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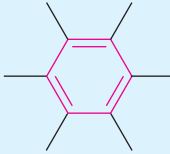



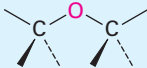
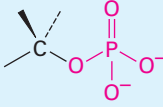
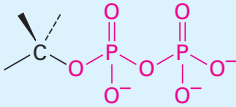

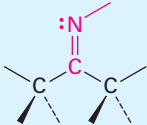



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## Structures of Some Common Functional Groups

Name	Structure*	Name ending	Example
Alkene (double bond)		<i>-ene</i>	H <sub>2</sub> C=CH <sub>2</sub> Ethene
Alkyne (triple bond)		<i>-yne</i>	HC≡CH Ethyne
Arene (aromatic ring)		None	 Benzene
Halide	 (X = F, Cl, Br, I)	None	CH <sub>3</sub> Cl Chloromethane
Alcohol		<i>-ol</i>	CH <sub>3</sub> OH Methanol
Ether		<i>ether</i>	CH <sub>3</sub> OCH <sub>3</sub> Dimethyl ether
Monophosphate		<i>phosphate</i>	CH <sub>3</sub> OPO <sub>3</sub> <sup>2-</sup> Methyl phosphate
Diphosphate		<i>diphosphate</i>	CH <sub>3</sub> OP <sub>2</sub> O <sub>6</sub> <sup>3-</sup> Methyl diphosphate
Amine		<i>-amine</i>	CH <sub>3</sub> NH <sub>2</sub> Methylamine
Imine (Schiff base)		None	 CH <sub>3</sub> CCH <sub>3</sub> Acetone imine
Nitrile		<i>-nitrile</i>	CH <sub>3</sub> C≡N Ethanenitrile
Thiol		<i>-thiol</i>	CH <sub>3</sub> SH Methanethiol

\*The bonds whose connections aren't specified are assumed to be attached to carbon or hydrogen atoms in the rest of the molecule.

Name	Structure*	Name ending	Example
Sulfide		<i>sulfide</i>	CH <sub>3</sub> SCH <sub>3</sub> Dimethyl sulfide
Disulfide		<i>disulfide</i>	CH <sub>3</sub> SSCH <sub>3</sub> Dimethyl disulfide
Sulfoxide		<i>sulfoxide</i>	 CH <sub>3</sub> SCH <sub>3</sub> Dimethyl sulfoxide
Aldehyde		<i>-al</i>	 CH <sub>3</sub> CH Ethanal
Ketone		<i>-one</i>	 CH <sub>3</sub> CCH <sub>3</sub> Propanone
Carboxylic acid		<i>-oic acid</i>	 CH <sub>3</sub> COH Ethanoic acid
Ester		<i>-oate</i>	 CH <sub>3</sub> COCH <sub>3</sub> Methyl ethanoate
Thioester		<i>-thioate</i>	 CH <sub>3</sub> CSCH <sub>3</sub> Methyl ethanethioate
Amide		<i>-amide</i>	 CH <sub>3</sub> CNH <sub>2</sub> Ethanamide
Acid chloride		<i>-oyl chloride</i>	 CH <sub>3</sub> CCl Ethanoyl chloride
Carboxylic acid anhydride		<i>-oic anhydride</i>	 CH <sub>3</sub> COCH <sub>3</sub> Ethanoic anhydride

\*The bonds whose connections aren't specified are assumed to be attached to carbon or hydrogen atoms in the rest of the molecule.