therefore,

$$[E] = \frac{(k_3 + k_4) \cdot (k_{-1} + k_2) \cdot [E]_0}{k_2 \cdot k_1 \cdot [I] + (k_3 + k_4) \cdot k_1 \cdot [I] + (k_3 + k_4) \cdot (k_{-1} + k_2)} \quad (\text{B.14})$$

Substituting Eq. B.13 into Eq. B.10 gives

$$[EI'] = \frac{k_2 \cdot k_1 \cdot [I] \cdot [E]}{(k_3 + k_4) \cdot (k_{-1} + k_2)} \cdot \frac{(k_3 + k_4) \cdot (k_{-1} + k_2) \cdot [E]_0}{k_2 \cdot k_1 \cdot [I] + (k_3 + k_4) \cdot k_1 \cdot [I] + (k_3 + k_4) \cdot (k_{-1} + k_2)} \quad (\text{B.15})$$

$$= \frac{k_2 \cdot k_1 \cdot [I] \cdot [E]_0}{k_2 \cdot k_1 \cdot [I] + (k_3 + k_4) \cdot k_1 \cdot [I] + (k_3 + k_4) \cdot (k_{-1} + k_2)} \quad (\text{B.16})$$

Dividing through by  $k_1$  gives

$$[EI'] = \frac{k_2 \cdot [I] \cdot [E]_0}{k_2 \cdot [I] + (k_3 + k_4) \cdot [I] + ((k_3 + k_4) \cdot (k_{-1} + k_2) / k_1)} \quad (\text{B.17})$$

Then dividing through by  $(k_2 + k_3 + k_4)$  gives

$$[EI'] = \frac{(k_2 / (k_2 + k_3 + k_4)) \cdot [I] \cdot [E]_0}{[I] + \frac{(k_3 + k_4) \cdot (k_{-1} + k_2)}{(k_2 + k_3 + k_4) \cdot k_1}} \quad (\text{B.18})$$

From Eq. B.18, we have

$$k_4 \cdot [EI'] = \frac{(k_2 \cdot k_4 / (k_2 + k_3 + k_4)) \cdot [I] \cdot [E]_0}{[I] + \frac{(k_3 + k_4) \cdot (k_{-1} + k_2)}{(k_2 + k_3 + k_4) \cdot k_1}} \quad (\text{B.19})$$

Substituting Eq. B.17 into Eq. B.3 gives

$$k_{\text{obs}} = \frac{(k_2 \cdot k_4 / (k_2 + k_3 + k_4)) \cdot [I]}{[I] + \frac{(k_3 + k_4) \cdot (k_{-1} + k_2)}{(k_2 + k_3 + k_4) \cdot k_1}} \quad (\text{B.20})$$

$$k_{\text{obs}} = \frac{k_{\text{inact}} \cdot [I]}{[I] + K_I} \quad (\text{B.21})$$

where  $k_{\text{inact}}$  is the maximal inactivation rate constant and  $K_1$  is the inhibitor concentration that supports half the maximal rate of inactivation and, referring to Fig. 22.3, can be described by

$$k_{\text{inact}} = k_2 \cdot k_4 / (k_2 + k_3 + k_4) \quad (\text{B.22})$$

$$K_1 = \frac{(k_3 + k_4) \cdot (k_{-1} + k_2)}{(k_2 + k_3 + k_4) \cdot k_1} \quad (\text{B.23})$$





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# 23

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## ***IN VIVO* METABOLISM IN PRECLINICAL DRUG DEVELOPMENT**

SEVIM ROLLAS

*Marmara University, Istanbul, Turkey*

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### **23.1 INTRODUCTION**

Drug biotransformation plays an important role in the absorption, distribution, metabolism, excretion, and toxicology (ADMET) properties. Undesirable ADMET properties are the cause of many drug development failures [1–3]. Many toxic side effects (carcinogenicity, tissue necrosis, apoptosis, hypersensitivity, teratogenicity) of

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drugs are directly attributable to the formation of chemically reactive metabolites [4–6]. The liver is a major site for the metabolism of xenobiotics (drugs and other exogenous compounds) and endogenous compounds. Other tissues such as kidney, lungs, adrenals, placenta, brain, intestinal mucosa, and skin have some degree of drug-metabolizing capability. Drug metabolism reactions have been divided into two main classes, phase I (functionalization) includes oxidation, reduction, and enzymatic hydrolysis reactions and phase II (conjugation) includes glucuronidation, sulfation, acetylation, methylation, amino acid and glutation conjugation, and mercapturic acid formation reactions. In phase I reactions, the cytochrome P450 (CYP) enzymes are responsible for oxidative biotransformation of many organic compounds [7]. Other drug-metabolizing enzymes, flavin-containing monooxygenase, monoamine oxidase, epoxide hydrolase, UDP-glucuronosyltransferase, glutation *S*-transferase, sulfotransferase, methyltransferase, and *N*-acetyltransferase may also be important when the drug is primarily metabolized by a non-P450 enzyme [8]. Glucuronidation is a well known major drug-metabolizing reaction (pathway) in humans and changes the structure of the parent drug, and thus its chemical and biological reactivity.

Individual differences in drug biotransformation exist. Some people metabolize a drug rapidly so that therapeutically effective blood and tissue levels are not achieved; others metabolize the same drug slowly so that toxicity results. Most CYP enzymes that belong to the CYP 1, 2, or 3 families are polymorphic [9]; so it may be difficult to predict a clinical response to a particular dose of a drug. Genetic polymorphisms have played an important role in these differences [10].

Metabolites can be obtained from plasma, urine, and bile after administration of compounds to laboratory animals and are used in preclinical *in vivo* studies as a reference standard [11–13]. The synthesis of metabolites is mostly done with parent compounds or is performed by total synthesis [14]. In addition, microbial methods can be used to obtain metabolites. The last two methods are very important for active metabolite synthesis.

Prodrug approaches are very valuable in drug development. The goal of prodrug development is to solve specific pharmaceutical or pharmacological problems. The prodrug itself is inactive and is converted to the active drug *in vivo*. Early screening of their *in vitro* and/or *in vivo* plasma stability provides vital information for a stability profile of prodrug candidates [15, 16].

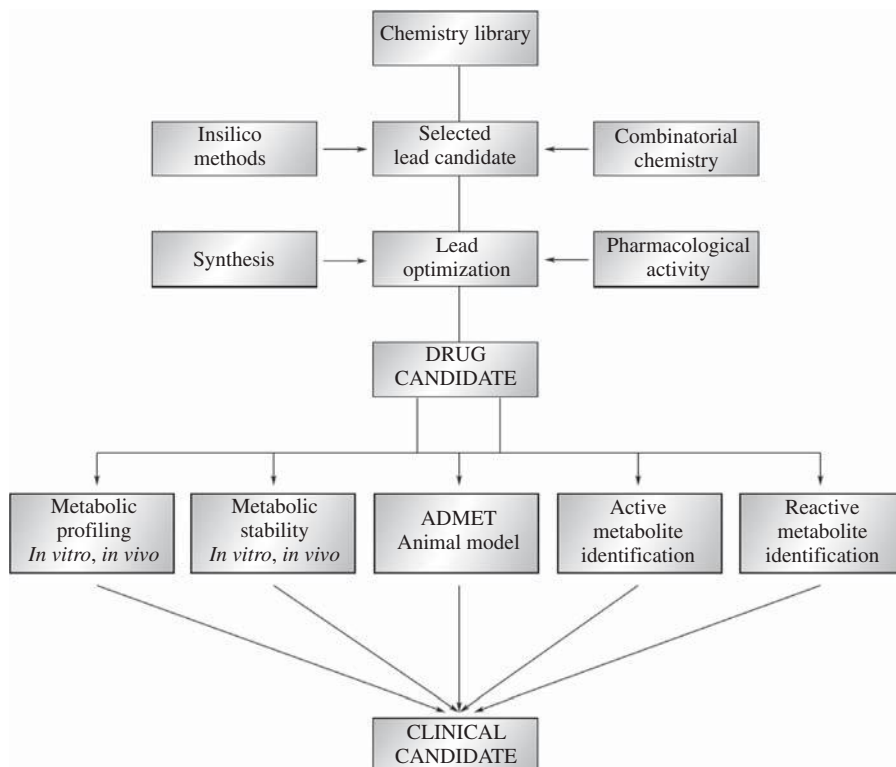
The application of well established preclinical *in vitro* and *in vivo* methodologies, such as gas chromatography–mass spectrometry (GC-MS) and liquid chromatography–mass spectrometry (LC-MS), is essential for drug discovery. Sensitive and specific analytical methods—for example liquid chromatography–high resolution nuclear magnetic resonance spectrometry [17, 18], LC-MS/MS (tandem mass spectrometry) [19–22], liquid chromatography–nuclear magnetic resonance/mass spectrometry (LC-NMR/MS) [23], and quadrupole time-of-flight mass spectrometry [24]—provide the identification and quantitation of metabolites. In general, this chapter is devoted to the preclinical *in vivo* metabolism of drug candidates. In the first part of this chapter, isolation, identification, synthesis, protein binding, and preclinical experimental studies are discussed, while in the following parts, *in vivo* metabolism of prodrugs, acyl glucuronides, and the role of active metabolites are covered.

## 23.2 IN VIVO METABOLISM OF DRUG CANDIDATES

In preclinical drug discovery, investigation of the *in vitro*, *in vivo* metabolism and toxicity screening of a drug candidate together play a very important role in the drug's success before entering into clinical use [25]. Rapid metabolism is one of the most important problems in achieving therapeutic drug levels [26]. Liver microsomal incubation has routinely been carried out by pharmaceutical companies to survey the metabolism of potential drugs. Phase I and phase II drug-metabolizing enzymes affect the overall therapeutic and toxicity profiles of a drug. Several *in vitro* models, such as isolated perfused livers, liver tissue slices, freshly isolated hepatocytes in suspension, primary hepatocyte cultures, cell lines, microsomes, mitochondria, and expression systems, in particular, have been developed to profile human metabolism [1]. *In silico* (computational) techniques are tools that are used to predict the metabolic profiles of drugs that are in the design phase, but there is still insufficient data on the efficacy of *in silico* techniques. That is why *in vivo* metabolism studies should be performed on pharmacophore groups for an *in silico* database. *In vivo* metabolism is still being studied for conventional drugs in humans or animals for the detection of unknown metabolites [27–31]. During drug development, *in vivo* drug absorption, metabolic fate, and hepatic first-pass metabolism studies generally are performed in rats, dogs, and/or monkeys [32–34]. Indeed, the results of *in vitro* metabolism may not be the same as results of *in vivo* metabolism in animals and humans. Reliable human tests should be carried out. Human hepatocytes are used to model biotransformation in the human liver to predict human clearance [35–37]. Dog hepatocytes have also been used. However, the most successful results have been provided using human hepatocytes [38, 39]. Investigating the stereoselective metabolism of optical isomers of drugs is crucial in distinguishing their pharmacokinetic behaviors. The formation of sulfone, hydroxy, and 5-*O*-demethyl metabolites from *S*- and *R*-omeprazole using liver microsomes has been reported [40]. The metabolism of the optical isomers of omeprazole exhibits a significant stereoselectivity. Current strategies in high throughput chemistry are focused on high quality lead compounds [41]. Therefore animal testing is the most important step for lead identification and *in vitro*–*in vivo* correlation. The preclinical evaluation of selected lead compounds is represented in Fig. 23.1.

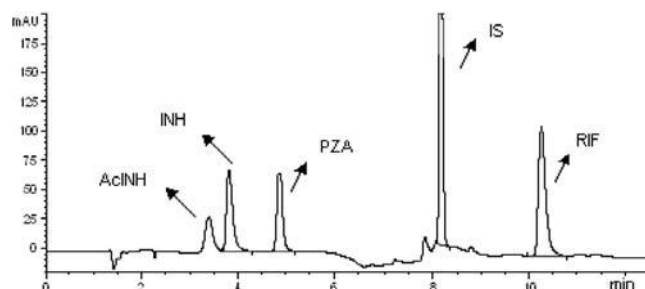
### 23.2.1 Isolation and Identification of Metabolites

Metabolite isolation and identification are important processes in discovery and development of new drugs. In many cases, the isolation and identification of metabolites are difficult because of the low concentration or instability of these compounds in biological matrix. Therefore metabolites are synthesized and characterized by spectroscopic methods. The drug metabolism methodologies may be divided into isolation technologies (e.g., extraction, fractioning, chromatography) and chemical characterization (by means of spectroscopic techniques). For *in vivo* characterization of metabolites, biological samples are collected and it is determined what extraction methods (e.g., preparative solid phase extraction, solid phase extraction, liquid–liquid extraction, or protein precipitation (deproteination)) are necessary for analysis of the samples. Several studies reported use of extraction processes [13, 14,



**FIGURE 23.1** Preclinical evaluation of selected lead candidates.

19, 34, 42–46]. Recently, the use of turbulent flow chromatography with online solid phase extraction and column switching for sample preparation has been reported by Ong et al. [47]. Development of an analytical methodology suitable for accurate determination of both drug and metabolite in biological fluids is an advantage for the examination of the metabolic profile of a drug. The well known technique of liquid chromatography–tandem mass spectrometry (LC-MS/MS), can provide information for the identification for drug metabolites [25, 48]. High resolution nuclear magnetic resonance (NMR) spectroscopy is widely used for structure elucidation of organic compounds and their metabolites in biological samples [49–51]. The most high throughput analytical method for substrate and metabolite analysis is liquid chromatography with parallel NMR and mass spectrometry (LC-NMR/MS) [23, 49]. Timm and colleagues [42] reported another method for the analysis of oral platelet aggregation inhibitor, a double protected prodrug, and its metabolites by using HPLC—column switching combined with turbo ion spray single quadrupole mass spectrometry after precipitation of plasma protein by 0.5M perchloric acid. Deproteinization is a suitable method for sample preparation. Acetonitrile, methanol, perchloric acid, and trichloroacetic acid are used as deproteinization agents [52–56]. Hydrophilic glucuronide conjugates may be hydrolyzed by  $\beta$ -glucuronidase in acetate buffer [28, 29, 57] or by acid or alkali [46, 58, 59] before being analyzed. In recent years, ultra performance liquid chromatography (UPLC) [60] has been used



**FIGURE 23.2** HPLC chromatogram of AcINH (acetylisoniazide), INH (isoniazide), PZA (pyrazinamide), IS (internal standard), and RIF (rifampin). Pore size is  $4\mu\text{m}$  Pore  $3.9 \times 150$  mm long Nova-Pak  $\text{C}_{18}$  column. Signals were monitored by diode array detector. Gradient elution was used.

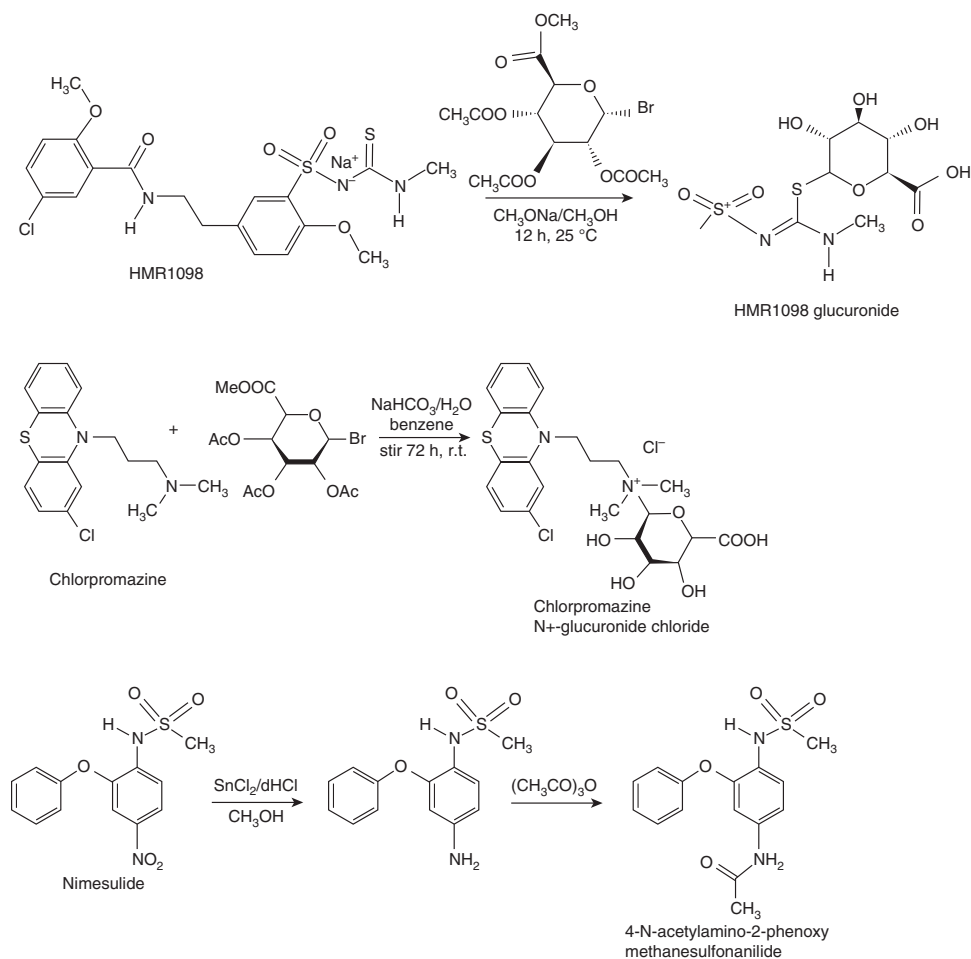
for drug and metabolite identification in biological fluids [61, 62]. Johnson and Plump [62] reported human metabolism of acetaminophen using UPLC with quadrupole time-of-flight mass analysis. Dear et al. [52] also reported a new approach by using UPLC for *in vivo* metabolite identification. These novel fast and successful methods may be applied to preclinical *in vivo* metabolite identification of drug candidates.

Metabolites alter the quantity of drug during pharmacokinetic or drug monitoring studies if an efficient separation is not provided. In our study on the monitoring of isoniazide, pyrazinamide, and rifampicin, for the separation of isoniazid and its major metabolite acetylisoniazid, we developed an HPLC analysis method represented in Fig. 23.2 for the drug monitoring study [55].

### 23.2.2 Synthesis of Metabolites

Metabolite identification and synthesis are important processes in the development of new lead candidates for drug metabolite profiling, pharmacokinetic studies (interference of metabolite), pharmacological activity testing (for active metabolite), metabolite quantification, CYP identification, and toxicity testing. Therefore a synthesis of methods are needed to obtain authentic metabolites. The synthesis of metabolites that are not easily obtained by chemistry methods may be produced by microbial biotransformation of the drugs. Chemical synthesis [14, 63] and microbial methods [64–66] are reported in the literature. Sulfadimethoxine *N*-glucuronide was prepared by Bridges et al. [67] starting from sodium sulfadimethoxine and methyl 2,3,4-tri-*O*-acetyl-1-bromoglucuronate. Ketotifen [68] and soraprazam [69] glucuronides have also been prepared synthetically. The *N*-oxide metabolite of the model compound *N*-benzyl-*N*-methylaniline was prepared by oxidizing with hydrogen peroxide [70]. Sulfinpyrazone sulfide metabolite was prepared by He et al. [71] by the reduction of sulfinpyrazone with iron powder. The hydroxamic acid metabolites are usually obtained from molecules containing an ester functional group [72].

The use of microbial models is a suitable method to produce a sufficient quantity of metabolites [73, 74]. Most bacterial species have P450 enzymes [75]. Enantiomeri-

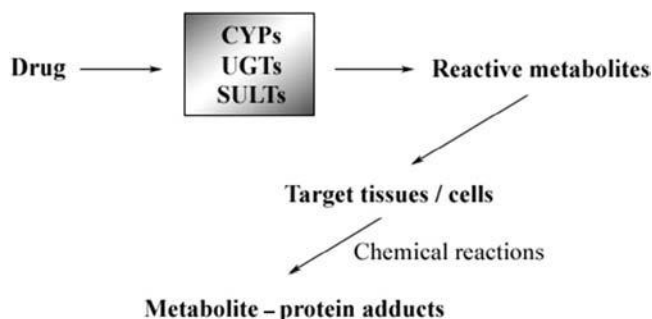


**FIGURE 23.3** Examples for synthesis of some metabolites.

cally pure sulfoxides may be synthesized by biocatalytic methods using microorganisms or isolated enzymes. Ricci et al. [76] prepared aromatic sulfoxides from sulfides employing Basidiomycetes. Zhang et al. [77] reported biotransformation of amitriptyline by the filamentous fungus *Cunninghamella elegans*. The amitriptyline N-oxide metabolite that is produced from fungus can also be obtained by the oxidation of amitriptyline with 3-chloroperoxybenzoic acid to compare with the microbial product. The biosynthesis of drug glucuronides may be performed using suitable animal microsomes to obtain glucuronide metabolites [78]. The synthesis route of some metabolites is shown in Fig. 23.3 [20, 79, 80].

### 23.2.3 Protein Binding of Metabolites

Most drugs and/or their metabolites bind to plasma proteins and other blood components such as red blood cells [81–84]. Generally, free drug molecules can arrive



**FIGURE 23.4** Formation of metabolite–protein adducts of drugs.

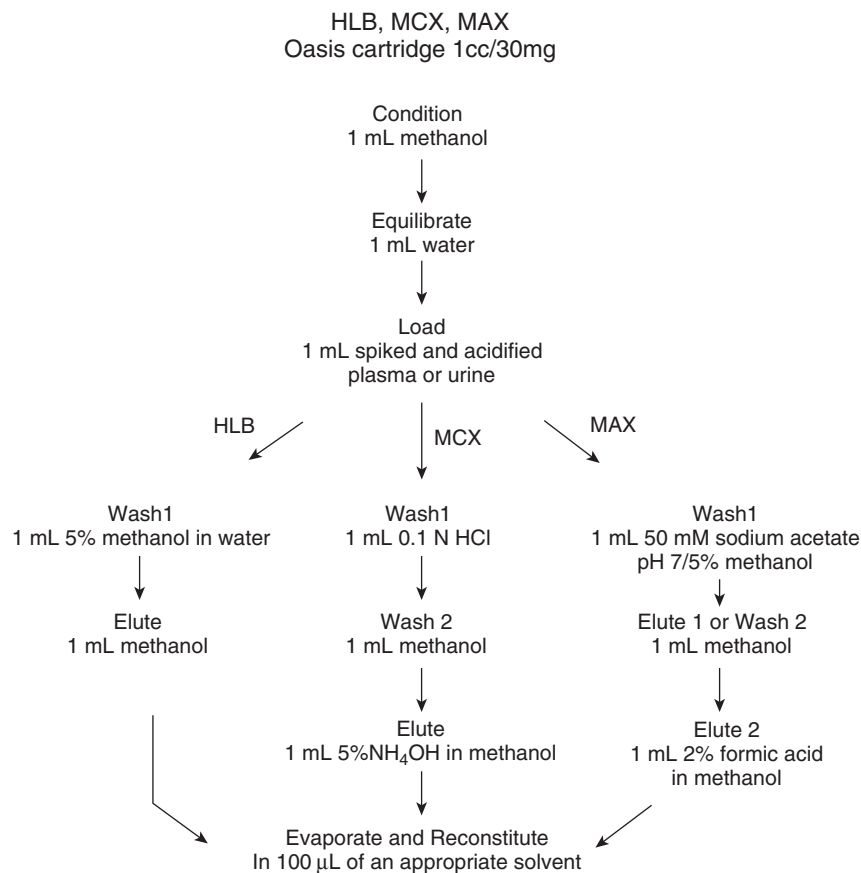
at the site of action. They are distributed into tissues for direct binding to tissue proteins [85]. Binding of a drug to plasma protein influences the drug's ADME processes. Yamazaki and Kanaoka [86] have developed a computational method that can provide precise and useful prediction of plasma protein binding for new drug candidates. However, preclinically, the *in vivo* studies must also be performed on drug candidates. The binding of drugs and/or their metabolites to plasma and tissue proteins is reversible. In some cases, they may covalently bind to plasma and tissue proteins, resulting of drug–protein or metabolite–protein adducts [87–89]. In most cases, the covalent binding by drugs is via their reactive metabolites [90–93]. Covalent binding to tissue proteins has been demonstrated for several metabolites such as acetaminophen–quinoneimine metabolite [94], carbamazepine–10,11-epoxide metabolite [95], furosemide–epoxide metabolite [88], ibuprofen–glucuronide metabolite [96], clozapine–nitrenium ion derived from clozapine N-oxide metabolite [97], and valproic acid–4-ene metabolite [98]. Formation of metabolite–protein adducts of drugs is represented in Fig. 23.4.

#### 23.2.4 Preclinical Experimental Studies

The application of human-based preclinical *in vitro* hepatic metabolism (hepatocyte or liver microsome) is a must for the evaluation of drug metabolism and the selection of drug candidates with a probability of clinical success. However, *in vitro* metabolic stability and metabolite profiling studies should be correlated with *in vivo* experiments with relevant laboratory animal species for prediction of *in vivo* human drug properties [99]. Sample preparation is the most important element in *in vivo* metabolism studies. Conventional solid phase extraction is performed with a C<sub>18</sub> cartridge [82, 100, 101]. Recently, Oasis HLB (hydrophilic–lipophilic balance), MCX (mixed mode, cation-exchange), and MAX (mixed mode, anion-exchange) cartridges have been developed to prepare samples [102–104]. Cleanup of the samples by means of solid phase extraction is shown Fig. 23.5.

Examples of *in vivo* metabolism studies are given next. The metabolite profile of 5-(2-ethyl-2*H*-tetrazol-5-yl)-1-methyl-1,2,3,6-tetrahydropyridine (Lu 25-109), a muscarine agonist, has been reported in mice, rats, dogs, and humans in plasma and



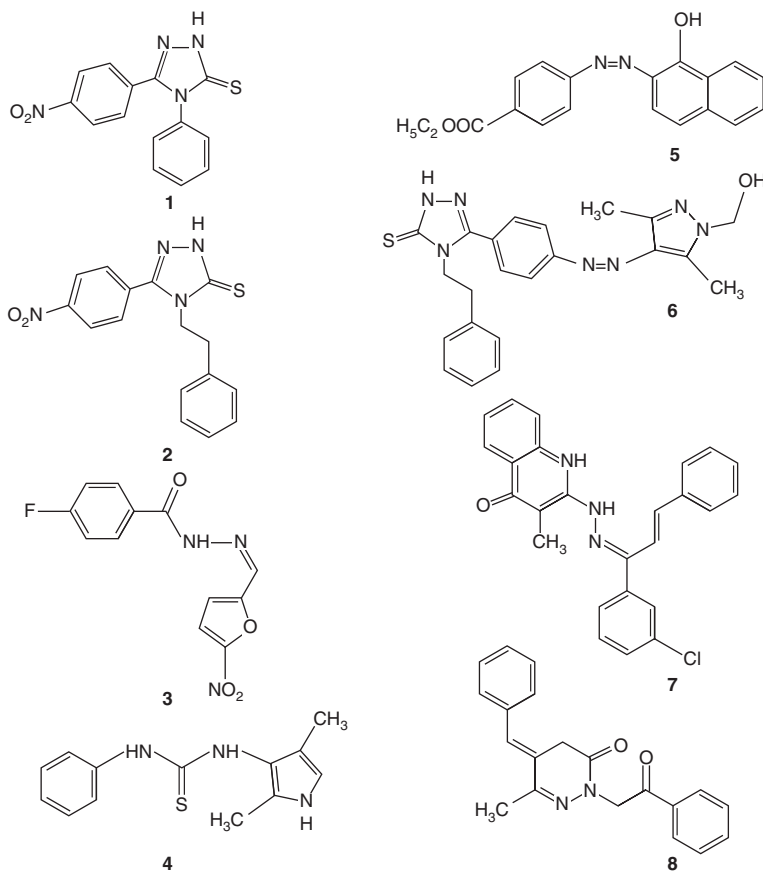


**FIGURE 23.5** Cleanup of the samples by HLB, MCX, and MAX cartridges.

urine after oral administration of [<sup>14</sup>C]Lu 25-109. The *N*-deethyl metabolite is a major metabolite in human plasma. The formation of a pyridine derivative has only been shown in rats [105].

The myocardial binding of the high potency antipsychotic benzisoxazol derivative risperidone and its active metabolite, 9-hydroxyrisperidone, was studied by using equilibrium dialysis *in vitro* and using intraperitoneal administration *in vivo* on plasma and tissue samples of guinea pig by Titier et al [106]. Risperidone and its metabolite concentrations have been determined by the HPLC method with UV detection. The plasma protein and tissue binding of 9-hydroxyrisperidone was lower than that of the more lipophilic risperidone. This method can be applied for other drugs and for therapeutic drug monitoring [106].

In our laboratory, *in vivo* metabolic pathway selected model compounds—1,2,4-triazole-3-thiones [82, 107], hydrazide [108], thiourea derivative [109], azo compounds [110, 111], and quinazolinone [112] and pyridazine [113] rings compounds—were investigated in rats (Fig. 23.6). The potential metabolites of substrate, which cannot be provided, were synthesized. Unknown metabolites were



**FIGURE 23.6** Examples of model compounds investigated for their *in vivo* metabolism: (1) 5-(4-nitrophenyl)-4-phenyl-2,4-dihydro-3H-1,2,4-triazole-3-thione [82]; (2) 5-(4-nitrophenyl)-4-(2-phenylethyl)-2,4-dihydro-3H-1,2,4-triazole-3-thione [107]; (3) 4-fluorobenzoic acid [(5-nitro-2-furanyl)methylene]hydrazide [108]; (4) *N*-phenyl-*N'*-(3,5-dimethyl-pyrazole-4-yl)thiourea [109]; (5) ethyl 4-[(2-hydroxy-1-naphthyl)azo]benzoate [110]; (6) 4-phenylethyl-5-[4-(1-(2-hydroxy-ethyl)-3,5-dimethyl-4-pyrazolylazo)-phenyl]-2,4-dihydro-3H-1,2,4-triazole-3-thione [111]; (7) 2-[1'-phenyl-3'-(3-chlorophenyl)-2'-propenylyden]hydrazino-3-methyl-4(3H)-quinazolinone [112]; (8) 3-oxo-5-benzylidene-6-methyl-(4H)-2-(benzoylmethyl)pyridazine [113].

collected during HPLC studies for mass analysis. Desulfurization reaction did not occur in the 1,2,4-triazole-3-thiones model compounds. Their oxygen analogues were synthesized and used as a reference standard in these studies. Examples of model compounds investigated for their *in vivo* metabolism are shown in Fig. 23.6.

4-Fluorobenzoic acid [(5-nitro-2-furanyl)methylene]hydrazide (substrate) is active against *Staphylococcus aureus* ATCC 29213 (MIC value 3.25 µg/mL) [114]. It was administered intraperitoneally in doses of 50 or 100 mg/kg. Blood samples were collected after administration. The substrate and its potential metabolites were synthesized and separated using HPLC [108]. The substrate and its potential metabolites were not detected in plasma. After plasma proteins were denatured, 4-

fluorobenzoic acid (1.5h) was detected. However, it rapidly metabolizes via a hydrolytic reaction in rats and shows poor metabolic stability *in vivo*.

### 23.3 IN VIVO METABOLISM OF PRODRUGS

Drug activation has been known since the identification of prontosil as a bioreductive prodrug of sulfanilamide. The aim of prodrug development is to solve specific pharmaceutical or pharmacological problems. Various drugs with groups that can be modified by chemical reactions can be converted into prodrugs. They can be synthesized to improve drug stability and absorption, to reduce diverse effects, to extend the duration of action, to increase water solubility or other desirable properties, and to allow site-specific delivery [115]. Generally, synthetic or natural compounds with therapeutic potential as bioactive agents contain alcohol, phenol, ester, thioether, amino, or nitro groups in their structure. In the synthesis of prodrugs, mainly esters, amides, ethers, phosphamides, hydroxamic acids, imines, N-oxides, Mannich bases, azo groups, glycosides, peptides, salts, polymers, and complexes from the parent drug are obtained. The anti-inflammatory drug sulindac is a prodrug and its two active metabolites—sulindac sulfide and sulindac sulfone—result from reduction and oxidation, respectively [116]. This type of prodrug is called a *bioprecursor*. Examples of other bioprecursors are diazepam, primidone, and clofibrate. Quantitative prediction of the effect of a prodrug's enzymatic catalysis *in vivo* is difficult due to the variety and complexity of enzyme systems [117]. The conversion of a prodrug into a parent drug is represented schematically in Fig. 23.7.

#### 23.3.1 Ester and Amide Prodrugs

Esterification and amidation are two important reactions for the synthesis of prodrugs. Many drugs contain free carboxylic acid and/or an amine group. Chandrasekaran et al. [118] reported that the nonsteroidal anti-inflammatory drug indomethacin conjugated with triethylene glycol at the carboxylic acid to obtain its ester and amide prodrugs. Sriram et al. [119] synthesized the amino acid ester prodrug of stavudine for the effective treatment of HIV/AIDS. Wu et al. [120] synthesized lipophilic metronidazol esters using enzymes as catalyst. The best enzymatic transesterification is provided by *Candida antarctica* lipase acrylic resin.

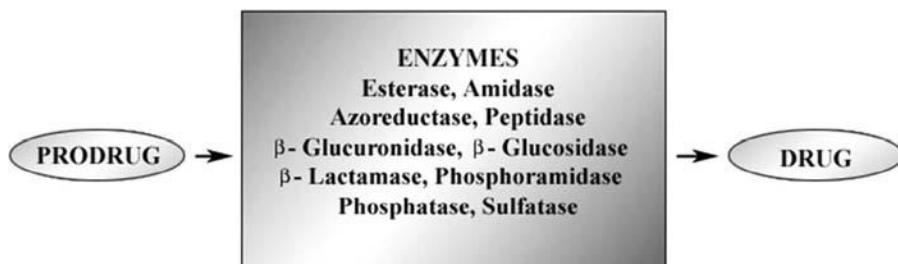


FIGURE 23.7 Conversion of prodrug to parent drug.

### 23.3.2 Other Prodrugs

Generally, prodrugs are designed for dealing with problems like rapid metabolism or poor absorption or reducing diverse effects seen with conventional drugs. In recent years, tumor-activated prodrug (TAP) strategies have become a very important approach in cancer therapy [121]. These prodrugs can be selectively activated in tumor tissue. The aromatic and aliphatic N-oxides, quinones, aromatic nitro groups, and cobalt complexes are the most important structures in developing hypoxia-selective TAPs [122]. Monoclonal antibody–enzyme conjugates (antibody-directed enzyme prodrugs) have recently been developed and rituximab, cetuximab, and trastuzumab are being used in the clinical setting [123,124].  $\beta$ -Lactamase-dependent prodrugs have also been developed for antibody-directed enzyme therapy [125].

Another class of prodrugs is the dendritic prodrugs, which can improve drug delivery and solubility: thus decreasing dosages and prolonging drug release [126–128]. Najlah et al. [129] reported that the design, synthesis, and characterization of polyamidoamine (PAMAM) dendrimer-based prodrugs are described by selecting as the model compound naproxen, a poorly water-soluble drug. Propranolol- $G_3$  PAMAM dendrimer conjugates were also synthesized to increase propranolol solubility [130]. Salamonczyk [131] prepared phosphorus-based dendrimers of acyclovir.

### 23.3.3 Metabolic Stability of Prodrugs

The metabolic stability of a drug candidate in liver microsomes of different animal species is determined in order to assess the potential of this drug to form undesired, potentially toxic, or pharmacologically inactive/active metabolites. In the determination of metabolic stability, *in vitro* models were widely used in order to investigate the metabolic fate of drug candidates. Liver microsomes and S9 (cytosolic and mitochondrial) fractions as drug-metabolizing enzyme sources are used for metabolic stability studies [132]. Di et al. [133] have suggested a single time-point microsomal stability assay and incubation time of 15 min for metabolic stability profiling at the early stage of drug discovery. The duration and/or the intensity of action of most drugs are determined by their rate of metabolism by hepatic enzymes [134]. The metabolic stability procedure involves the following [15, 16, 135]:

- Preparation of buffers
- Preparation of stock solution of test compound (in DMSO or water)
- Incubation (at 37°C)
- Sample preparation
- Determination of loss of parent compound (using LC-MS, LC-MS/MS)

Screening of plasma stability of the prodrugs is very important for rapid conversion in plasma [136]. However, *in vitro* hydrolysis rates of prodrugs of ester or amide must be correlated with *in vivo* hydrolysis rates [137]. Lai and Khojasteh-Bakht [138] reported a new method for prodrug stability—an automated online liquid chromatographic–mass spectrometric method.

### 23.4 ACYL GLUCURONIDES

Glucuronidation reaction represents the major route for elimination and detoxification of drugs. The biosynthesis of glucuronides is catalyzed by the uridine diphosphate glucuronosyltransferases, which are localized in hepatic endoplasmic reticulum. However, the glucuronidation of some drugs has been observed in human kidney microsomes [135]. Seventeen human UDP-glucuronosyltransferases have been identified to date [139]. The mechanism of glucuronidation is a nucleophilic substitution reaction and the resulting glucuronide has the 1 $\beta$ -glycosidic configuration.

The rate of glucuronidation is dependent on the nucleophilic character of the substrate and enzyme concentration. The most systematic investigations were initiated in the 1930s by R. T. Williams and collaborators [140, 141]. Acyl glucuronide metabolites of carboxylic acid may be hydrolyzed under physiological conditions. Therefore the glucuronides produced can potentially interfere with the pharmacokinetics of the parent drug. In this case, early pharmacokinetic study is important in animal models to regulate dosing and duration of action.

On the other hand, glucuronide conjugates are capable of cellular injury (e.g., hepatotoxicity, carcinogenesis) by facilitating the formation of reactive electrophilic intermediates and their transport into target tissues [87]. Acyl glucuronides are chemically unstable in aqueous solution and undergo the intramolecular acyl migration. The acyl group may migrate to the C-2, C-3, or C-4 position of the glucuronic acid from the 1 $\beta$  position [142]. In addition, unstable acyl glucuronides have also been shown to form covalent adducts with proteins under *in vivo* conditions. The existence of genetic polymorphism in the rate of drug acylation has important consequences in drug therapy. Chen et al. [143] developed a rapid *in vitro* method for the assessment of the stability of acyl glucuronides. Many nonsteroidal anti-inflammatory drugs, such as diclofenac, diflunisal, zomepirac, and naproxen, and other drugs, such as valproic acid, citalopram, and furosemide, which bear the carboxylic functional group, have been metabolized to form acyl glucuronides [49, 144, 145].

### 23.5 ROLE OF ACTIVE METABOLITES IN PRECLINICAL DRUG DEVELOPMENT

Drugs can metabolize inactive, reactive, or pharmacologically active molecules [146, 147]. Fura et al. [148] and Guengerich et al. [149] define an active metabolite as the active metabolic product possessing the same pharmacological activity as the parent drug. Pharmacologically active metabolites should be recognized at the drug discovery stage, because the potential contribution of an active metabolite to total activity is an important factor for pharmacokinetic properties of a drug candidate. Therefore active metabolite kinetics should be studied in animal models. Klein et al. [150] reported a population pharmacokinetic model for irinotecan and two of its metabolites, SN-38 and SN-38 glucuronide. The active metabolite 7-hydroxycamptothecin (SN-38) is 300–1000 times more potent than its parent molecule irinotecan. Therefore it is important to develop dosing strategies for irinotecan. Mak and Weglicki [151] reported on the antioxidant properties of 4-hydroxypropranolol, a major metabolite of propranolol. It is four- to eightfold more potent than vitamin E. As a result, 4-hydroxypropranolol may contribute to the cardiovascular therapeutic

benefits of propranolol because of its endothelial cytoprotective efficacy against cell injury.

Active metabolites play a role as lead candidates during lead optimization [148, 152]. For example, ezetimibe was found during the lead optimization phase [152]. During drug development, it is sometimes impossible to characterize the pharmacological activity of active metabolites because of the difficulties in their synthesis by common methods. They may be more active than their respective parent molecules. However, a number of active metabolites have been marketed as drugs: for example, acetaminophen, cetirizine, imipramine, desloratadine, fexofenadine, mesoridazole, nortriptyline, and oxazepam [148].

Metabolism of the antifungal drug itraconazole (ITZ) involves both phase I (oxidative) and phase II (conjugative) pathways. One of these metabolites, hydroxyitraconazole (OH-ITZ), possesses antifungal activity similar to that of ITZ [153]. Therefore, the pharmacokinetic behavior of ITZ may be affected by its active metabolites. Gasparro et al. [154] reported that the active metabolite *R*-desmethyl-deprenyl of *R*-deprenyl has been shown to be more active than *R*-deprenyl. Ciclesonide, an inhaled corticosteroid with prolonged anti-inflammatory activity, is being developed for the treatment of asthma. Nave et al. [155] reported ciclesonide is converted to an active metabolite, desisobutyryl-ciclesonide, which undergoes reversible fatty acid conjugation in *in vitro* studies. Cannell et al. [144] demonstrated that nonsteroidal anti-inflammatory drugs (diflunisal, zomepirac, and diclofenac), acyl glucuronides, and their rearrangement isomers had antiproliferative activity on human adenocarcinoma HT-29 cells in culture.

In conclusion, active metabolites may:

- Alter pharmacokinetic properties of parent compounds.
- Cause enzyme induction or inhibition.
- Alter pharmacokinetics of drugs when used together.
- Show different pharmacological activity.
- Be toxic.

### 23.6 CURRENT APPROACHES FOR EVALUATING PRECLINICAL *IN VIVO* METABOLISM

Metabolism is one of the most important steps in ADMET studies. Metabolites affect pharmacokinetic results [156]. *In vivo* pharmacokinetic behavior involves permeability, metabolic stability, inhibition, and induction of CYP isoenzymes and plasma protein binding. A new approach may be the use of ultra performance liquid chromatography [52], online hyphenated LC-NMR/MS [48], and an integrated high capacity solid phase extraction-MS/MS system [157] for *in vivo* metabolite identification in the future. Metabolism is also one of the most complicated of the pharmacokinetic properties. Genetic, environmental, and physiological factors may affect drug metabolism. Computational approaches for predicting CYP-related metabolism is very important for the evaluation of a successful new drug [158]. Therefore, CYP inhibitors, CYP inducers, CYP regioselectivity, and CYP substrates have been studied for the development of a reliable predictive model.

Computational or virtual ADME [2, 26, 159–162] and metabolism models derived from past *in vitro* and *in vivo* data have been developed to contribute to new drug discovery in the pharmaceutical industry [163, 164]. Klopman et al. [165] evaluated a computer model for prediction of intestinal absorption in humans. This model may be applied to particularly active metabolites. Recent studies have shown that one of the current trends will be drug-induced autoimmunity. Idiosyncratic drug reactions that do not occur in most patients are an important clinical problem [166]. Most of these reactions appear to be immune-mediated reactions. It is necessary to develop new animal models, in addition to the present models, for preclinical studies. However, the newest approaches of drug discovery may be the mechanism-based, the function-based, and the physiology-based approaches [167].

### 23.7 CONCLUSION

In the last few years, much effort in drug discovery and development has focused on defining the metabolic profile, toxicity, stereoselective metabolism, and pharmacokinetics of new drug candidates. Understanding the role of drug-metabolizing enzymes in drug metabolism is an important area of research for human toxicology testing. *In vitro* and *in vivo* models have been used as part of high throughput screening programs to characterize lead identification, drug metabolic profiling, metabolic stability, drug permeability, drug solubility, and drug safety. *In vitro* studies generally provide a valuable insight into a drug's metabolic profile in humans; however, sometimes there are discrepancies between *in vitro* results and *in vivo* findings. In the future, the use of *in silico* predictive models and ultra developed analysis techniques for preclinical *in vivo* studies will be an essential requirement for the discovery and development of successful new drugs.

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# 24

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## ***IN VITRO* EVALUATION OF METABOLIC DRUG–DRUG INTERACTIONS: SCIENTIFIC CONCEPTS AND PRACTICAL CONSIDERATIONS**

ALBERT P. LI

*In Vitro* ADMET Laboratories Inc., Columbia, Maryland

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## 24.1 INTRODUCTION

Simultaneous coadministration of multiple drugs to a patient is highly probable. A patient may be coadministered multiple drugs to allow effective treatment of a single disease (e.g., cancer, HIV infection) or multiple disease or disease symptoms. It is now known that drug–drug interactions may have serious, sometimes fatal, consequences. Serious drug–drug interactions have led to the necessity of a drug manufacturer to withdraw or limit the use of marketed drugs. Examples of fatal drug–drug interactions are shown in Table 24.1. As illustrated by the examples in Table 24.1, a major mechanism of adverse drug–drug interactions is the inhibition of the metabolism of a drug by a coadministered drug, thereby elevating the systemic burden of the affected drug to a toxic level.

Besides toxicity, loss of efficacy can also result from drug–drug interactions. In this case, the metabolic clearance of a drug is accelerated due to the inducing effects of a coadministered drug on drug metabolism. A well known example is the occurrence of breakthrough bleeding and contraceptive failures in women taking oral contraceptives who were coadministered the enzyme inducer rifampin [7]. Examples of drug–drug interactions leading to the loss of efficacy are shown in Table 24.2.

Estimation of drug–drug interaction potential is therefore an essential element of drug development. Screening for drug–drug interaction in early phases of drug development allows the avoidance of the development of drug candidates with high potential for adverse drug interactions. Estimation of drug–drug interaction potential is a regulatory requirement—it is required for New Drug Applications (NDAs) to the U.S. FDA [11]. In this chapter, the scientific principles, technologies, and experimental approaches for the preclinical evaluation of drug–drug interactions are reviewed.

## 24.2 MECHANISMS OF ADVERSE DRUG–DRUG INTERACTIONS

Adverse effects in a patient due to coadministration of multiple drugs can be due to pharmacological or pharmacokinetic drug–drug interactions defined as follows:

*Pharmacological Interactions.* Adverse effects that occur due to combined pharmacological activities, leading to exaggerated pharmacological effects. An example of pharmacological interaction is the serious, sometimes fatal, drop in blood pressure due to coadministration of nitroglycerin and Sildenafil [12].

**TABLE 24.1 Drugs that Have Been Withdrawn from the Market Due to Fatal Interactions with Coadministered Drugs**

Drug–Drug Interaction	Mechanism of Interactions	References
Terfenadine–ketoconazole interaction, leading to fatal arrhythmia (torsades de pointes). Terfenadine was withdrawn from the market in January 1997 and replaced by a safer alternative drug (fexofenadine), which is the active metabolite of terfenadine.	Terfenadine is metabolized mainly by CYP3A4 and has been found to interact with CYP3A4 inhibitors (e.g., ketoconazole), leading to elevation of plasma terfenadine level that reached cardiotoxic levels.	1–3; <a href="http://www.fda.gov/bbs/topics/answers/ans00853.html">www.fda.gov/bbs/topics/answers/ans00853.html</a>
Mibefradil interaction with multiple drugs, leading to serious adverse effects. Mibefradil interactions with statins has led to rhabdomyolysis. Mibefradil was withdrawn from the market in June 1998, less than a year after it was introduced to the market in August 1997.	Mibefradil is a potent CYP3A4 inhibitor known to elevate the plasma levels of over 25 coadministered drugs to toxic levels. Statins, especially simvastatin and cerivastatin, are known to cause rhabdomyolysis.	4; <a href="http://www.fda.gov/bbs/topics/answers/ans00876.html">www.fda.gov/bbs/topics/answers/ans00876.html</a>
Sorivudine–5-fluorouracil (5-FU) interaction, leading to severe or fatal gastrointestinal and bone marrow toxicities. Sorivudine was withdrawn from the market in 1993.	Sorivudine inhibits dihydropyrimidine dehydrogenase, an enzyme pathway responsible for fluoropyrimidine metabolism.	5
Gemfibrozil–cerivastatin interaction, leading to rhabdomyolysis. Cerivastatin was withdrawn from the market in August 2001.	Inhibition of cerivastatin metabolism by gemfibrozil, apparently due to CYP2C8 inhibitory effects of gemfibrozil.	6; <a href="http://www.fda.gov/medwatch/safety/2001/Baycol2.html">www.fda.gov/medwatch/safety/2001/Baycol2.html</a>

*Pharmacokinetic Interactions.* Adverse effects that occur due to altered body burden of a drug as a result of a coadministered drug that can occur because of the ability of one drug to alter the absorption, distribution, metabolism, and excretion (ADME properties) of the coadministered drug. Of the ADME properties, drug metabolism represents the most important and prevalent mechanism for pharmacokinetic interactions.

### 24.3 DRUG METABOLISM

All drugs administered to a patient are subject to biotransformation. Orally administered drugs are first subjected to metabolism by the intestinal epithelium and, upon

**TABLE 24.2 Drug-Drug Interactions Leading to Loss of Efficacy**

Drug-Drug Interaction	Mechanism	References
Oral contraceptive-rifampin interactions, leading to breakthrough bleeding and contraceptive failure	Rifampin accelerates the metabolism of the estrogenic component (e.g., 17 $\alpha$ -ethinylestradiol) of oral contraceptives via induction of the metabolizing enzymes (CYP3A4 and estrogen sulfotransferases)	7, 8
Cyclosporin-rifampin interaction, leading to rejection of transplanted organs	Rifampin induces CYP3A, leading to accelerated metabolic clearance of cyclosporine to nonimmunosuppressive level	9
St. John's wort (SJW) interactions with prescribed drugs, leading to loss of efficacy	SJW ( <i>Hypericum perforatum</i> ) is an herbal medicine found to contain ingredients that can induce CYP3A4, CYP2C9, CYP1A2, and various transporters, leading to clinically observed accelerated metabolic clearance and/or loss of efficacy of a large number of drugs including warfarin, phenprocoumon, cyclosporine, HIV protease inhibitors, theophylline, digoxin, and oral contraceptives. The incidents with SJW illustrate the importance of the evaluation of potential drug-drug interaction of herbal medicines.	10

absorption into the portal circulation, are metabolized by the liver before entering the systemic circulation. While multiple tissues have certain degrees of biotransformation capacity, it is generally accepted that hepatic metabolism represent the most important aspect of drug metabolism.

Drug metabolism can be classified into the following major categories:

*Phase I Oxidation.* This generally is described as the addition of an oxygen atom (e.g., as a hydroxyl moiety) to the parent molecule. Phase I oxidation is carried out by multiple enzyme pathways, including the various isoforms of the cytochrome P450 (CYP) family and the non-P450 biotransformation enzymes such as flavin-containing monooxygenase (FMO) and monamine oxidase (MAO).

*Phase II Conjugation.* Phase II conjugation represents enzyme reactions that lead to the addition of a highly water-soluble molecule to the chemical that is being metabolized, leading to highly water-soluble "conjugates" that allow efficient excretion. Examples of phase II enzymes are UDP-glucuronosyl transferase (UGT), sulfotransferase (ST), and glutathione-S-transferase (GST). Conjugation reactions often occur with the hydroxyl moiety of the parent structure or with the oxidative metabolites.

The major drug-metabolizing enzymes and subcellular locations are summarized in Table 24.3.



**TABLE 24.3 Major Pathways for Drug Metabolism, Enzymes, Subcellular Locations, and *In Vitro* Experimental System Containing the Enzymes<sup>a</sup>**

Major Classification	Enzyme	Subcellular Location	Representative <i>In Vitro</i> Experimental System
Phase I oxidation	Cytochrome P450 mixed function monooxygenases	Endoplasmic reticulum	Microsomes, S9, hepatocytes
	Monoamine oxidase	Mitochondria	Hepatocytes
	Flavin-containing monooxygenase	Endoplasmic reticulum	Microsomes, S9, hepatocytes
	Alcohol/aldehyde dehydrogenase	Cytosol	S9, hepatocytes
	Esterases	Cytosol and endoplasmic reticulum	Microsomes, S9, hepatocytes
Phase II conjugation	UDP-dependent glucuronyl transferase	Endoplasmic reticulum	Microsomes, S9, hepatocytes
	Phenol sulfotransferases; estrogen sulfotransferase	Cytosol	S9, hepatocytes
	N-acetyl transferase	Endoplasmic reticulum	Microsomes, S9, hepatocytes
	Soluble glutathione-S-transferases (GST)	Cytosol	S9, hepatocytes
	Membrane-bound GST	Endoplasmic reticulum	Microsomes, S9, hepatocytes

<sup>a</sup>These enzymes are grouped into phase I oxidation and phase II conjugation enzymes, although it is now believed that such classification may not be possible for all drug-metabolizing enzymes. Representative *in vitro* experimental systems containing these enzymes are shown to guide the selection of the most relevant approach for specific enzyme pathways. It is apparent that intact hepatocytes represent the most complete *in vitro* system for drug metabolism studies as they contain all the key hepatic drug-metabolizing enzyme pathways.

## 24.4 CYP ISOFORMS

Cytochrome P450-dependent monooxygenases (CYP) are the drug-metabolizing enzymes often involved in metabolic drug–drug interactions. The CYP family is represented by a large number of isoforms, each having selectivity for certain chemical structures. The major hepatic human CYP isoforms are CYP1A2, CYP2A6, CYP2B6, CYP2C8, CYP2C9, CYP2C19, CYP2D6, CYP2E1, and CYP3A4. Of these isoforms, the CYP3A isoforms are the most important in drug metabolism. CYP3A isoforms (CYP3A4 and CYP3A5) collectively represent the most abundant hepatic CYP isoforms (approximately 26%), followed by CYP2C isoforms (approximately 17%). In terms of the isoforms involved in drug metabolism, CYP3 isoforms are known to be involved in the metabolism of the most number of drugs (approximately 33%), followed by CYP2C isoforms (approximately 25%) [13].

P450 isoforms are known to have specific substrates, inhibitors, and inducers (Table 24.4).

**TABLE 24.4 Major Human P450 Isoforms<sup>a</sup> Involved in Drug Metabolism**

CYP Isoform	Substrate	Inhibitor	Inducer
CYP1A2	Phenytoin	Furafylline	Omeprazole
CYP2A6	Coumarin	Tranlycypromine	Rifampin
CYP2B6	Bupropion	Ticlopidine	Rifampin
CYP2C8	Taxol	Quercetin	Rifampin
CYP2C9	Tolbutamide	Sulfaphenazole	Rifampin
CYP2C19	S-mephenytoin	Omeprazole	Rifampin
CYP2D6	Dextromethorphan	Quinidine	None
CYP2E1	Chlorzoxazone	Diethyldithiocarbamate	Ethanol
CYP3A4	Testosterone	Ketoconazole	Rifampin

<sup>a</sup>The individual isoforms and examples of isoform-specific substrates, inhibitors, and inducers are shown.

## 24.5 HUMAN *IN VITRO* EXPERIMENTAL SYSTEMS FOR DRUG METABOLISM

Substantial species–species differences occur in drug metabolism pathways, especially for CYP isoforms. Because of the species–species differences, human *in vitro* hepatic experimental systems rather than nonhuman animals are viewed as the most relevant to the evaluation of xenobiotic properties, including human drug metabolism and metabolism-based drug–drug interactions [14–17]. The following are the commonly used *in vitro* experimental systems for the evaluation of metabolism-based drug–drug interactions.

### 24.5.1 Hepatocytes

Hepatocytes are the parenchymal cells of the liver, which are responsible for hepatic biotransformation of xenobiotics. Isolated hepatocytes represent the most physiologically relevant experimental system for drug metabolism studies as they contain all the major hepatic drug-metabolizing enzyme pathways that are not physically disrupted such as cell free fractions. Furthermore, the drug-metabolizing enzymes and cofactors in the hepatocytes are present at physiological concentrations. Freshly isolated hepatocytes and cryopreserved hepatocytes are generally believed to represent the most complete *in vitro* system for the evaluation of hepatic drug metabolism [18].

In the past, the use of human hepatocytes has been severely limited by their availability, as studies would be performed only if human livers were available for hepatocyte isolation. Furthermore, hepatocyte isolation from human livers is not a technology available to most drug metabolism laboratories. This limitation has been overcome in the past decade due to the advancements in the procurement of human livers for research, and the commercial availability of isolated human hepatocytes. The application of human hepatocytes in drug metabolism studies is also greatly aided by the successful cryopreservation of human hepatocytes to retain drug metabolism activities [8, 19, 20]. Recently, the usefulness of cryopreserved human hepatocytes was further extended through the development of technologies to cryopreserve human hepatocytes to retain their ability to be cultured as attached cultures (plateable cryopreserved hepatocytes), which can be used for longer term

**TABLE 24.5 Viability and Plateability (Ability of Hepatocytes to Be Cultured as Monolayer Cultures) of the Various Lots of Cryopreserved Human Hepatocytes<sup>a</sup>**

Lot	Yield (cells/vial)	Viability (trypan blue)	Plating	Confluency
HU4003	4.5 × 10 <sup>6</sup>	86%	Yes	100%
HU4001	6.0 × 10 <sup>6</sup>	80%	No	20%
HU4004	6.0 × 10 <sup>6</sup>	80%	No	30%
HU4000	7.2 × 10 <sup>6</sup>	93%	Yes	100%
HU4013	7.3 × 10 <sup>6</sup>	92%	Yes	75%
HU4016	6.2 × 10 <sup>6</sup>	81%	Yes	100%
HU4021	5.4 × 10 <sup>6</sup>	89%	Yes	70%
HU4022	5.5 × 10 <sup>6</sup>	91%	Yes	80%
HU4026	5.85 × 10 <sup>6</sup>	91%	No	10%
HU4027	5.9 × 10 <sup>6</sup>	92%	No	30%
HU4028	3.2 × 10 <sup>6</sup>	83%	Yes	50%
HU4023	2.1 × 10 <sup>6</sup>	89%	No	20%
HU4029	6.0 × 10 <sup>6</sup>	90%	Yes	80%

<sup>a</sup>Hepatocytes manufactured by APSciences Inc. in partnership with CellzDirect Inc.

studies such as enzyme induction studies [21]. Examples of the viability and plateability of cryopreserved human hepatocytes prepared in our laboratory are shown in Table 24.5.

#### 24.5.2 Liver Postmitochondrial Supernatant (PMS)

Liver PMS is prepared by first homogenizing the liver, and then centrifuging the homogenate at a speed of either 9000*g* or 10,000*g* to generate the supernatants S9 or S10, respectively. Liver PMS contains both cytosolic and microsomal drug-metabolizing enzymes but lacks mitochondrial enzymes.

#### 24.5.3 Human Liver Microsomes

Liver microsomes are the 100,000*g* pellet for the PMS. Microsome preparation procedures in general involve the homogenization of the liver, dilution of the homogenate with approximately 4 volumes of sample weight with a buffer (e.g., 0.1 M Tris-HCl, pH 7.4, 0.1 M KCl, 1.0 mM EDTA, 1.0 mM PMSF [22]) followed by centrifugation at 9000–14,000*g* to remove nonmicrosomal membranes, and then at 100,000–138,000*g* to pellet the microsomes [23]. Microsomes contain the smooth endoplasmic reticulum, which is the site of the major phase I oxidation pathway, the P450 isoforms, and esterases, as well as a major conjugating pathway, UGT.

#### 24.5.4 Recombinant P450 Isoforms (rCYP)

These are microsomes derived from organisms transfected with genes for individual human P450 isoforms (e.g., bacteria, yeast, mammalian cells) [24–26] and therefore contain only one specific human isoform. The major human P450 isoforms involved in drug metabolism are available commercially as rCYP. This experimental system is widely used to evaluate the drug-metabolizing activities of individual P450 isoforms [15, 16].

**TABLE 24.6 Comparison of the Key *In Vitro* Drug-Metabolizing Experimental Systems: Liver Microsomes (Microsomes), Liver Postmitochondrial Supernatant (S9), Liver Cytosol (Cytosol), and Hepatocytes and Their Major Drug-Metabolizing Enzymes<sup>a</sup>**

<i>In Vitro</i> System	P450	MAO	UGT	ST	GST
Microsomes	+	–	+ <sup>b</sup>	–	+ <sup>c</sup>
S9	+	–	+ <sup>b</sup>	+ <sup>b</sup>	+
Cytosol	–	–	– <sup>b</sup>	+ <sup>b</sup>	+ <sup>d</sup>
Hepatocytes	+	+	+	+	+

<sup>a</sup>P450, cytochrome P450; MAO, monoamine oxidase; UGT, UDP-glucuronosyl transferase; ST, sulfotransferase; GST, glutathione-S-transferase.

<sup>b</sup>Activity of this drug-metabolizing enzyme requires the addition of specific cofactors, for instance, UDP-glucuronic acid (UDPGA) for UGT activity, and 3'-phosphoadenosine 5'-phosphosulfate (PAPS) for ST activity.

<sup>c</sup>Membrane-bound GST but not the soluble GST is found in the microsomes.

<sup>d</sup>Soluble GST but not membrane-bound GST are found in the cytosol.

### 24.5.5 Cytosol

The supernatant after the 100,000g centrifugation for microsome preparation is the cytosol, which is practically devoid of all membrane associated enzymes. *N*-acetyl transferases, sulfotransferases, and dehydrogenases are examples of cytosolic enzymes. While drug–drug interaction studies are mainly studied using liver microsomes, there are cases of drug–drug interactions involving phase II pathways that can be studied using liver cytosol [27].

A comparison of the different *in vitro* experimental systems and their drug-metabolizing enzymes are shown in Table 24.6.

## 24.6 MECHANISMS OF METABOLIC DRUG–DRUG INTERACTIONS

Metabolic drug–drug interaction results from the alteration of the metabolic clearance of one drug by a coadministered drug. There are two major pathways of metabolic drug–drug interactions:

*Inhibitory Drug–Drug Interaction.* When one drug inhibits the drug metabolism enzyme responsible for the metabolism of a coadministered drug, the result is a decreased metabolic clearance of the affected drug, resulting in a higher than desired systemic burden. For drugs with a narrow therapeutic index, this may lead to serious toxicological concerns. Most fatal drug–drug interactions are due to inhibitory drug–drug interactions.

*Inductive Drug–Drug Interactions.* Drug–drug interactions can also be a result of the acceleration of the metabolism of a drug by a coadministered drug. Acceleration of metabolism is usually due to the induction of the gene expression, leading to higher rates of protein synthesis and therefore higher cellular content of the induced drug–metabolizing enzyme and a higher rate of metabolism of the substrates of the induced enzyme. Inductive drug–drug interactions can lead to a higher metabolic clearance of the affected drug, leading to a decrease in plasma concentration and loss of efficacy. Inductive drug–drug interactions can also lead to a higher systemic burden of metabolites, which, if toxic, may lead to safety concerns.

## 24.7 MECHANISM-BASED APPROACH FOR THE EVALUATION OF DRUG–DRUG INTERACTION POTENTIAL

Due to the realization that it is physically impossible to evaluate empirically the possible interaction between one drug and all marketed drugs, and that most drug-metabolizing enzyme pathways are well defined, a mechanism-based approach is used for the evaluation of drug–drug interaction potential of a new drug or drug candidate [15, 16, 28]. This mechanism-based approach is now also recommended by the U.S. FDA ([www.fda.gov/cber/gdlns/interactstud.htm](http://www.fda.gov/cber/gdlns/interactstud.htm)). The approach consists of the following major studies.

*Metabolic Phenotyping.* Metabolic phenotyping is defined as the identification of the major pathways involved in the metabolism of the drug in question. The reasoning is that if the pathways are known, one can estimate potential interaction of the drug in question with known inhibitors or inducers of the pathway.

*Evaluation of Inhibitory Potential for Drug-Metabolizing Enzymes.* The ability of the drug in question to inhibit the activities of known pathways for drug metabolism is evaluated. If a drug is an inhibitor of a drug-metabolizing enzyme pathway, it will have the potential to cause inhibitory drug interactions with coadministered drugs that are substrates of the inhibited pathway.

*Induction Potential for Drug-Metabolizing Enzymes.* The ability of the drug in question to induce drug-metabolizing enzyme activities is evaluated. If the drug in question is an inducer of a specific pathway, it will have the potential to cause inductive drug interactions with coadministered drugs that are substrates of the induced pathway.

## 24.8 EXPERIMENTAL APPROACHES FOR THE *IN VITRO* EVALUATION OF DRUG–DRUG INTERACTION POTENTIAL

Because of the known species–species differences in drug metabolism, it is now believed that *in vitro*, human-based, experimental systems are more appropriate than nonhuman animal models for the evaluation of drug–drug interactions. *In vitro* positive findings are usually confirmed with *in vivo* clinical studies. The typical preclinical studies for drug–drug interactions [15, 16, 28] ([www.fda.gov/cber/gdlns/interactstud.htm](http://www.fda.gov/cber/gdlns/interactstud.htm)) are described next.

### 24.8.1 Study 1: Metabolic Phenotyping 1—Metabolite Identification

The objective of this study is to identify the major metabolites of the drug in question. For this study, the drug in question is incubated with an appropriate *in vitro* metabolic system to allow the formation of metabolites [15, 16]. Metabolites are then identified using analytical chemical approaches. The *in vitro* experimental system of choice is human hepatocytes, with high performance liquid chromatography/mass spectrometry (HPLC/MS) or tandem mass spectrometry (HPLC/MS/MS) as the most convenient analytical tool to identify the metabolites.

The metabolites are generally identified as metabolites of phase I oxidation or phase II conjugation. If phase I oxidation is concluded as the major pathway for the

oxidative metabolism of the drug, Experiment 2 will be performed to evaluate which of the several oxidative pathways are involved. Phase II conjugation pathways can generally be identified by the identity of the metabolite and subsequent experiments to further identify the pathways may not be necessary. For instance, if the metabolite is a glucuronide, UGT can be identified as the enzyme involved.

A typical experimental design is as follows:

- *In vitro System.* Cryopreserved human hepatocytes pooled from two donors (male, female).
  - Three drug concentrations: 1, 10, and 100  $\mu$ M.
  - Hepatocyte concentration: 0.5–1.0 million hepatocytes per mL.
  - Three incubation times: 1, 2, and 4 hours (suspension culture); up to 24 hours (attached culture).
  - Incubation in 24 well plates at 37°C.
  - Organic solvent (e.g., acetonitrile) to terminate reaction and to extract medium and intracellular metabolites.
  - Stored frozen until analysis.
- *Analytical Chemistry.* HPLC-MS/MS.
  - Quantification of disappearance of parent chemical in all samples.
  - Identification of metabolites from 100  $\mu$ M samples.
  - Detection of metabolites in 1 and 10  $\mu$ M samples.

#### 24.8.2 Study 2: Metabolic Phenotyping 2—Identification of Major Metabolic Pathways

If oxidative metabolites are found to be the major metabolites, it is necessary to evaluate which major oxidative pathways are involved in the metabolism. This is performed via the use of liver microsomes and experimental conditions that would inhibit a specific pathway. The major pathways and experimental conditions are shown in Table 24.7.

As P450 pathways are considered the most important for metabolic drug–drug interactions, the study with the general P450 inhibitor, 1-aminobenzotriazole (ABT)

**TABLE 24.7 Experimental Conditions to Reduce the Activity of the Major Drug-Metabolizing Enzyme Pathways (P450 Isoforms (CYP), Flavin-Containing Monooxygenases (FMOs), and Monoamine Oxidase (MAO)) Using the *In Vitro* Experimental Systems for Drug Metabolism (Liver Microsomes (Microsomes), Postmitochondrial Supernatant (S9), and Hepatocytes)**

<i>In Vitro</i> System	Condition	Inactivated Pathway(s)
Microsomes	NADH omission	CYP, FMO
Microsomes or hepatocytes	1-Aminobenzotriazole treatment	CYP
Microsomes	Heat (45°C) inactivation	FMO
S9	Pargyline treatment	MAO

Source: Adapted from <http://www.fda.gov/cder/guidance/6695dft.pdf>.

is the one that should be performed. ABT is known to inhibit all eight human P450 isoforms involved in drug metabolism [29]. Inhibition of metabolism of a test article by ABT would indicate that the test article is metabolized by the P450 pathway. A typical study with ABT is as follows:

- Human liver microsomes (0.5 mg protein/mL).
- Experiment 1: Evaluation of experimental conditions for the accurate quantification of metabolic clearance.
  - Incubation with three concentrations of test article (e.g., 0.1, 1, 10  $\mu$ M) and three incubation times (e.g., 15, 30, and 60 minutes).
  - Quantification of test article disappearance.
- Experiment 2: Reaction phenotyping.
  - Incubation with one concentration of the test article at one incubation time (chosen from Experiment 1) in the presence and absence of three concentrations of ABT (100, 200, and 500  $\mu$ M).
  - Quantification of test article disappearance and evaluation of the effects of ABT treatment.

### 24.8.3 Study 3: Metabolic Phenotyping 3—Identification of P450 Isoform Pathways (P450 Phenotyping)

If ABT is found to inhibit the metabolism of the drug or drug candidate in Study 2, P450 metabolism is ascertained. The next step is to identify which P450 isoforms are involved in the metabolism, a process termed P450 phenotyping [30, 31]. There are several major approaches for this study.

***Liver Microsome and Isoform-Selective Inhibitors*** In this experiment, the test article is incubated with human liver microsomes in the presence and absence of individual selective inhibitors for the eight major CYP isoforms. The ability of an inhibitor to inhibit metabolism of the test article would indicate that the pathway inhibited by the inhibitor is involved in metabolism. For instance, if ketoconazole, a potent CYP3A4 inhibitor, is found to inhibit the metabolism of the test article, then CYP3A4 is concluded to be involved in the metabolism of the test article. It is also a common practice to assign the degree of involvement by the maximum percent inhibition. For instance, if the maximum inhibition, expressed as percentages of the total metabolism in the absence of inhibitor, by sulfaphenazole (CYP2C9 inhibitor) and ketoconazole (CYP3A4 inhibitor) are 20% and 80%, respectively, it can be concluded that the CYP2C9 is involved in 20% and CYP3A4 in 80% of the metabolism of the test article. It is important to realize that the inhibitors are isoform selective rather than isoform specific, so data interpretation must be performed carefully to avoid an inaccurate assignment of enzyme pathways [31]. It is always prudent to confirm the results of this study with results using a different approach (e.g., using rCYP).

***Incubation with Individual rCYPs*** In this experiment, individual rCYP isoforms are used to evaluate which P450 isoforms are involved in the metabolism [30]. The



test article is incubated with each rCYP and its disappearance quantified. A rCYP that leads to the disappearance of the test article would indicate that the isoform is involved in the metabolism of the test article. For instance, if rCYP2C19 incubation leads to the disappearance of the test article, then CYP2C19 is concluded to be involved in the metabolic clearance of the test article. It is important to realize that these studies are performed with a single P450 isoform and therefore lack competing enzyme pathways. Metabolism by a rCYP isoform may not be relevant *in vivo* because of higher affinity pathways.

**Correlation Study with Human Liver Microsomes** In this experiment, the test article is incubated with multiple lots of human liver microsomes that have been previously characterized for the activities of the individual CYP isoforms [32]. The rate of metabolic clearance of the test article is then plotted against the CYP activities of the different lots of microsomes. A linear correlation between activity and rate of disappearance for a specific CYP would indicate that this pathway is involved in the metabolism of the test article. This study requires the evaluation of at least 10 liver microsome lots with well-distributed gradations of activities.

**Liver Microsome/Inhibitor Study Design** In general, studies with liver microsomes are believed to be more relevant than studies with rCYP, as studies with individual rCYP does not allow competition in metabolism for isoforms with different affinities for the substrate and therefore may overemphasize the participation of low affinity pathways. It is important to use substrate concentrations similar to expected plasma concentrations. An artifactually high concentration would cause the substrate to be metabolized similarly by high and low affinity enzyme pathways [33]. Using liver microsomes with physiologically relevant substrate concentrations should provide the best results. A typical liver microsome experiment with inhibitors is as follows.

- Human liver microsomes (0.5 mg/mL).
- Experiment 1: Metabolic stability study.
  - Incubation with three concentrations of test article (e.g., 0.1, 1, 10  $\mu$ M) and three incubation times (e.g., 15, 30, and 60 minutes).
  - Quantification of test article disappearance.
- Experiment 2: Reaction phenotyping.
  - Incubation with one concentration of the test article at one incubation time (chosen from Experiment 1) in the presence and absence of isoform-specific inhibitors.
  - Quantification of test article disappearance.

The isoform-specific inhibitors suggested by the U.S. FDA are shown in Table 24.8.

**Evaluation of CYP Isoform Contributions Using Both Liver Microsomes and rCYP Isoforms** It is also possible to calculate the relative contribution of individual isoforms using data from both liver microsomes and rCYP isoforms using the following approach [34, 35]:

**TABLE 24.8 Preferred and Acceptable P450 Isoform-Specific Inhibitors Suggested by U.S. FDA in the September 2006 Draft Guidance Document for Drug–Drug Interaction Evaluation and Preferred Inhibitors Used in the *In Vitro* ADMET Laboratories (IVAL)**

CYP	FDA Preferred Inhibitor	FDA Acceptable Inhibitor	IVAL Preferred Inhibitor
1A2	Furafylline	$\alpha$ -Naphthoflavone	Furafylline
2A6	Tranlycypromine, methoxsalen	Pilocarpine, tryptamine	Tranlycypromine
2B6		Ticlopidine, sertraline	Ticlopidine
2C8	Quercetin	Trimethoprim, gemfibrozil, rosiglitazone	Quercetin
2C9	Sulfaphenazole	Fluconazole	Sulfaphenazole
2C19		Ticlopidine	Omeprazole
2D6	Quinidine		Quinidine
2E1		Diethyldithiocarbamate	Diethyldithiocarbamate
3A4/5	Ketoconazole, itraconazole	Troleandomycin, verapamil	Ketoconazole

- The relative activity factor for individual isoforms (using isoform-specific substrates) is calculated first. This is necessary as each lot of liver microsomes would have different relative amounts of each P450 isoform.  $V_{\max}$  and  $K_m$  values are determined for each isoform using isoform-specific substrates for both liver microsomes and rCYP isoforms. The relative activity factor (RAF) is calculated using the following equation:

$$RAF = \frac{(V_{\max}/K_m \text{ of CYP in microsomes})}{V_{\max}/K_m \text{ of rCYP}}$$

- Contribution of a specific CYP isoform to the metabolism of a test article is then calculated using the following equation:

$$\text{Contribution of CYP (\%)} = RAF \times \frac{V(\text{rCYP})}{V(\text{microsomes})}$$

#### 24.8.4 Study 4: CYP Inhibitory Potential

The objective of this study is to evaluate if the drug or drug candidate in question is an inhibitor of a specific P450 isoform. This study can be performed with rCYP, human liver microsomes, and human hepatocytes.

**rCYP Studies** rCYP studies represent the most convenient and rapid study for the evaluation of CYP inhibitory potential. As the study involves substrates that form metabolites that can be quantified by fluorescence, the laborious and time-consuming HPLC or LC/MS sample analysis is not required. For this reason, most drug development laboratories perform rCYP inhibition assays as a screen for P450

inhibitory potential of their drug candidates. The study involves the incubation of individual rCYP isoforms with the test article at various concentrations (e.g., seven concentrations plus solvent control) in triplicate, and a substrate that can be metabolized by the specific isoform. As the reaction contains only one isoform, isoform-specific substrates are not required to be used. The requirement is that the substrate generates metabolites that can be measured by a plate reader with the capability to quantify fluorescence.

**Liver Microsome Studies** Liver microsomes represent the most appropriate experimental system for the evaluation of the interaction of a drug with P450 isoforms. For the evaluation of CYP inhibitory potential, the test article is incubated with liver microsomes in the presence of individual isoform-specific substrates. The isoform-specific substrates and the metabolites quantified are shown in Table 24.9.

**Human Hepatocyte Studies** rCYP and human liver microsomes are cell-free systems, allowing direct interaction of the test article with the P450 isoforms. *In vivo*, the inhibitor is initially absorbed into the systemic circulation and then interacts with the enzymes after penetration through the hepatocyte plasma membrane. Once inside the cytoplasm, the inhibitor may be metabolized by phase I and/or phase II metabolism and/or actively transported out of the hepatocytes, for instance, via bile excretion. Furthermore, there may be transporters present to actively uptake the inhibitor. The result is that the intracellular concentration of the inhibitor may be substantially different from the plasma concentration. Results with rCYP and human liver microsomes may not be useful to estimate *in vivo* inhibitory effects based on plasma concentrations if the intracellular concentration of the inhibitor is not known.

The use of intact human hepatocytes may allow a more accurate extrapolation of *in vitro* results to *in vivo*. The study is performed using intact human hepatocytes incubated with isoform-specific substrate and the test article. The intact plasma membrane and the presence of all hepatic metabolic pathways and cofactors allow distribution and metabolism of the test article. The resulting inhibitory effect there-

**TABLE 24.9 P450 Isoform-Specific Substrates and Their Respective Metabolites<sup>a</sup>**

CYP	Substrate	Metabolite
1A2	Phenacetin	Acetaminophen
	Ethoxyresorufin	Resorufin
2A6	Coumarin	7-OH-coumarin
2B6	Bupropion	Hydroxypropion
2C8	Taxol	6- $\alpha$ -Hydroxypaclitaxel
2C9	Tolbutamide	4'-Hydroxytolbutamide
2C19	S-Mephenytoin	4-OH-mephenytoin
2D6	Dextromethorphan	Dextrophan
2E1	Chlorzoxazone	6-Hydroxychlorzoxazone
3A4/5	Testosterone	6- $\beta$ -Hydroxytestosterone

<sup>a</sup>These substrates are used in *in vitro* experimental systems such as liver microsomes, liver S9, or hepatocytes in which multiple isoforms are expressed.

fore should be physiologically more relevant to the *in vivo* situation than results with cell-free systems.

It is recommended that inhibition studies with intact hepatocytes be performed if inhibitory effects of a drug or drug candidate have been observed with rCYP or liver microsomes to allow a more accurate prediction of the extent of *in vivo* inhibitory effects. Time-dependent inhibition of P450 can also be studied using intact human hepatocytes [36]. One precaution with the use of intact hepatocytes is to concurrently measure cytotoxicity. As dead hepatocytes are not active in drug metabolism, without cytotoxicity information, cytotoxic drug concentrations could be interpreted as inhibitory concentrations.

A recent advancement is to use intact hepatocytes suspended in whole human plasma for inhibition studies to allow correction for plasma protein binding [37]. As drugs *in vivo* are always in contact with 100% human blood, this is conceptually sound and therefore deserves further investigation on its general applicability. One disturbing finding in our laboratory is that testosterone, a compound that is readily metabolized *in vivo*, is not metabolized by intact human hepatocytes in whole plasma (A. P. Li, unpublished).

***IC<sub>50</sub>, K<sub>i</sub>, K<sub>inact</sub> and [I]/K<sub>i</sub> Determinations*** Enzyme inhibition data are often presented as IC<sub>50</sub>, the concentration of the inhibitor to cause 50% inhibition at one chosen substrate concentration; K<sub>i</sub>, the inhibition constant (dissociation constant from the inhibitor–enzyme complex) determined by enzyme kinetic analysis (e.g., Dixon plot); and K<sub>inact</sub>, the time-dependent inhibition constant for mechanism-based inhibitors. IC<sub>50</sub> values can be estimated from the study described earlier. A positive inhibition, defined as dose-dependent inhibition, with the inhibited activity lower than 50% of that of the negative control, will require further experimentation to define K<sub>i</sub> for a better evaluation of *in vivo* inhibitory potential. Furthermore, study to determine K<sub>inact</sub> may be performed to evaluate if the inhibitor acts via covalent binding to the active site of the enzyme, leading to time-dependent irreversible inhibition.

IC<sub>50</sub> is generally determined by plotting the log of the relative activity (activity in the presence of the inhibitor as a percentage of the activity of the negative solvent control), and then estimating the concentration yielding 50% relative activity using linear regression analysis. IC<sub>50</sub> can also be calculated from the relationship between inhibitor concentrations and percent control activity with the aid of a nonlinear regression program such as SCIENTIST (Micromath, Salt Lake City, UT) [38].

K<sub>i</sub> can be determined using a Dixon plot with the reciprocal of the activity as the y-axis and inhibitor concentration as the x-axis. Results with at least two substrate concentrations below V<sub>max</sub> are plotted, with K<sub>i</sub> calculated as the negative of the x-intercept [39]. K<sub>i</sub> can also be estimated with the aid of nonlinear regression analysis software such as SYSTAT (SPPS, Inc., Chicago, IL) [40].

Most P450 inhibitors act via reversible (competitive or noncompetitive) mechanisms and their inhibitory potential can be estimated from their IC<sub>50</sub> or K<sub>i</sub> values. Some inhibitors are “mechanism-based” or “time-dependent” inhibitors, which can cause irreversible inhibition due to the formation of reactive metabolites by the CYP isoform, leading to covalent binding to the active site and thereby causing

irreversible inhibition of the affected enzyme molecule [41]. Irreversible inhibitors therefore will have prolonged inhibition of the enzyme even after clearance of the drug in question.  $K_{\text{inact}}$  is a measure of the potency of such “mechanism-based” inhibitors.

$K_{\text{inact}}$  can be determined using the following approach [42]:

1. Plot the relative activity (activity in the presence of the inhibitor as a percentage of the activity of the solvent or negative control) versus time and determine the slope at each inhibitor concentration.
2. Plot (1/slope) versus (1/inhibitor concentration) (Kitz–Wilson plot).  $K_{\text{inact}}$  is calculated as the reciprocal of the  $y$ -intercept, and  $K_i$  as the negative of the reciprocal of the  $x$ -intercept.

$[I]/K_i$ , the ratio of the anticipated or known steady-state plasma drug concentration to  $K_i$ , is generally used to determine the likelihood of clinical drug–drug interactions [43, 44]. A general rule of thumb suggested by the U.S. FDA (<http://www.fda.gov/cder/guidance/6695dft.pdf>) is as follows:

- $[I]/K_i < 0.1$ : Unlikely to cause *in vivo* drug–drug interactions.
- $[I]K_i = 1$ : Possible to cause *in vivo* drug–drug interactions.
- $[I]/K_i > 1$ : Likely to cause *in vivo* drug–drug interactions.

$K_i$  is estimated by an experiment with varying inhibitor and substrate concentrations. A typical  $K_i$  study is as follows:

- *In vitro* experimental system: rCYP, human liver microsomes or hepatocytes.
- Inhibitor concentrations: Five (ideally yielding 10–90% inhibition of activity).
- Substrate concentrations: Minimum of two for the Dixon plot; three is recommended.
- Timepoint: one (within the linear time course) if time course is known; multiple (e.g., 5, 10, and 15 minutes) if time course under the experimental conditions has not been established.
- $K_i$  is determined by a Dixon plot, plotting the reciprocal of activity versus inhibitor concentration. The negative of the  $x$ -coordinate value corresponding to the intercept of the plots for the low and high substrate concentrations is the  $K_i$ .

For mechanism-based inhibitors,  $K_{\text{inact}}$  is estimated by an experiment with varying inhibitor concentration and preincubation time. A typical  $K_{\text{inact}}$  study is as follows:

- *In vitro* experimental system: rCYP, human liver microsomes or hepatocytes.
- Preincubation time (preincubation of enzyme with inhibitor): Five values (e.g., 5, 10, 15, 20, 30 minutes).
- Inhibitor concentration: Five (ideally yielding 10–90% inhibition of activity).
- Substrate concentration: one.

- Substrate incubation time: one (within the linear time course) if time course is known; multiple (e.g., 5, 10, and 15 minutes) if time course under the experimental conditions has not been established.
- $K_{\text{inact}}$  is determined by the following approach:
  - Plot activity as a percentage of the solvent control versus time.
  - Estimate the first-order inactivation constants at each inhibitor concentration by multiplying the slope of the linear regression analysis by 2.303.
  - Determine  $t_{1/2}$  of the inactivation reaction as  $0.693/k$ .
  - Plot the Kitz–Wilson plot of  $t_{1/2}$  versus the reciprocal of the inhibitor concentration and estimate  $K_{\text{inact}}$  as the  $y$ -intercept, and  $K_i$  as the reciprocal of the  $x$ -intercept.

#### 24.8.5 Study 5: Enzyme Induction Potential

Enzyme induction is a major mechanism for drug–drug interactions. Induction of a drug-metabolizing enzyme by one drug would lead to the enhanced metabolism of coadministered drugs that are substrates of the induced enzyme.

Experimental evaluation of enzyme induction involves the treatment of human hepatocytes for several days with the test article followed by evaluation of enzyme activities using P450 isoform-specific substrates [45, 46]. As freshly isolated hepatocytes possess endogenous activities, which may be the result of inducers present in the donor’s systemic circulation, the isolated hepatocytes are cultured for 2–3 days to allow the P450 enzyme activities to return to a basal level. Testing for induction potential is initiated by treatment of the cultured hepatocytes for 2–3 days to allow full expression of the induced enzyme. Induction is generally evaluated by measuring enzyme activity, as activity represents the most relevant endpoint for drug–drug interaction. Both freshly isolated and plateable cryopreserved human hepatocytes can be used for the induction study [21, 47, 48].

As of this writing, all known inducers of P450 isoforms *in vivo* are inducers *in vitro* [21]. The known human P450 inducers are shown in Table 24.10.

**TABLE 24.10 Clinically Demonstrated Human Enzyme Inducers and Their Respective *In Vitro* Induction Results as Well as Their Association with Severe Hepatotoxicity**

<i>In Vivo</i> Enzyme Inducer	<i>In Vitro</i> Human Hepatocyte Induction Finding	Severe Clinical Hepatotoxicity
Carbamazepine	+	+
Dexamethasone	+	–
Isoniazid	+	+
Omeprazole	+	+
Phenobarbital	+	+
Phenytoin	+	+
Rifampin	+	+
Rifapentine	+	–
Rifabutin	+	–
Troglitazone	+	+
St. John’s wort	+	+

The typical experimental procedures for an enzyme induction study are as follows:

- Day 0: Plate human hepatocytes (freshly isolated or plateable cryopreserved human hepatocytes).
- Day 1: Refresh medium.
- Day 2: Refresh medium.
- Day 3: Change medium to that containing test article, solvent control, or positive controls.

Minimum of three test article concentrations, with the high concentration at least one order of magnitude greater than expected plasma concentration.

If plasma concentration not known, evaluate concentrations ranging over at least two orders of magnitude (e.g., 1, 10, 100  $\mu$ M).

- Day 4: Refresh treatment medium.
- Day 5: Refresh treatment medium.
- Day 6: Measure activity (*in situ* incubation with isoform-specific substrates).

The isoform-specific substrates described earlier for CYP inhibition studies are generally used for enzyme induction studies.

The known CYP inducers are now known to induce either CYP1A and/or CYP3A, with inducers of other inducible isoforms such as CYP2A6, CYP2C9, and CYP2C19 found also to be CYP3A inducers. For general enzyme induction evaluation for drug–drug interaction, it may be adequate to simply screen for CYP1A and CYP3A induction. If CYP3A induction is observed, then investigations into CYP2A6, CYP2C9, and CYP2C19 induction are warranted.

The two most common confounding factors for P450 induction studies are as follows:

1. *Inducers that Are Also Inhibitors.* The co-occurrence of P450 inhibition and induction (i. e., the compound is both an inhibitor and inducer) can confound induction results. Ritonavir is an example of a CYP3A4 inducer [49] that is also a potent CYP3A4 inhibitor [50]. The inhibitory effects can overcome any induction effects using activity as an endpoint. For the evaluation of enzyme induction potential of inhibitors, Western blotting for the amount of enzyme proteins would be most appropriate. Studies with mRNA expression would provide data to distinguish between induction of gene expression or protein stabilization as mechanisms. As in the case of ritonavir, induction effects persist after the clearance of the drug from the systemic circulation, leading to enhanced clearance of drugs that are substrates of the induced pathways. It is important to define the induction potential of a drug even if it is found to be an enzyme inhibitor.
2. *Cytotoxic Compounds.* Induction effects can be masked by the decrease of cell viability, as most induction assays quantify substrate metabolism *in situ* (in the same cell culture plate in which the cells are cultured) and assume that there is no change in cell number. Cytotoxicity evaluation therefore should always be performed concurrently with induction studies. In the presence of cytotoxicity, activity should be corrected by the viability for comparison with negative control activity to assess induction potential.



A compound is concluded to be an inducer if reproducible, statistically significant, and dose-dependent induction effects are observed. U.S. FDA recommends the use of the criterion of “40% or higher of the activity of positive controls” as a positive response ([www.fda.gov/cber/gdlns/interactstud.htm](http://www.fda.gov/cber/gdlns/interactstud.htm)).

#### 24.8.6 Study 6: *In Vitro* Empirical Drug–Drug Interactions

The physiological significance of the findings based on the mechanistic approach may be substantiated by *in vitro* drug–drug interactions between frequently coadministered drugs that are likely to have interaction with the drug in question [28]. This is particularly important if the drug in question is either a CYP3A4 substrate or a CYP3A4 inhibitor. As CYP3A4 is now known to have different affinities for different substrates and inhibitors [51], the interaction potential for a drug and a particular coadministered drug may be substantially different from that estimated by using a surrogate substrate of CYP3A4.

This study can be performed with liver microsomes or hepatocytes. The use of hepatocytes probably would allow the development of data more relevant to humans *in vivo*.

### 24.9 DATA INTERPRETATION

The studies described previously allow one to develop data for the estimation of drug–drug interaction potential of the drug or drug candidate in question. Accurate prediction of *in vivo* effects is possible only through thorough and scientifically sound interpretation of the data. While every novel chemical structure will provide a unique set of data and therefore requires individualized data interpretation and/or further experimentation, the following guidelines can be used to aid the evaluation of the data generated.

#### 24.9.1 Pathway Evaluation

Possible outcomes of a study are as follows:

1. The test article is not metabolized by liver microsomes or hepatocytes. This is indicated by the lack of metabolite formation and lack of parent disappearance in Studies 1 and 2. Hepatic metabolism is not involved in the metabolic clearance of the compound. There should be no concern with coadministered drugs that can alter drug-metabolizing enzyme activities.
2. The test article is metabolized but not metabolized by P450 isoforms. As P450-related drug–drug interactions are the most prevalent, non-P450 drug–drug interactions should be considered on a case-by-case basis. For instance, MAO interaction may be important if the drug in question may be coadministered with known MAO substrates or inhibitors. UGT substrates, for instance, may have drug interactions with UGT inhibitory drugs such as probenecid.
3. The test article is metabolized by a single P450 isoform. This represents the easiest data to interpret, albeit not a good scenario for a drug candidate. A drug that is metabolized predominantly by a single P450 isoform will very likely have drug–drug interactions with inhibitors of the isoform. The known cases of serious drug–drug interactions often involve a single P450 pathway,

with CYP3A4 being the most prominent. Drugs that have been withdrawn due to fatal drug–drug interactions are often CYP3A4 substrates or potent CYP3A4 inhibitors. Because of the role of CYP2C8 in the metabolism of statins, which are widely prescribed to combat hypercholesterolemia, CYP2C8 has become the second-most important isoform for drug–drug interactions. Cerivastatin, a CYP2C8 substrate, was withdrawn from the market in August 2001 after reports of fatal interactions with the CYP2C8 inhibitor gemfibrozil [52].

4. The test article is metabolized by multiple P450 isoforms. This is generally interpreted that the test article may not have serious interactions with a specific inhibitor of one of the P450 isoforms, as the metabolic clearance can be carried out by the unaffected pathways. However, there are examples of drugs that have been found to be metabolized by multiple pathways but later were found in clinical or postmarketing studies to have interactions with potent inhibitors of a specific pathway. An example is the antifungal terbinafine, which has been characterized using human liver microsomes and rCYP isoforms to be metabolized by multiple P450 isoforms: CYP1A2, CYP2C8, CYP2C9, CYP2C19, CYP2D6, and CYP3A4, leading to the conclusion that “the potential for terbinafine interaction with other drugs is predicted to be insignificant” [53]. In the same study, as terbinafine was a competitive inhibitor of CYP2D6, it was concluded that it would have interactions with CYP2D6 substrates. *In vivo* studies confirmed the CYP2D6 inhibitory effects as predicted by *in vitro* studies; however, it was also observed clinically that rifampin, a CYP3A4 inducer, caused a 100% increase in terbinafine clearance ([www.fda.gov/medwatch/safety/2004/jan\\_PI/Lamasil\\_PI.pdf](http://www.fda.gov/medwatch/safety/2004/jan_PI/Lamasil_PI.pdf)). One possible explanation of this is that, upon CYP3A4 induction, the total metabolism of terbinafine is greatly enhanced due to the high capacity of CYP3A4 for this substrate. It is therefore important to realize that if a drug is metabolized by multiple isoforms, it may still have significant drug interactions with inducers of isoforms with high capacity for the metabolism of the drug.

#### 24.9.2 P450 Inhibition

The outcomes of P450 inhibition studies may include the following:

1. No inhibition is observed. If no inhibitory effects are observed with rCYP, microsomes, and hepatocytes, the substance in question is considered not to have the potential to cause inhibitory metabolic drug–drug interactions *in vivo*. As of now, there are no examples of *in vivo* enzyme inhibitors that are not inhibitors *in vitro*.
2. Significant inhibition is observed. A practical definition of significant inhibition is that the test article is found to cause dose-dependent and >50% inhibition of one or more P450 isoforms at the concentrations evaluated. The conclusion is that the test article is a potent inhibitor. As described earlier, the physiological significance is determined by the  $[I]/K_i$  value, with any  $[I]/K_i$  value of 0.1 or higher as possible or likely to cause *in vivo* drug–drug interactions. It is recommended that  $[I]/K_i$  values obtained from cell-free systems (microsomes and rCYP) are confirmed with intact hepatocytes to aid an accurate prediction

of *in vivo* effects. If the results with hepatocytes are also determined to be significant, *in vivo* studies will need to be performed to estimate human *in vivo* drug–drug interaction potential.

3. No time-dependent inhibition is observed. The inhibitor is not a mechanism-based inhibitor.
4. Time-dependent inhibition is observed. The inhibitor is a time-dependent inhibitor. *In vivo* studies will need to be performed to further define its drug–drug interaction potential.
5. Additional safety concern. A time-dependent inhibitor may need to be further studied to define its hepatotoxic potential, as a number of time-dependent P450 inhibitors are found to cause idiosyncratic hepatotoxicity.

### 24.9.3 P450 Induction

The following outcomes may be observed:

1. No induction is observed. The substance evaluated is not an enzyme inducer if P450 inhibitory and cytotoxic potential are eliminated as confounding factors.
2. Induction is observed. The substance evaluated is observed to cause dose-dependent and physiologically significant induction (e.g., induced activity greater than twofold the negative control activity). If the doses found to be positive are within clinical plasma concentrations (e.g., within 10× of plasma  $C_{max}$ ), *in vivo* studies may be needed to further define the test article's *in vivo* enzyme induction and the subsequent drug–drug interaction potential.
3. Additional safety concern. Enzyme inducers may need to be further evaluated for their hepatotoxic potential, as a large number of enzyme-inducing drugs are found to cause severe hepatotoxicity.

## 24.10 CONCLUSION

Drug–drug interactions can have serious adverse consequences and therefore should be evaluated accurately before a new drug is introduced to the human population. Due to the scientific advances in the understanding of key human drug-metabolizing pathways, and the availability of human *in vitro* systems for drug metabolism studies, human drug–drug interaction evaluations, especially drug metabolism-related interactions, can be performed rapidly and efficiently. A scientific, mechanism-based approach to evaluate drug–drug interactions remains the most appropriate approach:

1. Understanding of the major drug-metabolizing pathways in the metabolism of the drug or drug candidate in question will allow assessment of its potential interactions with existing drugs that are inhibitors or inducers of the pathways involved.
2. A careful and exhaustive evaluation of the inhibitory potential of the drug or drug candidate in question toward the major human drug metabolism enzymes

will allow assessment of its potential to cause interactions with existing drugs that are substrates of the inhibited enzymes.

3. Evaluation of induction potential of the drug or drug candidate in question for the inducible human drug-metabolizing enzymes will allow assessment of potential interactions with drugs that are substrates of the induced enzymes.

This approach is mainly applied toward P450 isoforms but can also be applied to non-P450 drug-metabolizing enzyme pathways. The next wave of major advances in drug–drug interactions is anticipated to be approaches for the evaluation of the interactions between drugs and drug transporters.

The success achieved with the scientific-based approaches in the evaluation of drug–drug interactions is a result of the extensive scientific research in the identification and characterization of drug-metabolizing enzymes, the definition of the mechanisms of metabolism-based drug–drug interactions, and the development, characterization, and intelligent application of the human-based *in vitro* experimental models for drug metabolism. Similar approaches should be adopted for the evaluation of other major adverse drug effects (e.g., idiosyncratic drug toxicity), which so far have eluded the routine drug safety evaluation approaches. It is through an open mind—a willingness to venture toward the development of a hypothesis, the testing of a hypothesis, and the development and adoption of approaches to investigate a problem based on the best science—that the field of drug safety evaluation can move forward.

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